- 1 -

Substituted benzoylpyrazoles

The invention relates to novel substituted benzoylpyrazoles, to processes for their preparation and to their use as herbicides.

It is already known that certain substituted benzoylpyrazoles have herbicidal properties (cf. EP-A-352543, WO-A-96/26206, WO-A-97/35850, WO-A-97/41105, WO-A-97/41116, WO-A-97/41117, WO-A-97/41118, WO-A-97/46530, WO-A-98/28981, WO-A-98/31681, WO-A-98/31682, WO-A-99/07697). However, the activity of these compounds is not entirely satisfactory.

This invention now provides the novel substituted benzoylpyrazoles of the general formula (I)

(I)
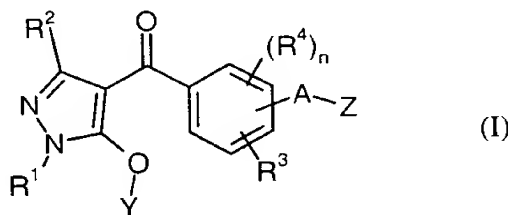
in which

n   represents the numbers 0, 1, 2 or 3,

A   represents a single bond or represents alkanediyl (alkylene),

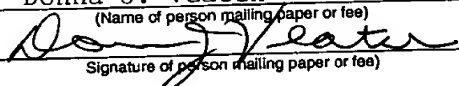R^1 represents in each case optionally substituted alkyl, alkenyl, alkinyl or cycloalkyl,

R^2 represents hydrogen, cyano, carbamoyl, thiocarbamoyl, halogen, or represents in each case optionally substituted alkyl, alkoxy, alkylthio, alkoxycarbonyl or cycloalkyl,

5   R^3 represents hydrogen, nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, halogen, or represents in each case optionally substituted alkyl, alkoxy, alkyl-thio, alkylsulfinyl, alkylsulfonyl, alkylamino, dialkylamino or dialkylamino-sulfonyl,

10   R^4 represents nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, halogen, or represents in each case optionally substituted alkyl, alkoxy, alkylthio, alkylsulfinyl, alkylsulfonyl, alkylamino, dialkylamino or dialkyl-aminosulfonyl,

15   Y   represents hydrogen or represents in each case optionally substituted alkyl, alkylcarbonyl, alkoxycarbonyl, alkylsulfonyl, alkylaminocarbonyl, dialkyl-aminocarbonyl, alkenyl, alkenylcarbonyl, alkenylsulfonyl, alkinyl, alkinyl-carbonyl, cycloalkyl, cycloalkylcarbonyl, cycloalkylalkyl, phenylcarbonyl, phenylsulfonyl, phenylalkyl or phenylcarbonylalkyl, and

20

Z   represents an optionally substituted 4- to 12-membered saturated or unsaturated monocyclic or bicyclic heterocyclic grouping which contains 1 to 4 heteroatoms (up to 4 nitrogen atoms and optionally - alternatively or additionally - one oxygen atom or one sulfur atom, or an SO grouping or an SO$_2$ grouping) and which additionally contains one to three oxo groups (C=O) and/or thioxo groups (C=S) as component of the heterocycle,

25

- including all possible tautomeric forms of the compounds of the general formula (I) and the possible salts of the compounds of the general formula (I).

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In the definitions, the hydrocarbon chains, such as alkyl or alkanediyl, are in each case straight-chain or branched - including in combination with heteroatoms, such as in alkoxy.

5 n preferably represents the numbers 0, 1 or 2.

 A preferably represents a single bond or represents alkanediyl (alkylene) having 1 to 4 carbon atoms.

10 R^1 preferably represents optionally cyano-, carboxyl-, carbamoyl-, halogen-, C_1-C_4-alkoxy-, C_1-C_4-alkyl-carbonyl-, C_1-C_4-alkoxy-carbonyl-, C_1-C_4-alkylthio-, C_1-C_4-alkylsulfinyl- or C_1-C_4-alkylsulfonyl-substituted alkyl having 1 to 6 carbon atoms, represents in each case optionally cyano-, carboxyl-, carbamoyl-, halogen- or C_1-C_4-alkoxy-carbonyl-substituted alkenyl

15 or alkinyl having in each case 2 to 6 carbon atoms, or represents optionally cyano-, carboxyl-, carbamoyl-, halogen-, C_1-C_4-alkyl- or C_1-C_4-alkoxy-carbonyl-substituted cycloalkyl having 3 to 6 carbon atoms.

 R^2 preferably represents hydrogen, cyano, carbamoyl, thiocarbamoyl, halogen, represents in each case optionally cyano-, halogen- or C_1-C_4-alkoxy-substituted alkyl, alkoxy or alkoxycarbonyl having in each case up to 6 carbon

20 atoms, represents optionally halogen-substituted alkylthio having 1 to 6 carbon atoms, or represents optionally cyano-, halogen- or C_1-C_4-alkyl-substituted cycloalkyl having 3 to 6 carbon atoms.

25

 R^3 preferably represents hydrogen, nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, halogen, represents in each case optionally halogen, C_1-C_4-alkoxy-, C_1-C_4-alkylthio-, C_1-C_4-alkylsulfinyl- or C_1-C_4-alkylsulfonyl-substituted alkyl, alkoxy, alkylthio, alkylsulfinyl or alkylsulfonyl having in each case up to 4 carbon atoms in the alkyl groups, or represents alkylamino,

30 dialkylamino or dialkylaminosulfonyl having in each case up to 4 carbon atoms in the alkyl groups.

R^4 preferably represents nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, halogen, represents in each case optionally halogen-, C_1-C_4-alkoxy-, C_1-C_4-alkylthio-, C_1-C_4-alkylsulfinyl- or C_1-C_4-alkylsulfonyl-substituted alkyl, alkoxy, alkylthio, alkylsulfinyl or alkylsulfonyl having in each case up to 4 carbon atoms in the alkyl groups, or represents alkylamino, dialkylamino or dialkylaminosulfonyl having in each case up to 4 carbon atoms in the alkyl groups.

Y   preferably represents hydrogen, represents in each case optionally cyano-, carboxyl-, carbamoyl-, halogen- or C_1-C_4-alkoxycarbonyl-substituted alkyl, alkylcarbonyl or alkoxycarbonyl having in each case up to 6 carbon atoms, represents in each case optionally halogen-substituted alkylsulfonyl, alkyl-aminocarbonyl or dialkylaminocarbonyl having in each case up to 6 carbon atoms in the alkyl groups, represents in each case optionally cyano-, carboxyl-, carbamoyl-, halogen- or C_1-C_4-alkoxy-carbonyl-substituted alkenyl, alkenylcarbonyl, alkinyl or alkinylcarbonyl having in each case 2 to 6 carbon atoms, represents optionally halogen-substituted alkenylsulfonyl having up to 6 carbon atoms represents in each case optionally cyano-, halogen- or C_1-C_4-alkyl-substituted cycloalkyl, cycloalkylcarbonyl or cycloalkylalkyl having in each case 3 to 6 carbon atoms in the cycloalkyl groups and optionally 1 to 3 carbon atoms in the alkyl moiety, or represents in each case optionally nitro-, cyano-, carboxyl-, carbamoyl-, halogen-, C_1-C_4-alkyl-, C_1-C_4-halogenoalkyl-, C_1-C_4-alkoxy- or C_1-C_4-halogenoalkoxy-substituted phenylcarbonyl, phenylsulfonyl, phenyl-C_1-C_4-alkyl or phenyl-carbonyl-C_1-C_4-alkyl.

Z   preferably represents one of the heterocyclic groupings below

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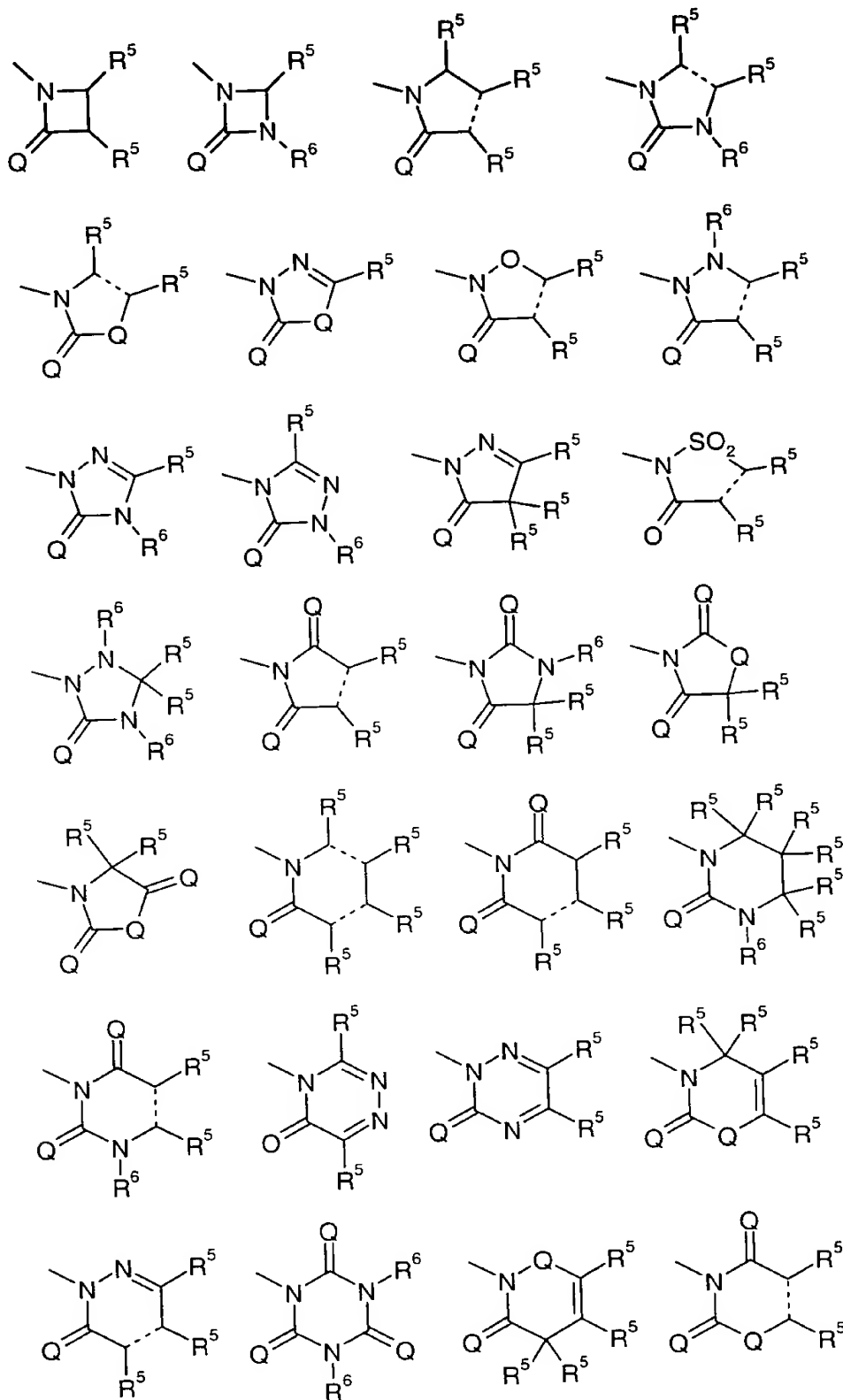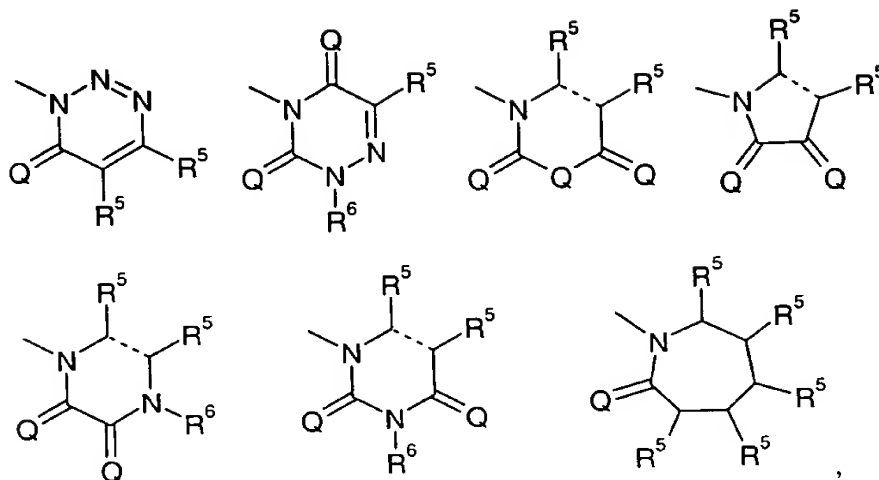
in which in each case the broken bond is a single bond or a double bond,

Q   represents oxygen or sulfur,

R^5 represents hydrogen, hydroxyl, mercapto, cyano, halogen, represents in each case optionally cyano-, halogen-, C_1-C_4-alkoxy-, C_1-C_4-alkyl-thio-, C_1-C_4-alkylsulfinyl- or C_1-C_4-alkylsulfonyl-substituted alkyl, alkylcarbonyl, alkoxy, alkoxycarbonyl, alkylthio, alkylsulfinyl or alkylsulfonyl having in each case up to 6 carbon atoms in the alkyl groups, represents propadienylthio, represents in each case optionally halogen-substituted alkylamino or dialkylamino having in each case up to 6 carbon atoms in the alkyl groups, represents in each case optionally halogen-substituted alkenyl, alkinyl, alkenyloxy, alkenyl-thio or alkenylamino having in each case up to 6 carbon atoms in the alkenyl or alkinyl groups, represents in each case optionally halogen-substituted cycloalkyl, cycloalkyloxy, cycloalkylthio, cycloalkylamino, cycloalkylalkyl, cycloalkylalkoxy, cycloalkylalkylthio or cycloalkyl-alkylamino having in each case 3 to 6 carbon atoms in the cycloalkyl groups and optionally up to 4 carbon atoms in the alkyl moiety, represents in each case optionally halogen-, C_1-C_4-alkyl- or C_1-C_4-alkoxy-substituted phenyl, phenyloxy, phenylthio, phenylamino, benzyl, benzyloxy, benzylthio or benzylamino, represents pyrrolidino,

piperidino or morpholino, or - if two adjacent radicals R^5 and R^5 are located on a double bond - together with the adjacent radical R^5 also represents a benzo grouping, and

R^6 represents hydrogen, hydroxyl, amino, alkylideneamino having up to 4 carbon atoms, represents in each case optionally halogen- or C_1-C_4-alkoxy-substituted alkyl, alkoxy, alkylamino, dialkylamino or alkanoylamino having in each case up to 6 carbon atoms in the alkyl groups, represents in each case optionally halogen-substituted alkenyl, alkinyl or alkenyloxy having in each case up to 6 carbon atoms in the alkenyl or alkinyl groups, represents in each case optionally halogen-substituted cycloalkyl, cycloalkylalkyl or cycloalkylamino having in each case 3 to 6 carbon atoms in the cycloalkyl groups and optionally up to 3 carbon atoms in the alkyl moiety, or represents in each case optionally halogen-, C_1-C_4-alkyl- or C_1-C_4-alkoxy-substituted phenyl or benzyl, or together with an adjacent radical R^5 or R^6 represents optionally halogen- or C_1-C_4-alkyl-substituted alkanediyl having 3 to 5 carbon atoms,

where the individual radicals R^5 and R^6 - if a plurality of these are attached to the same heterocyclic groupings, may have identical or different meanings within the scope of the above definition.

Q preferably represents oxygen.

R^5 preferably represents hydrogen, hydroxyl, mercapto, cyano, fluorine, chlorine, bromine, iodine, represents in each case optionally fluorine-, chlorine-, methoxy-, ethoxy-, n- or i-propoxy-, n-, i-, s- or t-butoxy-, methylthio-, ethylthio-, n- or i-propylthio-, n-, i-, s- or t-butylthio-, methylsulfinyl-, ethylsulfinyl-, n- or i-propylsulfinyl-, methylsulfonyl-, ethylsulfonyl-, n- or i-propylsulfonyl-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, methoxy, ethoxy, n- or i-propoxy, n-, i-, s-

- 8 -

or t-butoxy, methylthio, ethylthio, n- or i-propylthio, n-, i-, s- or t-butylthio, methylsulfinyl, ethylsulfinyl, n- or i-propylsulfinyl, methylsulfonyl, ethylsulfonyl, n- or i-propylsulfonyl, represents methylamino, ethylamino, n- or i-propylamino, n-, i-, s- or t-butylamino, dimethylamino, diethylamino, di-n-propylamino or di-i-propylamino, represents in each case optionally fluorine- and/or chlorine-substituted ethenyl, propenyl, butenyl, ethinyl, propinyl, butinyl, propenyloxy, butenyloxy, propenylthio, butenylthio, propenylamino or butenylamino, represents in each case optionally fluorine- and/or chlorine-substituted cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopropyloxy, cyclobutyloxy, cyclopentyloxy, cyclohexyloxy, cyclopropylthio, cyclobutylthio, cyclopentylthio, cyclohexylthio, cyclopropylamino, cyclobutylamino, cyclopentylamino, cyclohexylamino, cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl, cyclohexylmethyl, cyclopropylmethoxy, cyclobutylmethoxy, cyclopentylmethoxy, cyclohexylmethoxy, cyclopropylmethylthio, cyclobutylmethylthio, cyclopentylmethylthio, cyclohexylmethylthio, cyclopropylmethylamino, cyclobutylmethylamino, cyclopentylmethylamino or cyclohexylmethylamino, represents in each case optionally fluorine-, chlorine-, methyl-, ethyl-, n- or i-propyl-, n-, i-, s- or t-butyl-, methoxy-, ethoxy-, n- or i-propoxy-substituted phenyl, phenyloxy, phenylthio, phenylamino, benzyl, benzyloxy, benzylthio or benzylamino, represents pyrrolidino, piperidino or morpholino, or - if two radicals R^5 and R^5 are located on a double bond - together with the adjacent radical R^5 also represents a benzo grouping.

R^6 preferably represents hydrogen, hydroxyl, amino, represents in each case optionally fluorine- and/or chlorine-, methoxy- or ethoxy-substituted, methyl, ethyl, n- or i-propyl, n-, i- or s-butyl, methoxy, ethoxy, n- or i-propoxy, methylamino, ethylamino or dimethylamino, represents in each case optionally fluorine- and/or chlorine-substituted

ethenyl, propenyl, ethinyl, propinyl or propenyloxy, represents in each case optionally fluorine- and/or chlorine-substituted cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopropylmethyl, cyclobutyl-methyl, cyclopentylmethyl, cyclohexylmethyl, or represents in each case optionally fluorine-, chlorine-, methyl-, ethyl-, n- or i-propyl, n-, i-, s- or t-butyl-, methoxy-, ethoxy-, n- or i-propoxy-substituted phenyl or benzyl, or together with an adjacent radical R^5 or R^6 represents in each case optionally methyl- and/or ethyl-substituted propane-1,3-diyl (trimethylene), butane-1,4-diyl (tetramethylene) or pentane-1,5-diyl (pentamethylene).

n   particularly preferably represents the numbers 0 or 1.

A   particularly preferably represents a single bond, methylene, ethylidene (ethane-1,1-diyl) or dimethylene (ethane-1,2-diyl).

R^1 particularly preferably represents in each case optionally fluorine-, chlorine-, methoxy-, ethoxy-, n- or i-propoxy-, methylthio-, ethylthio-, n- or i-propyl-thio-, methylsulfinyl-, ethylsulfinyl-, n- or i-propylsulfinyl-, methylsulfonyl-, ethylsulfonyl-, n- or i-propylsulfonyl-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, represents in each case optionally fluorine-, chlorine- or bromine-substituted propenyl, butenyl, propinyl or butinyl, or represents in each case optionally cyano-, fluorine-, chlorine-, bromine-, methyl- or ethyl-substituted cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl.

R^2 particularly preferably represents hydrogen, cyano, carbamoyl, thiocarbamoyl, represents in each case optionally cyano-, fluorine-, chlorine-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, methoxy, ethoxy, n- or i-propoxy, methoxycarbonyl, ethoxycarbonyl, n- or i-propoxycarbonyl, represents in each case optionally fluorine- and/or chlorine-substituted methylthio, ethylthio, n- or i-propylthio, or represents in each case

optionally cyano-, fluorine-, chlorine-, bromine-, methyl- or ethyl-substituted cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl.

R^3 particularly preferably represents hydrogen, nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, fluorine, chlorine, bromine, iodine, represents in each case optionally fluorine- and/or chlorine-, methoxy-, ethoxy-, n- or i-propoxy-, methylthio-, ethylthio-, n- or i-propylthio-, methylsulfinyl-, ethylsulfinyl-, methylsulfonyl- or ethylsulfonyl-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, represents in each case optionally fluorine- and/or chlorine-, methoxy-, ethoxy-, n- or i-propoxy-substituted methoxy, ethoxy, n- or i-propoxy, represents in each case optionally fluorine- and/or chlorine-substituted methylthio, ethylthio, n- or i-propylthio, methylsulfinyl, ethylsulfinyl, n- or i-propylsulfinyl, methylsulfonyl, ethylsulfonyl, n- or i-propylsulfonyl, or represents methylamino, ethylamino, n- or i-propylamino, dimethylamino, diethylamino, dimethylaminosulfonyl or diethylaminosulfonyl.

R^4 particularly preferably represents nitro, cyano, carboxyl, carbamoyl, thiocarbamoyl, fluorine, chlorine, bromine, represents in each case optionally fluorine- and/or chlorine-, methoxy-, ethoxy-, n- or i-propoxy-, methylthio-, ethylthio-, n- or i-propylthio-, methylsulfinyl-, ethylsulfinyl-, methylsulfonyl- or ethylsulfonyl-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, represents in each case optionally fluorine- and/or chlorine-, methoxy-, ethoxy-, n- or i-propoxy-substituted methoxy, ethoxy, n- or i-propoxy, represents in each case optionally fluorine- and/or chlorine-substituted methylthio, ethylthio, n- or i-propylthio, methylsulfinyl, ethylsulfinyl, n- or i-propylsulfinyl, methylsulfonyl, ethylsulfonyl, n- or i-propylsulfonyl, or represents methylamino, ethylamino, n- or i-propylamino, dimethylamino, diethylamino, dimethylaminosulfonyl or diethylaminosulfonyl.

R^5 particularly preferably represents hydrogen, hydroxyl, chlorine, bromine, methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, difluoromethyl, dichloromethyl, trifluoromethyl, trichloromethyl, chlorodifluoromethyl, fluoro-

dichloromethyl, fluoroethyl, chloroethyl, difluoroethyl, dichloroethyl, fluoro-n-propyl, fluoro-i-propyl, chloro-n-propyl, chloro-i-propyl, methoxymethyl, ethoxymethyl, methoxyethyl, ethoxyethyl, methoxy, ethoxy, n- or i-propoxy, n-, i-, s- or t-butoxy, fluoroethoxy, chloroethoxy, difluoroethoxy, dichloro-ethoxy, trifluoroethoxy, trichloroethoxy, chlorofluoroethoxy, chlorodifluoro-ethoxy, fluorodichloroethoxy, methylthio, ethylthio, n- or i-propylthio, fluoro-ethylthio, chloroethylthio, difluoroethylthio, dichloroethylthio, chlorofluoro-ethylthio, chlorodifluoroethylthio, fluorodichloroethylthio, methylsulfinyl, ethylsulfinyl, n- or i-propylsulfinyl, methylsulfonyl, ethylsulfonyl, n- or i-propylsulfonyl, dimethylamino, propenylthio, butenylthio, propinylthio, butinylthio, cyclopropyl, cyclopropylmethyl, cyclopropylmethoxy, phenyl or phenoxy.

R^6 particularly preferably represents amino, methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, methoxy, ethoxy, methylamino, dimethylamino, cyclopropyl or cyclopropylmethyl, or together with R^5 represents propane-1,3-diyl (trimethylene), butane-1,4-diyl (tetramethylene) or pentane-1,5-diyl (penta-methylene).

Y particularly preferably represents hydrogen, represents in each case optionally cyano-, fluorine-, chlorine-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, acetyl, propionyl, n- or i-butyroyl, methoxycarbonyl or ethoxy-carbonyl, represents in each case optionally fluorine-, chlorine- and/or bromine-substituted methylsulfonyl-, ethylsulfonyl-, n- or i-propylsulfonyl-, n-, i-, s- or t-butylsulfonyl-, methylaminocarbonyl, ethylaminocarbonyl, n- or i-propylaminocarbonyl, dimethylaminocarbonyl or diethylaminocarbonyl, represents in each case optionally fluorine-, chlorine- or bromine-substituted propenyl, butenyl, propenylcarbonyl, butenylcarbonyl, propenylsulfonyl, butenylsulfonyl, propinyl, butinyl, propinylcarbonyl or butinylcarbonyl, represents in each case optionally cyano-, fluorine-, chlorine-, methyl- or ethyl-substituted cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclo-propylcarbonyl, cyclobutylcarbonyl, cyclopentylcarbonyl, cyclohexylcarbonyl,

- 12 -

cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl or cyclohexyl-methyl, or represents in each case optionally nitro-, cyano-, fluorine-, chlorine-, bromine-, methyl-, ethyl-, n- or i-propyl-, n-, i-, s- or t-butyl-, trifluoromethyl-, methoxy-, ethoxy-, n- or i-propoxy-, difluoromethoxy- or trifluoromethoxy-substituted phenylcarbonyl, phenylsulfonyl, benzyl or phenyl-carbonylmethyl.

Z   particularly preferably represents
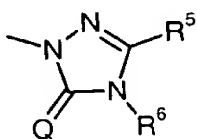
n   very particularly preferably represents O.

A   very particularly preferably represents a single bond or represents methylene.

R^1 very particularly preferably represents in each case optionally fluorine-, chlorine-, methoxy-, ethoxy-, methylthio-, ethylthio-, methylsulfinyl-, ethyl-sulfinyl-, methylsulfonyl- or ethylsulfonyl-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, or represents optionally cyano-, fluorine-, chlorine-, bromine-, methyl- or ethyl-substituted cyclopropyl.

R^2 very particularly preferably represents hydrogen, cyano, carbamoyl, fluorine, chlorine, bromine, represents in each case optionally cyano-, fluorine-, chlorine-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, methoxycarbonyl, ethoxycarbonyl, n- or i-propoxycarbonyl, represents in each case optionally fluorine- and/or chlorine-substituted methylthio, ethylthio, n- or i-propylthio, or represents cyclopropyl.

R^3 very particularly preferably represents hydrogen, nitro, cyano, fluorine, chlorine, bromine, iodine, methyl, ethyl, trifluoromethyl, methoxymethyl,

methylthiomethyl, methylsulfinylmethyl, methylsulfonylmethyl, methoxy, ethoxy, difluoromethoxy, trifluoromethoxy, methylthio, ethylthio, methylsulfinyl, ethylsulfinyl, methylsulfonyl, ethylsulfonyl or dimethylaminosulfonyl.

5

R^4 particularly preferably represents nitro, cyano, fluorine, chlorine, bromine, methyl, ethyl, trifluoromethyl, methoxymethyl, methylthiomethyl, methylsulfinylmethyl, methylsulfonylmethyl, methoxy, ethoxy, difluoromethoxy, trifluoromethoxy, methylthio, ethylthio, methylsulfinyl, ethylsulfinyl, methylsulfonyl, ethylsulfonyl, dimethylamino or dimethylaminosulfonyl.

10

R^6 very particularly preferably represents methyl, cyclopropyl, dimethylamino, methoxy or ethoxy.

15 Y very particularly preferably represents hydrogen.

A   most preferably represents methylene.

R^1 most preferably represents methyl or ethyl.

5

R^2 most preferably represents hydrogen or methyl.

R^3 most preferably represents hydrogen, fluorine, chlorine, bromine, trifluoro-

methyl or methylsulfonyl.

10

R^4 most preferably represents (2-)chlorine, (4-)chlorine, (6-)trifluoromethyl or

(2-)methylsulfonyl.

Preference according to the invention is given to compounds of the formula (I) which

contains a combination of the meanings listed above as being preferred.

15

Particular preference according to the invention is given to the compounds of the

formula (I) which contain a combination of the meanings listed above as being

particularly preferred.

20

Very particular preference is given to the compounds of the formula (I) which

contain a combination of the meanings listed above as being very particularly

preferred.

25   Most preference according to the invention is given to the compounds of the formula

(I) which contain a combination of the meanings listed above as being most

preferred.

The present invention in particular provides the compounds of the general formulae (IA), (IB) and (IC):

(IA)

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(IB)

(IC)
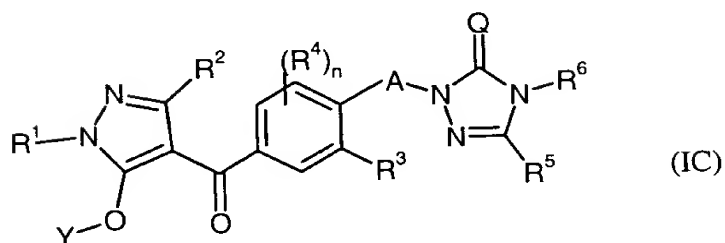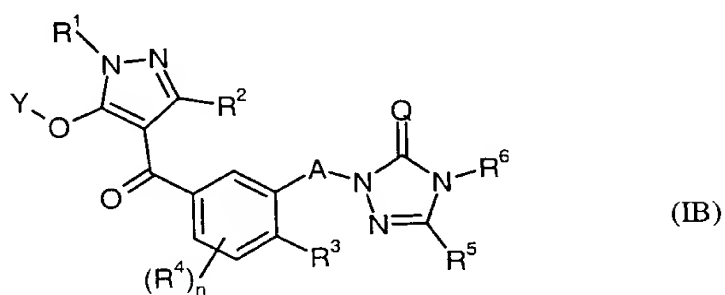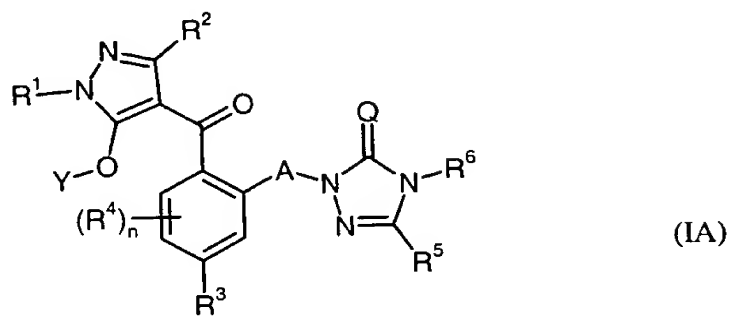
in which

15   n, A, Q, R^1, R^2, R^3, R^4, R^5, R^6 and Y   are as defined above.

The invention preferably also provides sodium, potassium, magnesium, calcium, ammonium, C_1-C_4-alkyl-ammonium, di-(C_1-C_4-alkyl)-ammonium, tri-(C_1-C_4-alkyl)-ammonium, tetra-(C_1-C_4-alkyl)-ammonium, tri-(C_1-C_4-alkyl)-sulfonium, C_5- or C_6-Cycloalkyl-ammonium and di-(C_1-C_2-alkyl)-benzylammonium salts of compounds of the formula (I), in which n, A, R^1, R^2, R^3, R^4, Y and Z are as defined above.

The abovementioned general or preferred radical definitions apply both to the end products of the formula (I) and, correspondingly to the starting materials or intermediates required in each case for the preparation. These radical definitions can be combined with one another as desired, i.e. including combinations between the given preferred ranges.

Examples of the compounds of the general formula (I) according to the invention are given in the groups below.

Group 1

(IA-1)

Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given in the table below:

R^3	(Position) $(R^4)_n$	R^5	R^6
H	-	CF_3	CH_3
F	-	CF_3	CH_3
Cl	-	CF_3	CH_3
Br	-	CF_3	CH_3
I	-	CF_3	CH_3
NO_2	-	CF_3	CH_3
CN	-	CF_3	CH_3
CH_3	-	CF_3	CH_3
OCH_3	-	CF_3	CH_3
CF_3	-	CF_3	CH_3
$OCHF_2$	-	CF_3	CH_3
OCF_3	-	CF_3	CH_3
SO_2CH_3	-	CF_3	CH_3
H	-	OCH_3	CH_3
F	-	OCH_3	CH_3
Cl	-	OCH_3	CH_3
Br	-	OCH_3	CH_3
I	-	OCH_3	CH_3
NO_2	-	OCH_3	CH_3
CN	-	OCH_3	CH_3
CH_3	-	OCH_3	CH_3
OCH_3	-	OCH_3	CH_3
CF_3	-	OCH_3	CH_3
$OCHF_2$	-	OCH_3	CH_3

R^3	(Position) (R^4)$_n$	R^5	R^6
OCF$_3$	-	OCH$_3$	CH$_3$
SO$_2$CH$_3$	-	OCH$_3$	CH$_3$
H	-	SCH$_3$	CH$_3$
F	-	SCH$_3$	CH$_3$
Cl	-	SCH$_3$	CH$_3$
Br	-	SCH$_3$	CH$_3$
I	-	SCH$_3$	CH$_3$
NO$_2$	-	SCH$_3$	CH$_3$
CN	-	SCH$_3$	CH$_3$
CH$_3$	-	SCH$_3$	CH$_3$
OCH$_3$	-	SCH$_3$	CH$_3$
CF$_3$	-	SCH$_3$	CH$_3$
OCHF$_2$	-	SCH$_3$	CH$_3$
OCF$_3$	-	SCH$_3$	CH$_3$
SO$_2$CH$_3$	-	SCH$_3$	CH$_3$
H	-	OC$_2$H$_5$	CH$_3$
F	-	OC$_2$H$_5$	CH$_3$
Cl	-	OC$_2$H$_5$	CH$_3$
Br	-	OC$_2$H$_5$	CH$_3$
I	-	OC$_2$H$_5$	CH$_3$
NO$_2$	-	OC$_2$H$_5$	CH$_3$
CN	-	OC$_2$H$_5$	CH$_3$
CH$_3$	-	OC$_2$H$_5$	CH$_3$
OCH$_3$	-	OC$_2$H$_5$	CH$_3$
CF$_3$	-	OC$_2$H$_5$	CH$_3$

R^3	(Position) $(R^4)_n$	R^5	R^6
$OCHF_2$	-	OC_2H_5	CH_3
OCF_3	-	OC_2H_5	CH_3
SO_2CH_3	-	OC_2H_5	CH_3
H	-	$N(CH_3)_2$	CH_3
F	-	$N(CH_3)_2$	CH_3
Cl	-	$N(CH_3)_2$	CH_3
Br	-	$N(CH_3)_2$	CH_3
I	-	$N(CH_3)_2$	CH_3
NO_2	-	$N(CH_3)_2$	CH_3
CN	-	$N(CH_3)_2$	CH_3
CH_3	-	$N(CH_3)_2$	CH_3
OCH_3	-	$N(CH_3)_2$	CH_3
CF_3	-	$N(CH_3)_2$	CH_3
$OCHF_2$	-	$N(CH_3)_2$	CH_3
OCF_3	-	$N(CH_3)_2$	CH_3
SO_2CH_3	-	$N(CH_3)_2$	CH_3
H	-	OCH_3	
F	-	OCH_3	
Cl	-	OCH_3	
Br	-	OCH_3	
I	-	OCH_3	

R^3	(Position) (R^4)$_n$	R^5	R^6
NO$_2$	-	OCH$_3$	cyclopropyl
CN	-	OCH$_3$	cyclopropyl
CH$_3$	-	OCH$_3$	cyclopropyl
OCH$_3$	-	OCH$_3$	cyclopropyl
CF$_3$	-	OCH$_3$	cyclopropyl
OCHF$_2$	-	OCH$_3$	cyclopropyl
OCF$_3$	-	OCH$_3$	cyclopropyl
SO$_2$CH$_3$	-	OCH$_3$	cyclopropyl
H	(5-) Cl	CF$_3$	CH$_3$
F	(5-) Cl	CH$_3$	CH$_3$
Cl	(5-) Cl	OCH$_3$	CH$_3$
Br	(5-) Cl	Br	cyclopropyl
Cl	(5-) Cl	CF$_3$	CH$_3$
NO$_2$	(5-) Cl	CH$_3$	CH$_3$
Cl	(5-) Cl	SCH$_3$	CH$_3$
CH$_3$	(5-) Cl	Cl	CH$_3$
OCH$_3$	(5-) Cl	OCH$_3$	CH$_3$

R³	(Position) (R⁴)ₙ	R⁵	R⁶
CF₃	(5-) Cl	CF₃	CH₃
OCHF₂	(5-) Cl	CH₃	CH₃
OCF₃	(5-) Cl	CH₃	CH₃
SO₂CH₃	(5-) Cl	OCH₃	CH₃

Group 2

(IA-2)
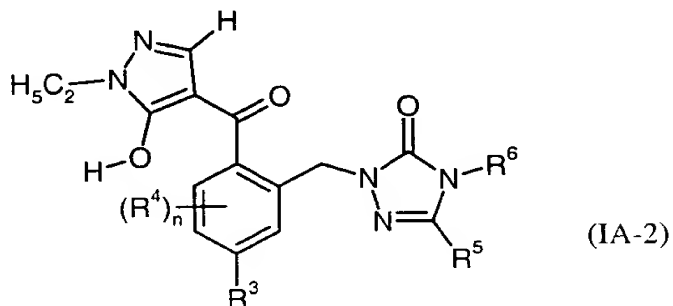
Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 1.

Group 3

(IA-3)
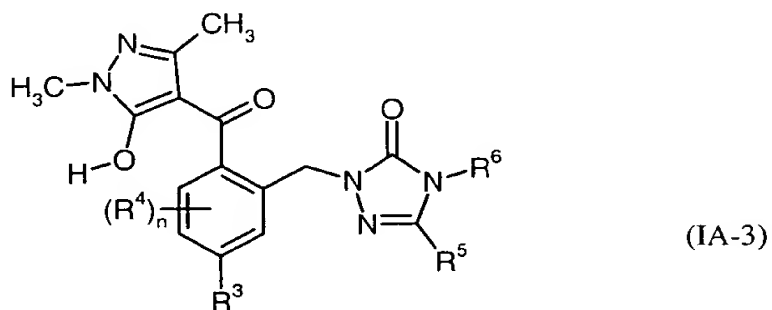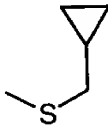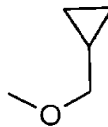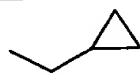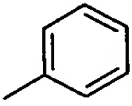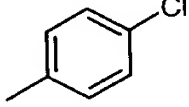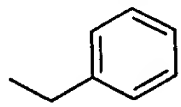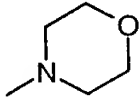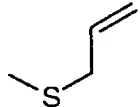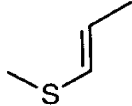
Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 1.

Group 4
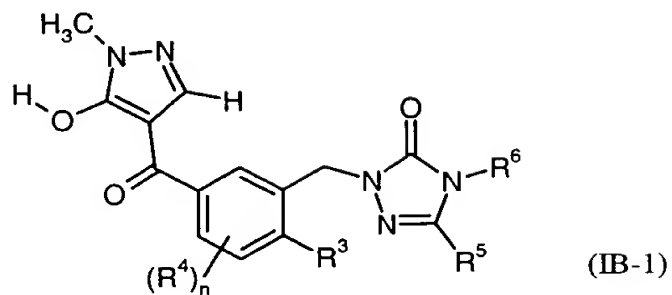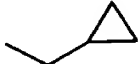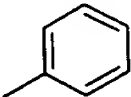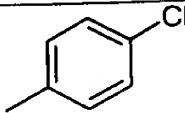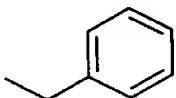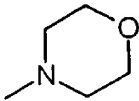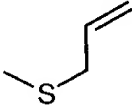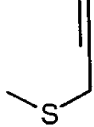
(IB-1)

5   Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given in the table below:
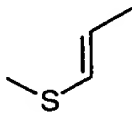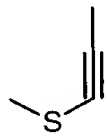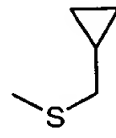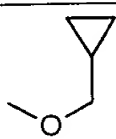
R^3	(Position) $(R^4)_n$	R^5	R^6
Cl	(2-) Cl	CF_3	CH_3
Cl	(2-) Cl	SCH_3	CH_3
Cl	(2-) Cl	SC_2H_5	CH_3
Cl	(2-) Cl	SC_3H_7	CH_3
Cl	(2-) Cl	SC_3H_7-i	CH_3
Cl	(2-) Cl		CH_3
Cl	(2-) Cl		CH_3
Cl	(2-) Cl		CH_3
Cl	(2-) Cl		CH_3

R^3	(Position) (R^4)$_n$	R^5	R^6
Cl	(2-) Cl	(cyclopropyl-CH$_2$-S- structure)	CH$_3$
Cl	(2-) Cl	SCH=C=CH$_2$	CH$_3$
Cl	(2-) Cl	SCH$_2$CN	CH$_3$
Cl	(2-) Cl	SCH$_2$CH$_2$CN	CH$_3$
Cl	(2-) Cl	OCH$_3$	CH$_3$
Cl	(2-) Cl	OC$_2$H$_5$	CH$_3$
Cl	(2-) Cl	OC$_3$H$_7$	CH$_3$
Cl	(2-) Cl	OC$_3$H$_7$-i	CH$_3$
Cl	(2-) Cl	OC$_4$H$_9$	CH$_3$
Cl	(2-) Cl	OCH$_2$CF$_3$	CH$_3$
Cl	(2-) Cl	(cyclopropyl-CH$_2$-O- structure)	CH$_3$
Cl	(2-) Cl	OC$_6$H$_5$	CH$_3$
Cl	(2-) Cl	H	CH$_3$
Cl	(2-) Cl	CH$_3$	CH$_3$
Cl	(2-) Cl	C$_2$H$_5$	CH$_3$
Cl	(2-) Cl	C$_3$H$_7$	CH$_3$
Cl	(2-) Cl	C$_3$H$_7$-i	CH$_3$
Cl	(2-) Cl	C$_4$H$_9$	CH$_3$
Cl	(2-) Cl	C$_4$H$_9$-i	CH$_3$
Cl	(2-) Cl	C$_4$H$_9$-s	CH$_3$
Cl	(2-) Cl	C$_4$H$_9$-t	CH$_3$
Cl	(2-) Cl	(cyclopropyl structure)	CH$_3$
Cl	(2-) Cl	(cyclopropyl-methyl structure)	CH$_3$
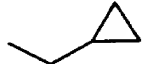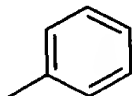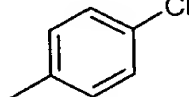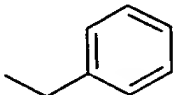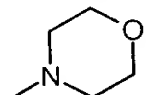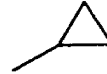
R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) Cl	CH=CHCH₃	CH₃
Cl	(2-) Cl	(methylphenyl)	CH₃
Cl	(2-) Cl	(chloro-methylphenyl)	CH₃
Cl	(2-) Cl	(ethylphenyl)	CH₃
Cl	(2-) Cl	N(CH₃)₂	CH₃
Cl	(2-) Cl	(N-methylmorpholine)	CH₃
Cl	(2-) Cl	Cl	CH₃
Cl	(2-) Cl	Br	CH₃
SO₂CH₃	(2-) Cl	CF₃	CH₃
SO₂CH₃	(2-) Cl	SCH₃	CH₃
SO₂CH₃	(2-) Cl	SC₂H₅	CH₃
SO₂CH₃	(2-) Cl	SC₃H₇	CH₃
SO₂CH₃	(2-) Cl	SC₃H₇-i	CH₃
SO₂CH₃	(2-) Cl	(CH₂=CHCH₂-S-)	CH₃
SO₂CH₃	(2-) Cl	(HC≡CCH₂-S-)	CH₃
SO₂CH₃	(2-) Cl	(CH₃CH=CH-S-)	CH₃

R^3	(Position) $(R^4)_n$	R^5	R^6
SO_2CH_3	(2-) Cl		CH_3
SO_2CH_3	(2-) Cl		CH_3
SO_2CH_3	(2-) Cl	$SCH=C=CH_2$	CH_3
SO_2CH_3	(2-) Cl	SCH_2CN	CH_3
SO_2CH_3	(2-) Cl	SCH_2CH_2CN	CH_3
SO_2CH_3	(2-) Cl	OCH_3	CH_3
SO_2CH_3	(2-) Cl	OC_2H_5	CH_3
SO_2CH_3	(2-) Cl	OC_3H_7	CH_3
SO_2CH_3	(2-) Cl	OC_3H_7-i	CH_3
SO_2CH_3	(2-) Cl	OC_4H_9	CH_3
SO_2CH_3	(2-) Cl	OCH_2CF_3	CH_3
SO_2CH_3	(2-) Cl		CH_3
SO_2CH_3	(2-) Cl	OC_6H_5	CH_3
SO_2CH_3	(2-) Cl	H	CH_3
SO_2CH_3	(2-) Cl	CH_3	CH_3
SO_2CH_3	(2-) Cl	C_2H_5	CH_3
SO_2CH_3	(2-) Cl	C_3H_7	CH_3
SO_2CH_3	(2-) Cl	C_3H_7-i	CH_3
SO_2CH_3	(2-) Cl	C_4H_9	CH_3
SO_2CH_3	(2-) Cl	C_4H_9-i	CH_3
SO_2CH_3	(2-) Cl	C_4H_9-s	CH_3
SO_2CH_3	(2-) Cl	C_4H_9-t	CH_3

R³	(Position) (R⁴)ₙ	R⁵	R⁶
SO₂CH₃	(2-) Cl	cyclopropyl	CH₃
SO₂CH₃	(2-) Cl	cyclopropylmethyl	CH₃
SO₂CH₃	(2-) Cl	CH=CHCH₃	CH₃
SO₂CH₃	(2-) Cl	2-methylphenyl	CH₃
SO₂CH₃	(2-) Cl	4-chloro-2-methylphenyl	CH₃
SO₂CH₃	(2-) Cl	2-(phenyl)ethyl	CH₃
SO₂CH₃	(2-) Cl	N(CH₃)₂	CH₃
SO₂CH₃	(2-) Cl	4-methylmorpholin-4-yl	CH₃
SO₂CH₃	(2-) Cl	Cl	CH₃
SO₂CH₃	(2-) Cl	Br	CH₃
Cl	(2-) SO₂CH₃	CF₃	CH₃
Cl	(2-) SO₂CH₃	SCH₃	CH₃
Cl	(2-) SO₂CH₃	SC₂H₅	CH₃
Cl	(2-) SO₂CH₃	SC₃H₇	CH₃
Cl	(2-) SO₂CH₃	SC₃H₇-i	CH₃
Cl	(2-) SO₂CH₃	S-CH₂-CH=CH₂	CH₃
Cl	(2-) SO₂CH₃	S-CH₂-C≡CH	CH₃
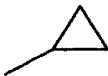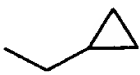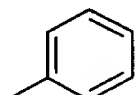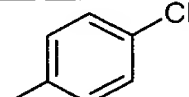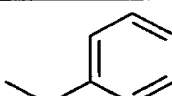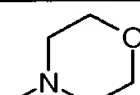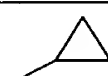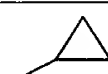
R^3	(Position) $(R^4)_n$	R^5	R^6
Cl	(2-) SO_2CH_3	$S-CH=CH-CH_3$	CH_3
Cl	(2-) SO_2CH_3	$S-C\equiv C-CH_3$	CH_3
Cl	(2-) SO_2CH_3	$S-CH_2-\text{cyclopropyl}$	CH_3
Cl	(2-) SO_2CH_3	$SCH=C=CH_2$	CH_3
Cl	(2-) SO_2CH_3	SCH_2CN	CH_3
Cl	(2-) SO_2CH_3	SCH_2CH_2CN	CH_3
Cl	(2-) SO_2CH_3	OCH_3	CH_3
Cl	(2-) SO_2CH_3	OC_2H_5	CH_3
Cl	(2-) SO_2CH_3	OC_3H_7	CH_3
Cl	(2-) SO_2CH_3	$OC_3H_7\text{-}i$	CH_3
Cl	(2-) SO_2CH_3	OC_4H_9	CH_3
Cl	(2-) SO_2CH_3	OCH_2CF_3	CH_3
Cl	(2-) SO_2CH_3	$O-CH_2-\text{cyclopropyl (epoxide)}$	CH_3
Cl	(2-) SO_2CH_3	OC_6H_5	CH_3
Cl	(2-) SO_2CH_3	H	CH_3
Cl	(2-) SO_2CH_3	CH_3	CH_3
Cl	(2-) SO_2CH_3	C_2H_5	CH_3
Cl	(2-) SO_2CH_3	C_3H_7	CH_3
Cl	(2-) SO_2CH_3	$C_3H_7\text{-}i$	CH_3
Cl	(2-) SO_2CH_3	C_4H_9	CH_3
Cl	(2-) SO_2CH_3	$C_4H_9\text{-}i$	CH_3
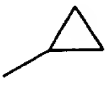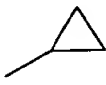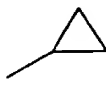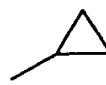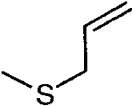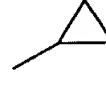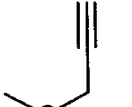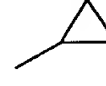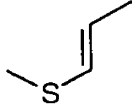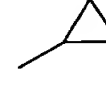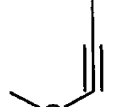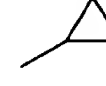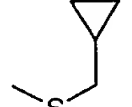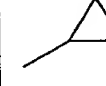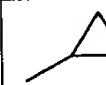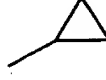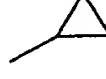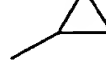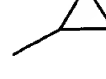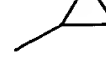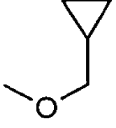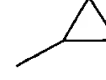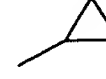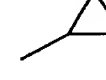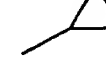
R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) SO_2CH_3	C_4H_9-s	CH_3
Cl	(2-) SO_2CH_3	C_4H_9-t	CH_3
Cl	(2-) SO_2CH_3	(cyclopropyl)	CH_3
Cl	(2-) SO_2CH_3	(cyclopropylmethyl/ethyl)	CH_3
Cl	(2-) SO_2CH_3	$CH=CHCH_3$	CH_3
Cl	(2-) SO_2CH_3	(4-methylphenyl)	CH_3
Cl	(2-) SO_2CH_3	(4-chloro-methylphenyl, Cl)	CH_3
Cl	(2-) SO_2CH_3	(ethylphenyl)	CH_3
Cl	(2-) SO_2CH_3	$N(CH_3)_2$	CH_3
Cl	(2-) SO_2CH_3	(N-methylmorpholine, O)	CH_3
Cl	(2-) SO_2CH_3	Cl	CH_3
Cl	(2-) SO_2CH_3	Br	CH_3
Cl	(2-) Cl	CF_3	(cyclopropyl)
Cl	(2-) Cl	SCH_3	(cyclopropyl)
Cl	(2-) Cl	SC_2H_5	(cyclopropyl)
Cl	(2-) Cl	SC_3H_7	(cyclopropyl)

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) Cl	SC_3H_7-i	(cyclopropyl)
Cl	(2-) Cl	$SCH_2CH=CH_2$ (allyl)	(cyclopropyl)
Cl	(2-) Cl	$SCH_2C{\equiv}CH$ (propargyl)	(cyclopropyl)
Cl	(2-) Cl	$SCH=CHCH_3$ (propenyl)	(cyclopropyl)
Cl	(2-) Cl	$SC{\equiv}CCH_3$	(cyclopropyl)
Cl	(2-) Cl	SCH_2-cyclopropyl	(cyclopropyl)
Cl	(2-) Cl	$SCH=C=CH_2$	(cyclopropyl)
Cl	(2-) Cl	SCH_2CN	(cyclopropyl)
Cl	(2-) Cl	SCH_2CH_2CN	(cyclopropyl)
Cl	(2-) Cl	OCH_3	(cyclopropyl)
Cl	(2-) Cl	OC_2H_5	(cyclopropyl)

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) Cl	OC_3H_7	(cyclopropylmethyl)
Cl	(2-) Cl	OC_3H_7-i	(cyclopropylmethyl)
Cl	(2-) Cl	OC_4H_9	(cyclopropylmethyl)
Cl	(2-) Cl	OCH_2CF_3	(cyclopropylmethyl)
Cl	(2-) Cl	(cyclopropylmethoxy)	(cyclopropylmethyl)
Cl	(2-) Cl	OC_6H_5	(cyclopropylmethyl)
Cl	(2-) Cl	H	(cyclopropylmethyl)
Cl	(2-) Cl	CH_3	(cyclopropylmethyl)
Cl	(2-) Cl	C_2H_5	(cyclopropylmethyl)
Cl	(2-) Cl	C_3H_7	(cyclopropylmethyl)
Cl	(2-) Cl	C_3H_7-i	(cyclopropylmethyl)
Cl	(2-) Cl	C_4H_9	(cyclopropylmethyl)
Cl	(2-) Cl	C_4H_9-i	(cyclopropylmethyl)

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) Cl	C_4H_9-s	[cyclopropyl]
Cl	(2-) Cl	C_4H_9-t	[cyclopropyl]
Cl	(2-) Cl	[cyclopropyl]	[cyclopropyl]
Cl	(2-) Cl	[cyclopropylmethyl]	[cyclopropyl]
Cl	(2-) Cl	$CH=CHCH_3$	[cyclopropyl]
Cl	(2-) Cl	[phenyl]	[cyclopropyl]
Cl	(2-) Cl	[4-chlorophenyl]	[cyclopropyl]
Cl	(2-) Cl	[benzyl]	[cyclopropyl]
Cl	(2-) Cl	$N(CH_3)_2$	[cyclopropyl]
Cl	(2-) Cl	[N-methylmorpholino]	[cyclopropyl]
Cl	(2-) Cl	Cl	[cyclopropyl]
Cl	(2-) Cl	Br	[cyclopropyl]

R³	(Position) (R⁴)ₙ	R⁵	R⁶
SO_2CH_3	(2-) Cl	CF_3	cyclopropyl
SO_2CH_3	(2-) Cl	SCH_3	cyclopropyl
SO_2CH_3	(2-) Cl	SC_2H_5	cyclopropyl
SO_2CH_3	(2-) Cl	SC_3H_7	cyclopropyl
SO_2CH_3	(2-) Cl	SC_3H_7-i	cyclopropyl
SO_2CH_3	(2-) Cl	$SCH_2CH=CH_2$	cyclopropyl
SO_2CH_3	(2-) Cl	$SCH_2C{\equiv}CH$	cyclopropyl
SO_2CH_3	(2-) Cl	$SCH=CHCH_3$	cyclopropyl
SO_2CH_3	(2-) Cl	$SCH_2CH=C{:}$	cyclopropyl
SO_2CH_3	(2-) Cl	SCH_2-cyclopropyl	cyclopropyl
SO_2CH_3	(2-) Cl	$SCH=C=CH_2$	cyclopropyl

R³	(Position) (R⁴)ₙ	R⁵	R⁶
SO_2CH_3	(2-) Cl	SCH_2CN	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	SCH_2CH_2CN	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OCH_3	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OC_2H_5	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OC_3H_7	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	$OC_3H_7\text{-}i$	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OC_4H_9	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OCH_2CF_3	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	(cyclopropylmethoxymethyl)	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	OC_6H_5	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	H	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	CH_3	(cyclopropylmethyl)
SO_2CH_3	(2-) Cl	C_2H_5	(cyclopropylmethyl)

R³	(Position) (R⁴)ₙ	R⁵	R⁶
SO₂CH₃	(2-) Cl	C₃H₇	cyclopropyl
SO₂CH₃	(2-) Cl	C₃H₇-i	cyclopropyl
SO₂CH₃	(2-) Cl	C₄H₉	cyclopropyl
SO₂CH₃	(2-) Cl	C₄H₉-i	cyclopropyl
SO₂CH₃	(2-) Cl	C₄H₉-s	cyclopropyl
SO₂CH₃	(2-) Cl	C₄H₉-t	cyclopropyl
SO₂CH₃	(2-) Cl	cyclopropyl	cyclopropyl
SO₂CH₃	(2-) Cl	cyclopropylmethyl	cyclopropyl
SO₂CH₃	(2-) Cl	CH=CHCH₃	cyclopropyl
SO₂CH₃	(2-) Cl	phenyl	cyclopropyl
SO₂CH₃	(2-) Cl	4-Cl-phenyl	cyclopropyl
SO₂CH₃	(2-) Cl	benzyl (phenylethyl)	cyclopropyl
SO₂CH₃	(2-) Cl	N(CH₃)₂	cyclopropyl

R^3	(Position) (R^4)$_n$	R^5	R^6
SO$_2$CH$_3$	(2-) Cl	N-methylmorpholine	cyclopropyl
SO$_2$CH$_3$	(2-) Cl	Cl	cyclopropyl
SO$_2$CH$_3$	(2-) Cl	Br	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	CF$_3$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	SCH$_3$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	SC$_2$H$_5$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	SC$_3$H$_7$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	SC$_3$H$_7$-i	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	S-CH$_2$-CH=CH$_2$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	S-CH$_2$-C≡CH	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	S-CH=CH-CH$_3$	cyclopropyl
Cl	(2-) SO$_2$CH$_3$	S-CH=C=CH$_2$	cyclopropyl

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) SO_2CH_3		
Cl	(2-) SO_2CH_3	$SCH=C=CH_2$	
Cl	(2-) SO_2CH_3	SCH_2CN	
Cl	(2-) SO_2CH_3	SCH_2CH_2CN	
Cl	(2-) SO_2CH_3	OCH_3	
Cl	(2-) SO_2CH_3	OC_2H_5	
Cl	(2-) SO_2CH_3	OC_3H_7	
Cl	(2-) SO_2CH_3	OC_3H_7-i	
Cl	(2-) SO_2CH_3	OC_4H_9	
Cl	(2-) SO_2CH_3	OCH_2CF_3	
Cl	(2-) SO_2CH_3		
Cl	(2-) SO_2CH_3	OC_6H_5	
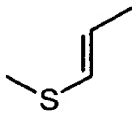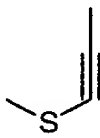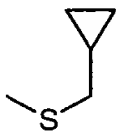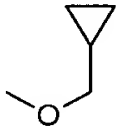
R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) SO₂CH₃	H	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	CH₃	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₂H₅	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₃H₇	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₃H₇-i	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₄H₉	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₄H₉-i	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₄H₉-s	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	C₄H₉-t	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	(cyclopropylmethyl structure)	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	(cyclopropylethyl structure)	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	CH=CHCH₃	(cyclopropylmethyl structure)
Cl	(2-) SO₂CH₃	(tolyl structure)	(cyclopropylmethyl structure)
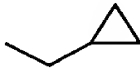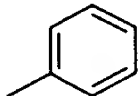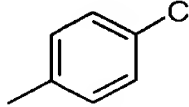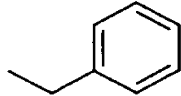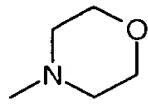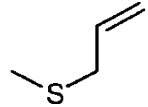
R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) SO₂CH₃	4-chloro-2-methylphenyl	cyclopropyl
Cl	(2-) SO₂CH₃	2-ethylphenyl	cyclopropyl
Cl	(2-) SO₂CH₃	$N(CH_3)_2$	cyclopropyl
Cl	(2-) SO₂CH₃	N-methylmorpholine	cyclopropyl
Cl	(2-) SO₂CH₃	Cl	cyclopropyl
Cl	(2-) SO₂CH₃	Br	cyclopropyl
Cl	(2-) Cl	CF_3	$N(CH_3)_2$
Cl	(2-) Cl	SCH_3	$N(CH_3)_2$
Cl	(2-) Cl	SC_2H_5	$N(CH_3)_2$
Cl	(2-) Cl	SC_3H_7	$N(CH_3)_2$
Cl	(2-) Cl	$SC_3H_7\text{-}i$	$N(CH_3)_2$
Cl	(2-) Cl	allyl-S-CH₃ (—S—CH₂—CH=CH₂)	$N(CH_3)_2$
Cl	(2-) Cl	propargyl-S-CH₃ (—S—CH₂—C≡CH)	$N(CH_3)_2$
Cl	(2-) Cl	propenyl-S-CH₃ (—S—CH=CH—CH₃)	$N(CH_3)_2$

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) Cl	$SCH_2C\equiv CH$	$N(CH_3)_2$
Cl	(2-) Cl	SCH_2-cyclopropyl	$N(CH_3)_2$
Cl	(2-) Cl	$SCH=C=CH_2$	$N(CH_3)_2$
Cl	(2-) Cl	SCH_2CN	$N(CH_3)_2$
Cl	(2-) Cl	SCH_2CH_2CN	$N(CH_3)_2$
Cl	(2-) Cl	OCH_3	$N(CH_3)_2$
Cl	(2-) Cl	OC_2H_5	$N(CH_3)_2$
Cl	(2-) Cl	OC_3H_7	$N(CH_3)_2$
Cl	(2-) Cl	OC_3H_7-i	$N(CH_3)_2$
Cl	(2-) Cl	OC_4H_9	$N(CH_3)_2$
Cl	(2-) Cl	OCH_2CF_3	$N(CH_3)_2$
Cl	(2-) Cl	OCH_2-cyclopropyl	$N(CH_3)_2$
Cl	(2-) Cl	OC_6H_5	$N(CH_3)_2$
Cl	(2-) Cl	H	$N(CH_3)_2$
Cl	(2-) Cl	CH_3	$N(CH_3)_2$
Cl	(2-) Cl	C_2H_5	$N(CH_3)_2$
Cl	(2-) Cl	C_3H_7	$N(CH_3)_2$
Cl	(2-) Cl	C_3H_7-i	$N(CH_3)_2$
Cl	(2-) Cl	C_4H_9	$N(CH_3)_2$
Cl	(2-) Cl	C_4H_9-i	$N(CH_3)_2$
Cl	(2-) Cl	C_4H_9-s	$N(CH_3)_2$
Cl	(2-) Cl	C_4H_9-t	$N(CH_3)_2$
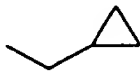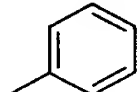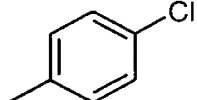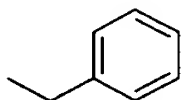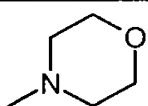
R^3	(Position) $(R^4)_n$	R^5	R^6
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl	$CH=CHCH_3$	$N(CH_3)_2$
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl	$N(CH_3)_2$	$N(CH_3)_2$
Cl	(2-) Cl		$N(CH_3)_2$
Cl	(2-) Cl	Cl	$N(CH_3)_2$
Cl	(2-) Cl	Br	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	CF_3	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	SCH_3	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	SC_2H_5	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	SC_3H_7	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	SC_3H_7-i	$N(CH_3)_2$
SO_2CH_3	(2-) Cl		$N(CH_3)_2$
SO_2CH_3	(2-) Cl		$N(CH_3)_2$

R^3	(Position) (R^4)$_n$	R^5	R^6
SO$_2$CH$_3$	(2-) Cl	S–CH=CH–CH$_3$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	S–CH=CH$_2$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	S–CH$_2$–cyclopropyl	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	SCH=C=CH$_2$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	SCH$_2$CN	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	SCH$_2$CH$_2$CN	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OCH$_3$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OC$_2$H$_5$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OC$_3$H$_7$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OC$_3$H$_7$-i	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OC$_4$H$_9$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OCH$_2$CF$_3$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	O–CH$_2$–cyclopropyl	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	OC$_6$H$_5$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	H	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	CH$_3$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	C$_2$H$_5$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	C$_3$H$_7$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	C$_3$H$_7$-i	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	C$_4$H$_9$	N(CH$_3$)$_2$
SO$_2$CH$_3$	(2-) Cl	C$_4$H$_9$-i	N(CH$_3$)$_2$

R^3	(Position) $(R^4)_n$	R^5	R^6
SO_2CH_3	(2-) Cl	C_4H_9-s	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	C_4H_9-t	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	cyclopropyl (structure)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	cyclopropylmethyl (structure)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	$CH=CHCH_3$	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	4-methylphenyl (structure)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	4-chloro-methylphenyl (structure, Cl)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	phenylethyl (structure)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	$N(CH_3)_2$	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	N-methylmorpholin-yl (structure)	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	Cl	$N(CH_3)_2$
SO_2CH_3	(2-) Cl	Br	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	CF_3	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SCH_3	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SC_2H_5	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SC_3H_7	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SC_3H_7-i	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	allylthiomethyl (structure, S)	$N(CH_3)_2$

R^3	(Position) $(R^4)_n$	R^5	R^6
Cl	(2-) SO_2CH_3	$SCH_2C{\equiv}CH$	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	$SCH=CHCH_3$	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	$SC(CH_3)=CH_2$	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SCH_2-cyclopropyl	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	$SCH=C=CH_2$	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SCH_2CN	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	SCH_2CH_2CN	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OCH_3	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OC_2H_5	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OC_3H_7	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OC_3H_7-i	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OC_4H_9	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OCH_2CF_3	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OCH_2-cyclopropyl	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	OC_6H_5	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	H	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	CH_3	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	C_2H_5	$N(CH_3)_2$
Cl	(2-) SO_2CH_3	C_3H_7	$N(CH_3)_2$

R³	(Position) (R⁴)ₙ	R⁵	R⁶
Cl	(2-) SO₂CH₃	C₃H₇-i	N(CH₃)₂
Cl	(2-) SO₂CH₃	C₄H₉	N(CH₃)₂
Cl	(2-) SO₂CH₃	C₄H₉-i	N(CH₃)₂
Cl	(2-) SO₂CH₃	C₄H₉-s	N(CH₃)₂
Cl	(2-) SO₂CH₃	C₄H₉-t	N(CH₃)₂
Cl	(2-) SO₂CH₃	(cyclopropyl)	N(CH₃)₂
Cl	(2-) SO₂CH₃	(cyclopropylmethyl)	N(CH₃)₂
Cl	(2-) SO₂CH₃	CH=CHCH₃	N(CH₃)₂
Cl	(2-) SO₂CH₃	(methylphenyl)	N(CH₃)₂
Cl	(2-) SO₂CH₃	(chloro-methylphenyl)	N(CH₃)₂
Cl	(2-) SO₂CH₃	(ethylphenyl)	N(CH₃)₂
Cl	(2-) SO₂CH₃	N(CH₃)₂	N(CH₃)₂
Cl	(2-) SO₂CH₃	(N-methylmorpholine)	N(CH₃)₂
Cl	(2-) SO₂CH₃	Cl	N(CH₃)₂
Cl	(2-) SO₂CH₃	Br	N(CH₃)₂
Cl	(2-) Cl	CH₃	OCH₃
Cl	(2-) Cl	C₂H₅	OCH₃
Cl	(2-) Cl	C₃H₇	OCH₃
Cl	(2-) Cl	SCH₃	OCH₃
Cl	(2-) Cl	SC₂H₅	OCH₃
Cl	(2-) Cl	OCH₃	OCH₃

R^3	(Position) $(R^4)_n$	R^5	R^6
Cl	(2-) Cl	OC_2H_5	OCH_3
Cl	(2-) Cl	CH_3	OC_2H_5
Cl	(2-) Cl	C_2H_5	OC_2H_5
Cl	(2-) Cl	C_3H_7	OC_2H_5
Cl	(2-) Cl	SCH_3	OC_2H_5
Cl	(2-) Cl	SC_2H_5	OC_2H_5
Cl	(2-) Cl	OCH_3	OC_2H_5
Cl	(2-) Cl	OC_2H_5	OC_2H_5
Cl	(2-) SO_2CH_3	CH_3	OCH_3
Cl	(2-) SO_2CH_3	C_2H_5	OCH_3
Cl	(2-) SO_2CH_3	C_3H_7	OCH_3
Cl	(2-) SO_2CH_3	SCH_3	OCH_3
Cl	(2-) SO_2CH_3	SC_2H_5	OCH_3
Cl	(2-) SO_2CH_3	OCH_3	OCH_3
Cl	(2-) SO_2CH_3	OC_2H_5	OCH_3
Cl	(2-) SO_2CH_3	CH_3	OC_2H_5
Cl	(2-) SO_2CH_3	C_2H_5	OC_2H_5
Cl	(2-) SO_2CH_3	C_3H_7	OC_2H_5
Cl	(2-) SO_2CH_3	SCH_3	OC_2H_5
Cl	(2-) SO_2CH_3	SC_2H_5	OC_2H_5
Cl	(2-) SO_2CH_3	OCH_3	OC_2H_5
Cl	(2-) SO_2CH_3	OC_2H_5	OC_2H_5
SO_2CH_3	(2-) Cl	Cl	OCH_3
SO_2CH_3	(2-) Cl	Br	OCH_3
SO_2CH_3	(2-) Cl	CH_3	OCH_3
SO_2CH_3	(2-) Cl	C_2H_5	OCH_3
SO_2CH_3	(2-) Cl	C_3H_7	OCH_3
SO_2CH_3	(2-) Cl	SCH_3	OCH_3
SO_2CH_3	(2-) Cl	SC_2H_5	OCH_3

R^3	(Position) (R^4)$_n$	R^5	R^6
SO$_2$CH$_3$	(2-) Cl	OCH$_3$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	OC$_2$H$_5$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	CH$_3$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	C$_2$H$_5$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	C$_3$H$_7$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	SCH$_3$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	SC$_2$H$_5$	OC$_2$H$_5$
SO$_2$CH$_3$	(2-) Cl	OCH$_3$	OC$_2$H$_5$

Group 5

(IB-2)
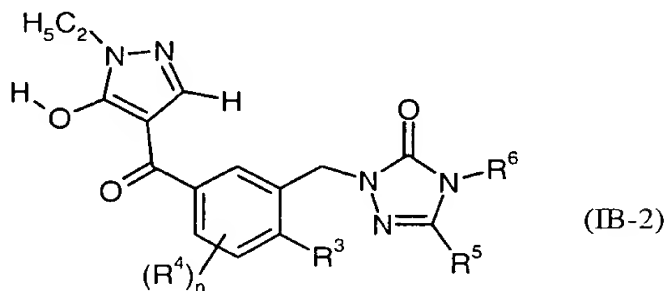
Here, R^3, (R^4)$_n$, R^5 and R^6 have, for example, the meanings given above in group 4.

Group 6

(IB-3)
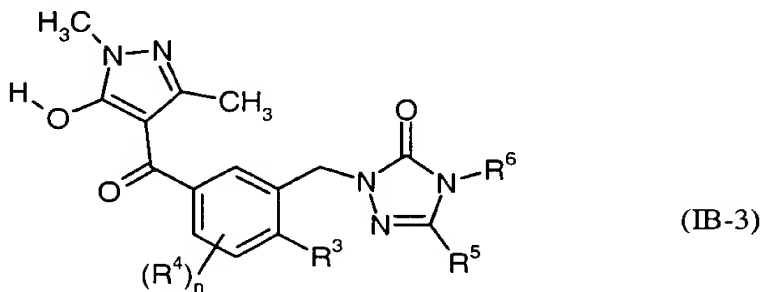
Here, R^3, (R^4)$_n$, R^5 and R^6 have, for example, the meanings given above in group 4.

Group 7
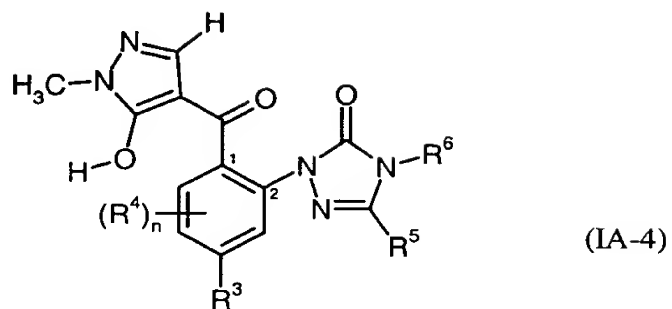
(IA-4)

5   Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given in the table below:

R^3	(Position) $(R^4)_n$	R^5	R^6
H	-	CF_3	CH_3
F	-	CF_3	CH_3
Cl	-	CF_3	CH_3
Br	-	CF_3	CH_3

Group 8

(IA-5)
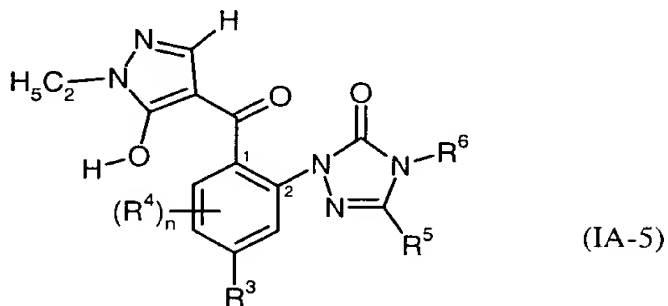
10

Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 7.

Group 9

(IA-6)
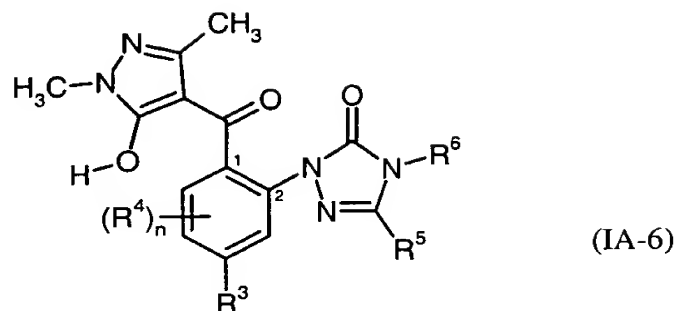
Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 7.

5

Group 10

(IC-1)
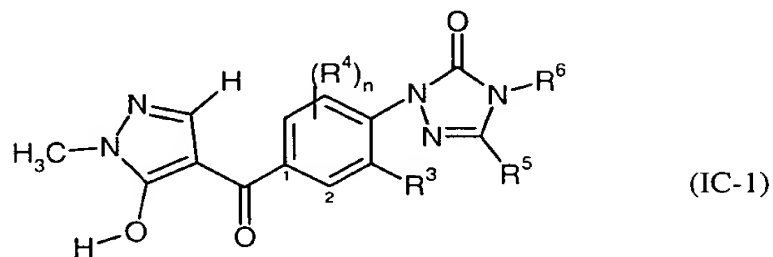
Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given in the table below:

10

R^3	(Position) $(R^4)_n$	R^5	R^6
H	(2-) F	CF_3	CH_3
H	(2-) Cl	CF_3	CH_3
H	(2-) Br	CF_3	CH_3
H	-	CF_3	CH_3

Group 11
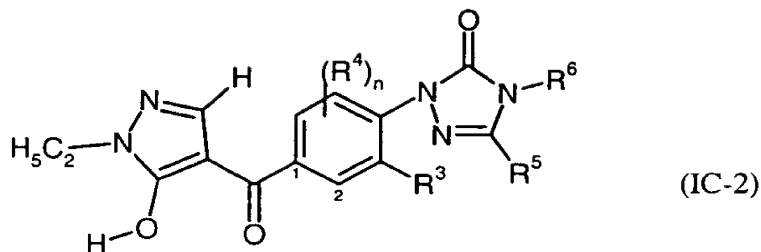
(IC-2)

Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 10.

Group 12

(IC-3)
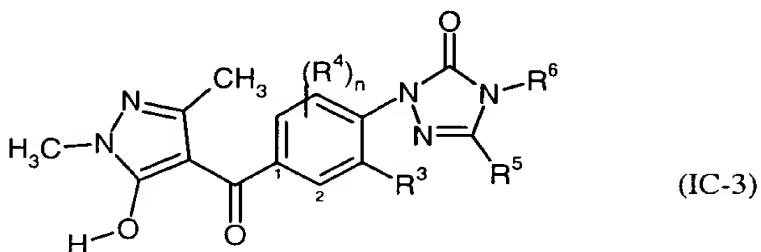
Here, R^3, $(R^4)_n$, R^5 and R^6 have, for example, the meanings given above in group 10.

The novel substituted benzoylpyrazoles of the general formula (I) have strong and selective herbicidal activity.

Novel substituted benzoylpyrazoles of the general formula (I) are obtained when

(a) pyrazoles of the general formula (II)
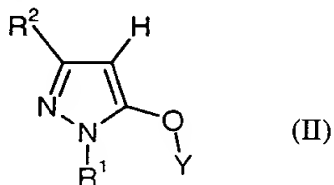
(II)

in which

R^1, R^2 and Y are as defined above,

- 50 -

are reacted with substituted benzoic acids of the general formula (III),

(III)
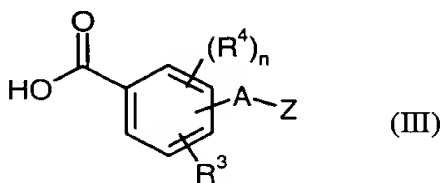
5   in which

n, A, R^3, R^4 and Z are as defined above,

in the presence of a dehydrating agent, if appropriate in the presence of one or more

10   reaction auxiliaries and if appropriate in the presence of a diluent,

or when

(b) pyrazoles of the general formula (II)
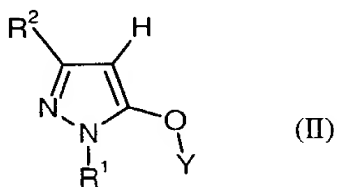
15

(II)

in which

R^1, R^2 and Y are as defined above,

20

are reacted with substituted benzoic acid derivatives of the general formula (IV)

(IV)
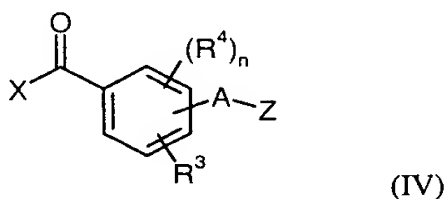
in which

n, A, R^3, R^4 and Z are as defined above, and

5   X   represents cyano, halogen or alkoxy,

- or with corresponding carboxylic anhydrides -

if appropriate in the presence of one or more reaction auxiliaries and if appropriate in

10   the presence of a diluent,

or when

(c) substituted benzoylpyrazoles of the general formula (Ia)

15

(Ia)
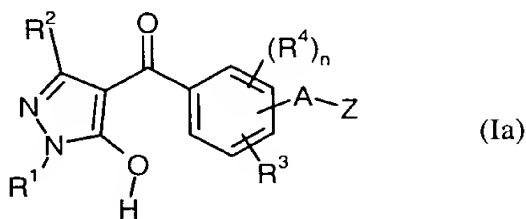
in which

n, A, R^1, R^2, R^3, R^4 and Z are as defined above,

20

are reacted with compounds of the general formula (V)

H-Y   (V)

in which

25

Y   is as defined above, except for hydrogen,

- or, if appropriate, with corresponding isocyanates or isothiocyanates -

if appropriate in the presence of one or more reaction auxiliaries and if appropriate in the presence of a diluent,

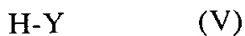5   and, if appropriate, the resulting compounds of the formula (I) are subsequently subjected in a customary manner to electrophilic or nucleophilic and/or oxidation or reduction reactions within the scope of the definition of the substituents, or the compounds of the formula (I) are converted in a customary manner into salts.

10   The compounds of the formula (I) can be converted by customary methods into other compounds of the formula (I) in accordance with the above definition, for example by nucleophilic substitution (for example R^5: Cl $\rightarrow$ OC_2H_5, SCH_3) or by oxidation (for example R^5: CH_2SCH_3 $\rightarrow$ $CH_2S(O)CH_3$).

15   Using, for example, 3-chloro-5-hydroxy-1-methyl-pyrazole and 2-(3-carboxy-5-fluoro-benzyl)-5-ethyl-4-methoxy-2,4-dihydro-3H-1,2,4-triazol-3-one as starting materials, the course of the reaction in the process (a) according to the invention can be illustrated by the following formula scheme:
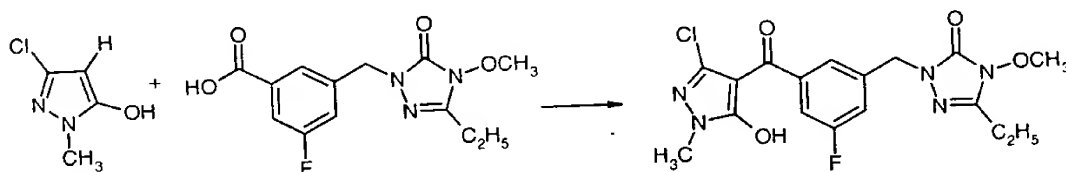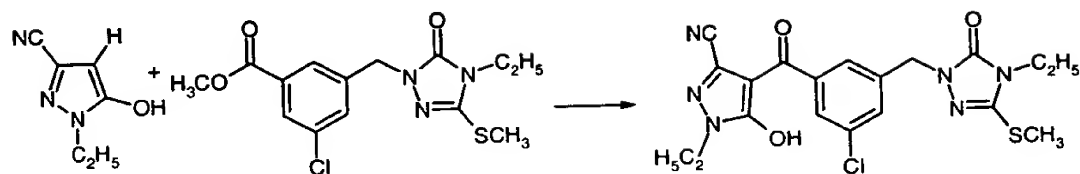

20

Using, for example, 3-cyano-5-hydroxy-1-ethyl-pyrazole and 2-(3-methoxycarbonyl-5-chloro-benzyl)-4-ethyl-5-methylthio-2,4-dihydro-3H-1,2,4-triazol-3-one as starting materials, the course of the reaction in the process (b) according to the invention can be illustrated by the following formula scheme:

25

Using, for example, 4-methyl-5-trifluoromethyl-2-[3-chloro-4-(1-ethyl-5-hydroxy-pyrazol-4-yl-carbonyl)-phenyl]-2,4-dihydro-3H-1,2,4-triazol-3-one and benzoyl chloride as starting materials, the course of the reaction in the process (c) according to the invention can be illustrated by the following formula scheme:
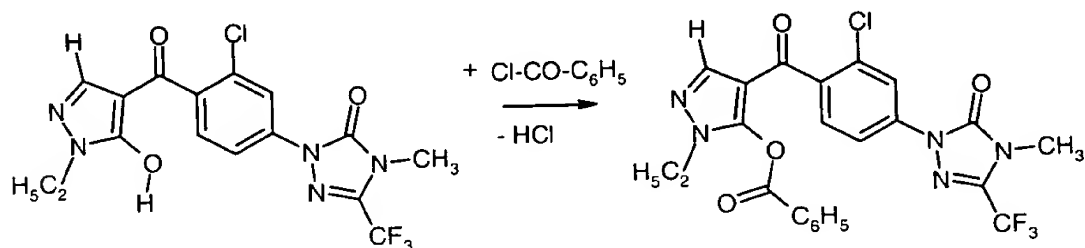
The formula (II) provides a general definition of the pyrazoles to be used as starting materials in the process (a) according to the invention for preparing compounds of the general formula (I). In the general formula (II), R^1, R^2 and Y preferably have those meanings which have already been mentioned above, in connection with the description of the compounds of the general formula (I) according to the invention, as being preferred, particularly preferred, very particularly preferred or most preferred for R^1, R^2 and Y.

The starting materials of the general formula (II) are known and/or can be prepared by processes known per se (cf. EP-A-240001).
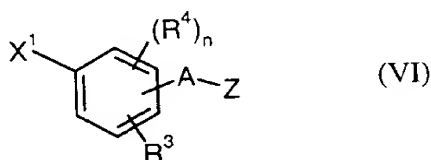
The formula (III) provides a general definition of the benzoic acids further to be used as starting materials in the process (a) according to the invention. In the formula (III), n, A, R^3, R^4 and Z preferably have those meanings which have already been mentioned above, in connection with the description of the compounds of the

formula (I) according to the invention, as being preferred, particularly preferred, very particularly preferred or most preferred for n, A, R^3, R^4 and Z.

Except for 2-(5-carboxy-2,4-dichlorophenyl)-4-difluoromethyl-5-methyl-2,4-dihydro-3H-1,2,4-triazol-3-one - alias 2,4-dichloro-5-(4-difluoromethyl-4,5-dihydro-3-methyl-5-oxo-1H-1,2,4-triazol-1-yl)-benzoic acid (CAS Reg. No. 90208-77-8) and 2-(5-carboxy-2,4-dichloro-phenyl)-4,5-dimethyl-2,4-dihydro-3H-1,2,4-triazol-3-one - alias 2,4-dichloro-5-(4,5-dihydro-3,4-dimethyl-5-oxo-1H-1,2,4-triazol-1-yl)-benzoic acid (CAS Reg. No. 90208-76-7) - the starting materials of the general formula (III) have hitherto not been disclosed in the literature. However, except for 2-(5-carboxy-2,4-dichloro-phenyl)-4-difluoromethyl-5-methyl-2,4-dihydro-3H-1,2,4-triazol-3-one and 2-(5-carboxy-2,4-dichloro-phenyl)-4,5-dimethyl-2,4-dihydro-3H-1,2,4-triazol-3-one (cf. JP-A-58225070 - cited in Chem. Abstracts 100:209881, JP-A-02015069 - cited in Chem. Abstracts 113:23929), they are the subject of an earlier application which, however, has not been published earlier (cf. DE-A-19833360).

The substituted benzoic acids of the general formula (III) are obtained when benzoic acid derivatives of the general formula (VI)

(VI)

in which

n, A, R^3 and R^4 and Z are as defined above, and

X^1 represents cyano, carbamoyl, halogenocarbonyl or alkoxycarbonyl,

are reacted with water, if appropriate in the presence of a hydrolysis auxiliary, such as, for example, sulfuric acid, at temperatures between 50°C and 120°C (cf. the Preparation Examples).

The formula (IV) provides a general definition of the substituted benzoic acid derivatives to be used as starting materials in the process (b) according to the invention for preparing compounds of the general formula (I). In the general formula (IV), n, A, R^3, R^4 and Z preferably have those meanings which have already been mentioned above, in connection with the description of the compounds of the general formula (I) according to the invention, as being preferred, particularly preferred, very particularly preferred or most preferred for n, A, R^3, R^4 and Z; X preferably represents cyano, fluorine, chlorine, bromine or C_1-C_4-alkoxy, in particular chlorine, methoxy or ethoxy.

The starting materials of the general formula (IV) – and the precursors of the general formula (VI) - are known and/or can be prepared by processes known per se (cf. DE-A-3839480, DE-A-4239296, EP-A-597360, EP-A-609734, DE-A-4303676, EP-A-617026, DE-A-4405614, US-A-5378681).

The formula (Ia) provides a general definition of the substituted benzoylpyrazoles to be used as starting materials in the process (c) according to the invention for preparing compounds of the general formula (I). In the general formula (Ia), n, A, R^1, R^2, R^3, R^4 and Z preferably have those meanings which have already been mentioned above, in connection with the description of the compounds of the general formula (I) according to the invention, as being preferred, particularly preferred, very particularly preferred or most preferred for n, A, R^1, R^2, R^3, R^4 and Z.

The starting materials of the general formula (Ia) are novel compounds according to the invention; they can be prepared by the processes (a) and (b) according to the invention.

The formula (V) provides a general definition of the compounds further to be used as starting materials in the [lacuna] (c) according to the invention. In the general formula (V), Y preferably has that meaning which has already been mentioned above, in connection with the description of the compounds of the general formula (I)

according to the invention, as being preferred, particularly preferred, very particularly preferred or most preferred for Y.

The starting materials of the general formula (V) are known chemicals for synthesis.

5

The process (a) according to the invention for preparing the novel substituted benzoylpyrazoles of the general formula (I) is carried out using a dehydrating agent. Suitable dehydrating agents are the customary chemicals suitable for binding water.

10 Examples which may be mentioned are dicyclohexylcarbodiimide and carbonyl-bis-imidazole.

A particularly suitable dehydrating agent which may be mentioned is dicyclohexylcarbodiimide.

15

The process (a) according to the invention for preparing the novel substituted benzoylpyrazoles of the general formula (I) is, if appropriate, carried out using a reaction auxiliary.

20 Examples of suitable reaction auxiliaries which may be mentioned are sodium cyanide, potassium cyanide, acetone cyanohydrin, 2-cyano-2-(trimethylsilyloxy)-propane and trimethylsilyl cyanide.

A particularly suitable reaction auxiliary which may be mentioned is trimethylsilyl

25 cyanide.

The process (b) according to the invention for preparing the novel substituted benzoylpyrazoles of the general formula (I) is, if appropriate, carried out using reaction auxiliaries.

30

Examples of suitable reaction auxiliaries which may be mentioned are (conc.) sulfuric acid, zinc chloride, aluminum chloride, and boron fluoride.

The processes according to the invention for preparing the novel substituted benzoylpyrazoles of the general formula (I) are, if appropriate, carried out using further reaction auxiliaries. Suitable (further) reaction auxiliaries for the processes according to the invention are, in general, basic organic nitrogen compounds, such as, for example, trimethylamine, triethylamine, tripropylamine, tributylamine, ethyl-di-isopropylamine, N,N-dimethyl-cyclohexylamine, dicyclohexylamine, ethyl-dicyclo-hexylamine, N,N-dimethyl-aniline, N,N-dimethyl-benzylamine, pyridine, 2-methyl-, 3-methyl-, 4-methyl-, 2,4-dimethyl-, 2,6-dimethyl-, 3,4-dimethyl- and 3,5-dimethyl-pyridine, 5-ethyl-2-methyl-pyridine, 4-dimethylamino-pyridine, N-methyl-piperidine, 1,4-diazabicyclo[2,2,2]-octane (DABCO), 1,5-diazabicyclo[4,3,0]-non-5-ene (DBN), or 1,8-diazabicyclo[5,4,0]-undec-7-ene (DBU).

Suitable diluents for carrying out the processes (a), (b) and (c) according to the invention are especially inert organic solvents. These include, in particular, aliphatic, alicyclic or aromatic, optionally halogenated hydrocarbons, such as, for example, benzine, benzene, toluene, xylene, chlorobenzene, dichlorobenzene, petroleum ether, hexane, cyclohexane, dichloromethane, chloroform, carbon tetrachloride or 1,4-dichloro-ethane; ethers, such as diethyl ether, diisopropyl ether, dioxane, tetrahydrofuran, ethylene glycol dimethyl ether or ethylene glycol diethyl ether; ketones, such as acetone, butanone or methyl isobutyl ketone; nitriles, such as acetonitrile, propionitrile or butyronitrile; amides, such as N,N-dimethylformamide, N,N-dimethylacetamide, N-methyl-formanilide, N-methyl-pyrrolidone or hexa-methylphosphoric triamide; esters such as methyl acetate or ethyl acetate, sulfoxides, such as dimethyl sulfoxide.

When carrying out the processes (a), (b) and (c) according to the invention, the reaction temperatures can be varied within a relatively wide range. In general, the processes are carried out at temperatures between 0°C and 150°C, preferably between 10°C and 120°C.

The processes (a), (b) and (c) according to the invention are generally carried out under atmospheric pressure. However, it is also possible to carry out the processes according to the invention under elevated or reduced pressure - in general between 0.1 bar and 10 bar.

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For carrying out the processes (a), (b) and (c) according to the invention, the starting materials are generally employed in approximately equimolar amounts. However, it is also possible to use a relatively large excess of one of the components. The reaction is generally carried out in a suitable diluent in the presence of a dehydrating agent, and the reaction mixture is generally stirred for a number of hours at the required temperature. Work-up is carried out by customary methods (cf. the Preparation Examples).

The active compounds according to the invention can be used as defoliants, desiccants, haulm killers and, especially, as weed killers. By weeds in the broadest sense there are to be understood all plants which grow in locations where they are undesired. Whether the substances according to the invention act as total or selective herbicides depends essentially on the amount used.

According to the invention, it is possible to treat all plants and parts of plants. By plants are understood here all plants and plant populations such as desired and undesired wild plants or crop plants (including naturally occurring crop plants). Crop plants can be plants which can be obtained by conventional breeding and optimization methods or by biotechnological and genetic engineering methods or combinations of these methods, including transgenic plants and including plant varieties which may or may not be protected by plant variety protection rights. Parts of plants are to be understood as meaning all above-ground and below-ground parts and organs of plants, such as shoot, leaf, flower and root, examples which may be mentioned being leaves, needles, stems, trunks, flowers, fruit-bodies, fruits and seeds and also roots, tubers and rhizomes. Parts of plants also include crops, and vegetative and generative propagation material, for example seedlings, tubers, rhizomes, cuttings and seeds.

The treatment of the plants and parts of plants according to the invention with the active compounds is carried out directly or by action on their environment, habitat or storage area according to customary treatment methods, for example by dipping, spraying, evaporating, atomizing, broadcasting, brushing-on and, in the case of propagation material, in particular in the case of seeds, furthermore by single- or multi-layer coating.

The active compounds according to the invention can be used, for example, in connection with the following plants:

Dicotyledonous weeds of the genera: Sinapis, Lepidium, Galium, Stellaria, Matricaria, Anthemis, Galinsoga, Chenopodium, Urtica, Senecio, Amaranthus, Portulaca, Xanthium, Convolvulus, Ipomoea, Polygonum, Sesbania, Ambrosia, Cirsium, Carduus, Sonchus, Solanum, Rorippa, Rotala, Lindernia, Lamium, Veronica, Abutilon, Emex, Datura, Viola, Galeopsis, Papaver, Centaurea, Trifolium, Ranunculus, Taraxacum.

Dicotyledonous crops of the genera: Gossypium, Glycine, Beta, Daucus, Phaseolus, Pisum, Solanum, Linum, Ipomoea, Vicia, Nicotiana, Lycopersicon, Arachis, Brassica, Lactuca, Cucumis, Cucurbita.

Monocotyledonous weeds of the genera: Echinochloa, Setaria, Panicum, Digitaria, Phleum, Poa, Festuca, Eleusine, Brachiaria, Lolium, Bromus, Avena, Cyperus, Sorghum, Agropyron, Cynodon, Monochoria, Fimbristylis, Sagittaria, Eleocharis, Scirpus, Paspalum, Ischaemum, Sphenoclea, Dactyloctenium, Agrostis, Alopecurus, Apera, Aegilops, Phalaris.

Monocotyledonous crops of the genera: Oryza, Zea, Triticum, Hordeum, Avena, Secale, Sorghum, Panicum, Saccharum, Ananas, Asparagus, Allium.

However, the use of the active compounds according to the invention is in no way restricted to these genera, but also extends in the same manner to other plants.

Depending on the concentration, the active compounds according to the invention are suitable for total weed control, for example on industrial terrain and rail tracks and on paths and areas with or without tree growth. Equally, the active compounds according to the invention can be employed for controlling weeds in perennial crops, for example forests, ornamental tree plantings, orchards, vineyards, citrus groves, nut orchards, banana plantations, coffee plantations, tea plantations, rubber plantations, oil palm plantations, cocoa plantations, soft fruit plantings and hop fields, on lawns and turf and pastures and for selective weed control in annual crops.

The compounds of the formula (I) according to the invention have strong herbicidal activity and a broad activity spectrum when applied to the soil and on above-ground parts of plants. To a certain extent, they are also suitable for selective control of monocotyledonous and dicotyledonous weeds in monocotyledonous and dicotyledonous crops, both by the pre-emergence and by the post-emergence method.

The active compounds can be converted into the customary formulations, such as solutions, emulsions, wettable powders, suspensions, powders, dusts, pastes, soluble powders, granules, suspoemulsion concentrates, natural and synthetic substances impregnated with active compound, and microencapsulations in polymeric substances.

These formulations are produced in a known manner, for example by mixing the active compounds with extenders, that is to say liquid solvents and/or solid carriers, optionally with the use of surfactants, that is to say emulsifiers and/or dispersants and/or foam formers.

If the extender used is water, it is also possible to use, for example, organic solvents as auxiliary solvents. Liquid solvents which are mainly suitable are: aromatics, such as xylene, toluene or alkylnaphthalenes, chlorinated aromatics and chlorinated

aliphatic hydrocarbons, such as chlorobenzenes, chloroethylenes or methylene chloride, aliphatic hydrocarbons, such as cyclohexane or paraffins, for example petroleum fractions, mineral and vegetable oils, alcohols, such as butanol or glycol, and also their ethers and esters, ketones, such as acetone, methyl ethyl ketone, methyl isobutyl ketone or cyclohexanone, strongly polar solvents, such as dimethylformamide and dimethyl sulfoxide, and water.

Suitable solid carriers are: for example ammonium salts and ground natural minerals, such as kaolins, clays, talc, chalk, quartz, attapulgite, montmorillonite or diatomaceous earth, and ground synthetic minerals, such as finely divided silica, alumina and silicates; suitable solid carriers for granules are: for example crushed and fractionated natural rocks, such as calcite, marble, pumice, sepiolite, dolomite and synthetic granules of inorganic and organic meals, and granules of organic material, such as sawdust, coconut shells, maize cobs and tobacco stalks; suitable emulsifiers and/or foam formers are: for example nonionic and anionic emulsifiers such as polyoxyethylene fatty acid esters, polyoxyethylene fatty alcohol ethers, for example alkylaryl polyglycol ethers, alkylsulfonates, alkyl sulfates, arylsulfonates and protein hydrolyzates; suitable dispersants are: for example lignosulfite waste liquors and methylcellulose.

Tackifiers, such as carboxymethylcellulose, natural and synthetic polymers in the form of powders, granules or latices, such as gum arabic, polyvinyl alcohol and polyvinyl acetate, and also natural phospholipids, such as cephalins and lecithins, and synthetic phospholipids can be used in the formulations. Other possible additives are mineral and vegetable oils.

It is possible to use colorants, such as inorganic pigments, for example iron oxide, titanium oxide, Prussian blue, and organic dyes, such as alizarin dyes, azo dyes and metal phthalocyanine dyes, and trace nutrients, such as salts of iron, manganese, boron, copper, cobalt, molybdenum and zinc.

The formulations generally comprise between 0.1 and 95 per cent by weight of active compound, preferably between 0.5 and 90%.

For controlling weeds, the active compounds according to the invention, as such or in the form of their formulations, can also be used as mixtures with known herbicides, finished formulations or tank mixes being possible.

Possible components for the mixtures are known herbicides, for example

acetochlor, acifluorfen(-sodium), aclonifen, alachlor, alloxydim(-sodium), ametryne, amidochlor, amidosulfuron, anilofos, asulam, atrazine, azafenidin, azimsulfuron, benazolin(-ethyl), benfuresate, bensulfuron(-methyl), bentazon, benzofenap, benzoyl-prop(-ethyl), bialaphos, bifenox, bispyribac(-sodium), bromobutide, bromofenoxim, bromoxynil, butachlor, butroxydim, butylate, cafenstrole, caloxydim, carbetamide, carfentrazone(-ethyl), chlomethoxyfen, chloramben, chloridazon, chlorimuron(-ethyl), chlornitrofen, chlorsulfuron, chlortoluron, cinidon(-ethyl), cinmethylin, cino-sulfuron, clethodim, clodinafop(-propargyl), clomazone, clomeprop, clopyralid, clo-pyrasulfuron(-methyl), cloransulam(-methyl), cumyluron, cyanazine, cybutryne, cycloate, cyclosulfamuron, cycloxydim, cyhalofop(-butyl), 2,4-D, 2,4-DB, 2,4-DP, desmedipham, diallate, dicamba, diclofop(-methyl), diclosulam, diethatyl(-ethyl), di-fenzoquat, diflufenican, diflufenzopyr, dimefuron, dimepiperate, dimethachlor, di-methametryn, dimethenamid, dimexyflam, dinitramine, diphenamid, diquat, dithio-pyr, diuron, dymron, epoprodan, EPTC, esprocarb, ethalfluralin, ethametsulfuron(-methyl), ethofumesate, ethoxyfen, ethoxysulfuron, etobenzanid, fenoxaprop(-P-ethyl), flamprop(-isopropyl), flamprop(-isopropyl-L), flamprop(-methyl), flazasulfuron, fluazifop(-P-butyl), fluazolate, flucarbazone, flufenacet, flumetsulam, flumiclorac(-pentyl), flumioxazin, flumipropyn, flumetsulam, fluometuron, fluoro-chloridone, fluoroglycofen(-ethyl), flupoxam, flupropacil, flurpyrsulfuron(-methyl, -sodium), flurenol(-butyl), fluridone, fluroxypyr(-meptyl), flurprimidol, flurtamone, fluthiacet(-methyl), fluthiamide, fomesafen, glufosinate(-ammonium), glyphosate(-isopropylammonium), halosafen, haloxyfop(-ethoxyethyl), haloxyfop(-P-methyl), hexazinone, imazamethabenz(-methyl), imazamethapyr, imazamox, imazapic, imaza-

pyr, imazaquin, imazethapyr, imazosulfuron, iodosulfuron, ioxynil, isopropalin, isoproturon, isouron, isoxaben, isoxachlortole, isoxaflutole, isoxapyrifop, lactofen, lenacil, linuron, MCPA, MCPP, mefenacet, mesotrione, metamitron, metazachlor, methabenzthiazuron, metobenzuron, metobromuron, (alpha-)metolachlor, meto-sulam, metoxuron, metribuzin, metsulfuron(-methyl), molinate, monolinuron, napro-anilide, napropamide, neburon, nicosulfuron, norflurazon, orbencarb, oryzalin, oxadiargyl, oxadiazon, oxasulfuron, oxaziclomefone, oxyfluorfen, paraquat, pelargonsäure, pendimethalin, pentoxazone, phenmedipham, picolinafen, piperophos, pretilachlor, primisulfuron(-methyl), prometryn, propachlor, propanil, propaquizafop, propisochlor, propyzamide, prosulfocarb, prosulfuron, pyraflufen(-ethyl), pyrazolate, pyrazosulfuron(-ethyl), pyrazoxyfen, pyribenzoxim, pyributicarb, pyridate, pyriminobac(-methyl), pyrithiobac(-sodium), quinchlorac, quinmerac, quinoclamine, quizalofop(-P-ethyl), quizalofop(-P-tefuryl), rimsulfuron, sethoxydim, simazine, simetryn, sulcotrione, sulfentrazone, sulfometuron(-methyl), sulfosate, sulfosulfuron, tebutam, tebuthiuron, tepraloxydim, terbuthylazine, terbutryn, thenylchlor, thiafluamide, thiazopyr, thidiazimin, thifensulfuron(-methyl), thiobencarb, tiocarb-azil, tralkoxydim, triallate, triasulfuron, tribenuron(-methyl), triclopyr, tridiphane, tri-fluralin, triflusulfuron and tritosulfuron.

A mixture with other known active compounds, such as fungicides, insecticides, acaricides, nematicides, bird repellents, plant nutrients and agents which improve soil structure, is also possible.

The active compounds can be used as such, in the form of their formulations or in the use forms prepared therefrom by further dilution, such as ready-to-use solutions, suspensions, emulsions, powders, pastes and granules. They are used in the customary manner, for example by watering, spraying, atomizing, scattering.

The active compounds according to the invention can be applied both before and after emergence of the plants. They can also be incorporated into the soil before sowing.

The amount of active compound used can vary within a relatively wide range. It depends essentially on the nature of the desired effect. In general, the amounts used are between 1 g and 10 kg of active compound per hectare of soil surface, preferably between 5 g and 5 kg per ha.

The preparation and the use of the active compounds according to the invention can be seen from the examples below.

Preparation Examples:

Example 1
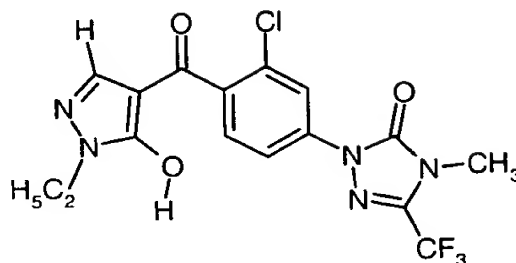
At room temperature (about 20°C), a mixture of 1.64 g (5 mmol) of 4-methyl-5-trifluoromethyl-2-(3-chloro-4-carboxy-phenyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 0.62 g (5.5 mmol) of 1-ethyl-5-hydroxy-pyrazole and 40 ml of acetonitrile is admixed with stirring with 1.13 g (5.5 mmol) of dicyclohexylcarbodiimide, and the reaction mixture is stirred at room temperature for 16 hours. 1.0 g (10 mmol) of triethylamine and 0.2 g (2 mmol) of trimethylsilyl cyanide are then added, and the mixture is stirred at room temperature for three days. 60 ml of a 2% strength aqueous sodium carbonate solution are then added, and the mixture is stirred at room temperature for three hours. The precipitated dicyclohexylurea is removed by filtration with suction, and the mother liquor is extracted twice with diethyl ether. With stirring, the aqueous phase is adjusted by addition of conc. hydrochloric acid to a pH of about 1. The oily product that separates off during the addition is extracted with methylene chloride, and the extraction solution is dried with magnesium sulfate and filtered. From the filtrate, the solvent is carefully distilled off under waterpump vacuum.

This gives 1.5 g (72% of theory) of 4-methyl-5-trifluoromethyl-2-[3-chloro-4-(1-ethyl-5-hydroxy-pyrazol-4-yl-carbonyl)-phenyl]-2,4-dihydro-3H-1,2,4-triazol-3-one as an amorphous product.

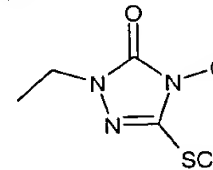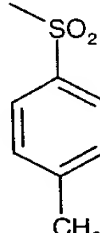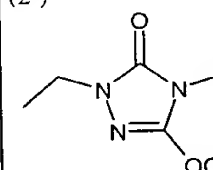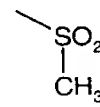logP (determined at pH≈2): 2.63.

Similarly to Example 1, and in accordance with the general description of the preparation process according to the invention, it is also possible to prepare, for example, the compounds of the general formula (I) - or those of the formulae (IA), (IB) or (IC) - listed in Table 1 below.
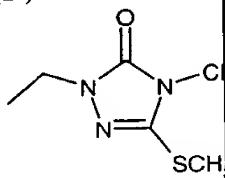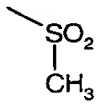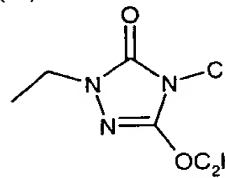
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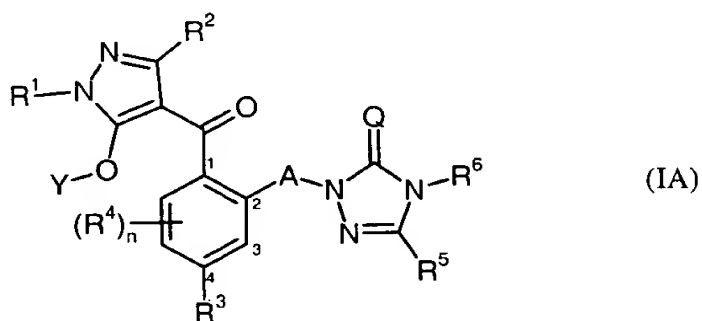
(IA)
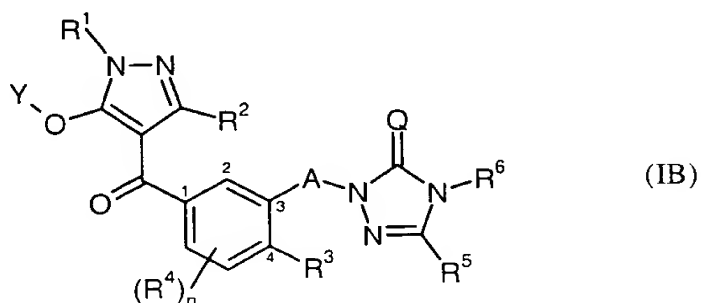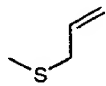
(IB)
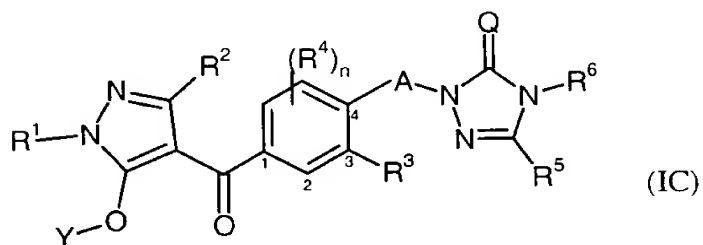
(IC)

Table 1: Examples of the compounds of the formula (I), (IA), (IB), (IC)

Here, Y in each case represents hydrogen

Ex. No.	A	Q	R^1	R^2	R^3	(Position) $(R^4)_n$	R^5	R^6	(Formula) Physical data
2	CH_2	O	C_2H_5	H	CF_3	-	OC_2H_5	CH_3	(IA) logP = 2.34[a]
3	CH_2	O	C_2H_5	H	CF_3	-	SCH_3	CH_3	(IA) logP = 2.22 [a]
4	CH_2	O	C_2H_5	H	SO_2CH_3	-	SCH_3	CH_3	(IA) logP = 1.24 [a]
5	CH_2	O	C_2H_5	H	CF_3	-	SC_2H_5	CH_3	(IA) logP = 2.58 [a]
6	CH_2	O	C_2H_5	H	CF_3	-	SC_3H_7-i	CH_3	(IA) logP = 2.90 [a]
7	CH_2	O	C_2H_5	H	CF_3	-	OCH_3	(cyclopropylmethyl)	(IA) logP = 2.28 [a]
8	CH_2	O	C_2H_5	H	F	-	$N(CH_3)_2$	CH_3	(IA) logP = 1.61 [a]
9	CH_2	O	CH_3	CH_3	F	-	$N(CH_3)_2$	CH_3	(IA) logP = 1.32 [a]
10	CH_2	O	CH_3	CH_3	F	-	OCH_3	(cyclopropylmethyl)	(IA) logP = 1.50 [a]
11	CH_2	O	CH_3	CH_3	F	-	OC_2H_5	(cyclopropylmethyl)	(IA) logP = 1.80 [a]
12	CH_2	O	C_2H_5	H	Br	-	(S-CH₂-CH=CH₂)	CH_3	(IA) logP = 2.69 [a]
13	-	O	C_2H_5	H	H	(6-) CF_3	CF_3	CH_3	(IB) logP = 2.83 [a]
14	-	O	C_2H_5	H	H	(2-) Cl	CH_3	CH_3	(IC) logP = 1.71 [a]
15	-	O	C_2H_5	H	H	-	CF_3	CH_3	(IA) logP = 1.95 [a]

Ex. No.	A	Q	R¹	R²	R³	(Position) (R⁴)ₙ	R⁵	R⁶	(Formula) Physical data
16	-	O	C_2H_5	H	Cl	-	CF_3	CH_3	(IA) logP = 2.47 [a]
17	CH_2	O	C_2H_5	H	Cl	(2-) Cl	CF_3	CH_3	(IB) logP = 2.30 [a]
18	CH_2	O	C_2H_5	H	Cl	(2-) Cl	SCH_3	CH_3	(IB) logP = 1.91 [a]
19	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OC_2H_5	CH_3	(IB) logP = 2.01 [a]
20	CH_2	O	C_2H_5	H	Cl	(2-) Cl	◁	◁	(IB) logP = 2.14 [a]
21	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OCH_3	CH_3	(IB) logP = 1.69 [a]
22	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OC_3H_7-i	CH_3	(IB) logP = 2.31 [a]
23	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OCH_2CF_3	CH_3	(IB) logP = 2.33 [a]
24	CH_2	O	C_2H_5	H	Cl	(2-) Cl	Br	CH_3	(IB) logP = 1.81 [a]
25	CH_2	O	C_2H_5	H	Cl	(2-) Cl	H	CH_3	(IB) logP = 1.28 [a]
26	CH_2	O	C_2H_5	H	Cl	(2-) Cl	◁	CH_3	(IB) logP = 1.82 [a]
27	-	O	C_2H_5	H	Br	-	CF_3	CH_3	(IA) logP = 2.55 [a]
28	CH_2	O	C_2H_5	H	Cl	(2-) Cl	$N(CH_3)_2$	CH_3	(IB) logP = 1.77 [a]
29	CH_2	O	C_2H_5	H	Cl	(2-) Cl	CH_3	CH_3	(IB) logP = 1.38 [a]
30	CH_2	O	C_2H_5	H	Cl	(2-) Cl	R⁵ + R⁶: $(CH_2)_4$	(cF. R⁵)	(IB) logP = 1.55 [a]
31	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OCH_3	◁	(IB) logP = 1.99 [a]

Ex. No.	A	Q	R^1	R^2	R^3	(Position) (R^4)$_n$	R^5	R^6	(Formula) Physical data
32	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OC_2H_5	△	(IB) logP = 2.31 [a]
33	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OC_3H_7-i	△	(IB) logP = 4.64 [a]
34	CH_2	O	C_2H_5	H	Cl	(2-) Cl	OCH_2CF_3	△	(IB) logP = 2.65 [a]
35	CH_2	O	C_2H_5	H	Cl	(2-) Cl	SCH_3	△	(IB) logP = 2.27 [a]
36	CH_2	O	C_2H_5	H	Cl	(2-) Cl	CH_3	△	(IB) logP = 1.64 [a]
37	CH_2	O	C_2H_5	H	Cl	(2-) Cl	$N(CH_3)_2$	△	(IB) logP = 2.04 [a]
38	CH_2	O	C_2H_5	H	Cl	(2-) Cl	C_2H_5	OC_2H_5	(IB) logP = 2.16 [a]
39	CH_2	O	CH_3	CH_3	Cl	(2-) Cl	Br	CH_3	(IB) logP = 1.52 [a]
40	CH_2	O	CH_3	H	Cl	(2-) Cl	Br	CH_3	(IB) logP = 1.53 [a]
41	CH_2	O	C_2H_5	CH_3	Cl	(2-) Cl	SCH_3	CH_3	(IB) logP = 1.91 [a]
42	CH_2	O	C_2H_5	CH_3	Cl	(2-) Cl	OC_2H_5	CH_3	(IB) logP = 2.02 [a]
43	CH_2	O	C_2H_5	CH_3	Cl	(2-) Cl	OCH_3	CH_3	(IB) logP = 1.71 [a]
44	CH_2	O	C_2H_5	CH_3	Cl	(2-) Cl	Br	CH_3	(IB) logP = 1.81 [a]
45	CH_2	O	C_2H_5	CH_3	Cl	(2-) Cl	CH_3	CH_3	(IB) logP = 1.40 [a]
46	CH_2	O	t-C_4H_9	CH_3	Cl	(2-) Cl	SCH_3	CH_3	(IB) logP = 3.30 [a]
47	CH_2	O	t-C_4H_9	CH_3	Cl	(2-) Cl	OC_2H_5	CH_3	(IB) logP = 3.44 [a]
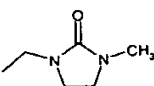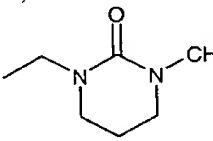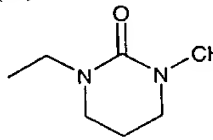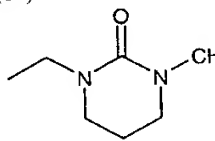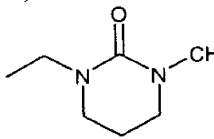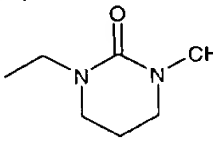
Ex. No.	A	Q	R^1	R^2	R^3	(Position) (R^4)$_n$	R^5	R^6	(Formula) Physical data
48	CH$_2$	O	t-C$_4$H$_9$	CH$_3$	Cl	(2-) Cl	OCH$_3$	CH$_3$	(IB) logP = 3.02 [a]
49	CH$_2$	O	t-C$_4$H$_9$	CH$_3$	Cl	(2-) Cl	Br	CH$_3$	(IB) logP = 3.19 [a]
50	CH$_2$	O	t-C$_4$H$_9$	CH$_3$	Cl	(2-) Cl	CH$_3$	CH$_3$	(IB) logP = 2.53 [a]
51	CH$_2$	O	CH$_3$	CH$_3$	Cl	(2-) Cl	SCH$_3$	CH$_3$	(IB) logP = 1.66 [a]
52	CH$_2$	O	CH$_3$	CH$_3$	Cl	(2-) Cl	OC$_2$H$_5$	CH$_3$	(IB) logP = 1.76 [a]
53	CH$_2$	O	CH$_3$	CH$_3$	Cl	(2-) Cl	OCH$_3$	CH$_3$	(IB) logP = 1.48 [a]
54	CH$_2$	O	CH$_3$	CH$_3$	Cl	(2-) Cl	CH$_3$	CH$_3$	(IB) logP = 1.20 [a]
55	CH$_2$	O	CH$_3$	H	Cl	(2-) Cl	SCH$_3$	CH$_3$	(IB) logP = 1.67 [a]
56	CH$_2$	O	CH$_3$	H	Cl	(2-) Cl	OC$_2$H$_5$	CH$_3$	(IB) logP = 1.77 [a]
57	CH$_2$	O	CH$_3$	H	Cl	(2-) Cl	OCH$_3$	CH$_3$	(IB) logP = 1.48 [a]
58	CH$_2$	O	CH$_3$	H	Cl	(2-) Cl	CH$_3$	CH$_3$	(IB) logP = 1.19 [a]
59	CH$_2$	O	C$_2$H$_5$	H	OCH$_3$	(2-) NO$_2$	OC$_2$H$_5$	CH$_3$	(IC) logP = 1.99 [a]
60	CH$_2$	O	C$_2$H$_5$	H	OCH$_3$	(2-) NO$_2$	SCH$_3$	CH$_3$	(IC) logP = 1.92 [a]
61	CH$_2$	O	C$_2$H$_5$	H	CF$_3$	-	SCH$_3$	CH$_3$	(IA-Na salt)
62	CH$_2$	O	C$_2$H$_5$	H	Cl	(2-) F	SCH$_3$	CH$_3$	(IB) logP = 1.99 [a]
63	CH$_2$	O	C$_2$H$_5$	H	CF$_3$	-	H	CH$_3$	(IA)
64	CH$_2$	O	C$_2$H$_5$	H	CF$_3$	-	CH$_3$	CH$_3$	(IA) logP = 1.80 [a]

Ex. No.	A	Q	R^1	R^2	R^3	(Position) $(R^4)_n$	R^5	R^6	(Formula) Physical data
65	CH_2	O	C_2H_5	H	CF_3	-	CH_2OCH_3	CH_3	(IA) logP = 1.98 [a)]
66	CH_2	O	C_2H_5	H	CF_3	-	OCH_3	CH_3	(IA) logP = 2.27 [a)]
67	CH_2	O	C_2H_5	H	SO_2CH_3	-	CF_3	CH_3	(IA) logP = 1.60 [a)]
68	CH_2	O	C_2H_5	H	SO_2CH_3	-	OCH_2CF_3	CH_3	(IA) logP = 1.73 [a)]
69	CH_2	O	C_2H_5	H	F	(2-) Cl	CH_3	CH_3	(IB) logP = 1.27 [a)]
70	CH_2	O	C_2H_5	H	F	(2-) Cl	SCH_3	CH_3	(IB) logP = 1.76 [a)]
71	CH_2	O	C_2H_5	H	F	(2-) Cl	OCH_3	CH_3	(IB) logP = 1.55 [a)]
72	CH_2	O	C_2H_5	H	F	(2-) Cl	$N(CH_3)_2$	CH_3	(IB) logP = 1.62 [a)]
73	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	SCH_3	CH_3	(IB) m.p.: 204°C
74	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	OCH_3	CH_3	(IB) m.p.: 183°C
75	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	OCH_2CF_3	CH_3	(IB) m.p.: 192°C
76	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	CH_3	CH_3	(IB) m.p.: 200°C
77	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	OCH_3	(cyclopropylmethyl)	(IB) m.p.: 205°C
78	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	SCH_3	(cyclopropylmethyl)	(IB) m.p.: 233°C
79	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	CH_3	(cyclopropylmethyl)	(IB) m.p.: 223°C
80	CH_2	O	C_2H_5	H	SO_2CH_3	(2-) Cl	C_2H_5	OC_2H_5	(IB) m.p.: 163°C
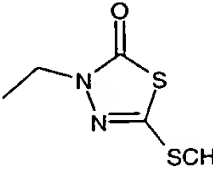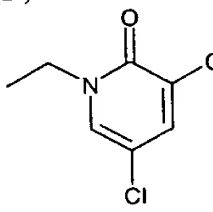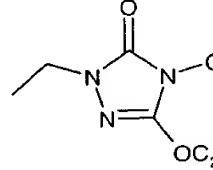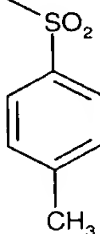
Similarly to Example 1, and in accordance with the general description of the preparation process according to the invention, it is also possible to prepare, for example, the compounds of the general formula (I) - or of the formula (ID) - listed in

5   Table 2 below.

(ID)

Table 2: Further examples of the compounds of the formula (I)
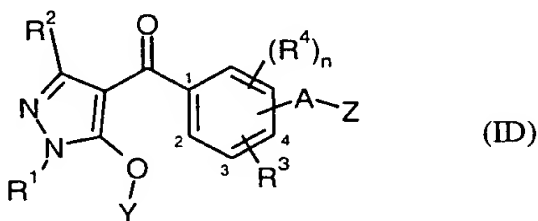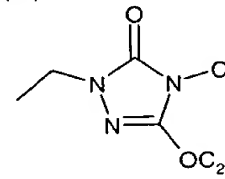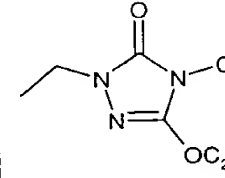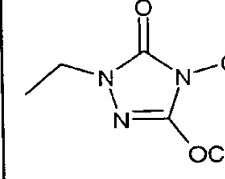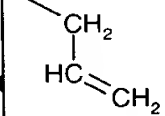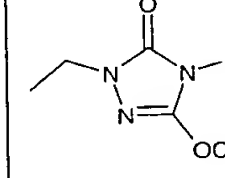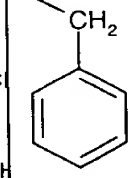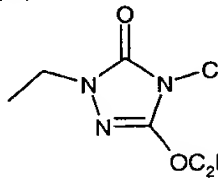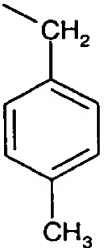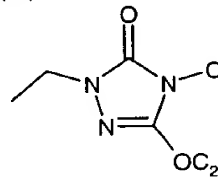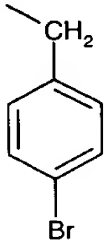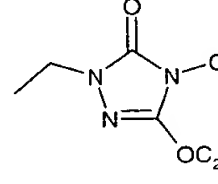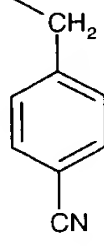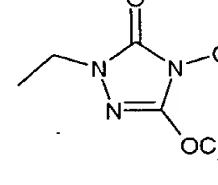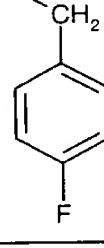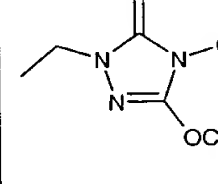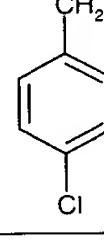
Ex. No.	R¹	R²	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Y	Physical data
ID-1	C_2H_5	H	(2-) Cl	(4-) Cl	(3-)	H	logP = 1.64 [a]
ID-2	C_2H_5	H	(2-) Cl	(4-) Cl	(3-)	H	logP = 1.77 [a]
ID-3	CH_3	H	(2-) Cl	(4-) Cl	(3-)	H	logP = 1.52 [a]
ID-4	CH_3	CH_3	(2-) Cl	(4-) Cl	(3-)	H	logP = 1.50 [a]
ID-5	C_2H_5	CH_3	(2-) Cl	(4-) Cl	(3-)	H	logP = 1.72 [a]
ID-6	t-C_4H_9	CH_3	(2-) Cl	(4-) Cl	(3-)	H	logP = 3.01 [a]
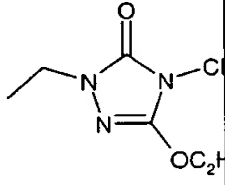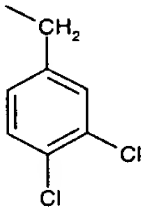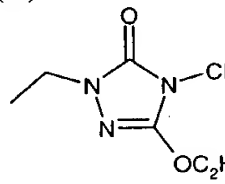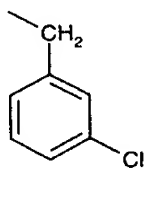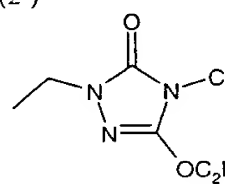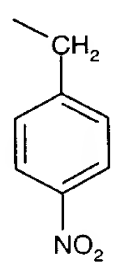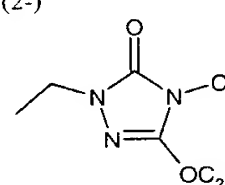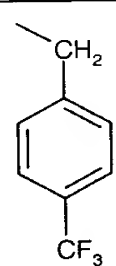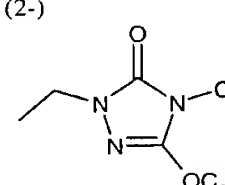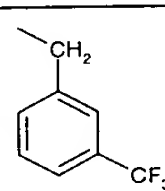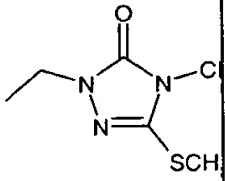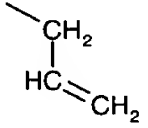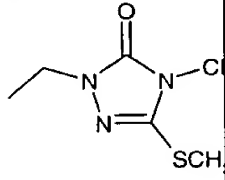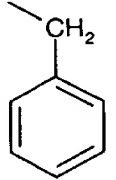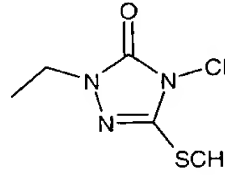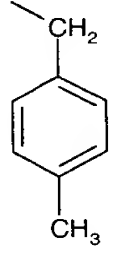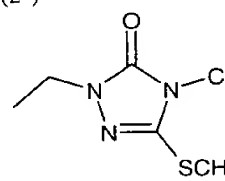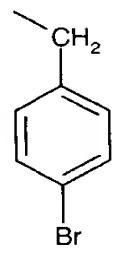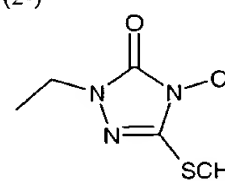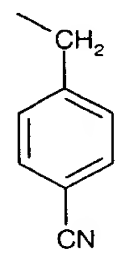
Ex. No.	R^1	R^2	(Position) R^3	(Position) $(R^4)_n$	(Position) -A-Z	Y	Physical data
ID-7	C_2H_5	H	(4-) CF_3	-	(2-) [1,3,4-thiadiazol-2(3H)-one ring, N-ethyl, S, —SCH]	H	logP = 2.98 [a]
ID-8	C_2H_5	H	(4-) CF_3	-	(2-) [N-ethyl pyridin-2(1H)-one ring, Cl, Cl]	H	logP = 2.75 [a]
ID-9	C_2H_5	H	(4-) CF_3	-	(2-) [1,2,4-triazol-3-one ring, N-ethyl, Cl, OC_2H]	[—SO_2—C$_6$H$_4$—CH_3]	logP = 3.82 [a]
ID-10	C_2H_5	H	(4-) CF_3	-	(2-) [1,2,4-triazol-3-one ring, N-ethyl, Cl, SCH]	[—SO_2—C$_6$H$_4$—CH_3]	logP = 3.73 [a]
ID-11	C_2H_5	H	(4-) CF_3	-	(2-) [1,2,4-triazol-3-one ring, N-ethyl, Cl, OC_2H]	[—SO_2—CH_3]	logP = 3.25 [a]
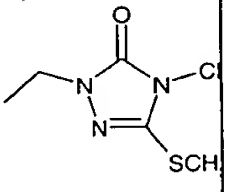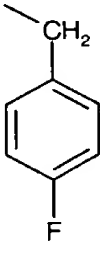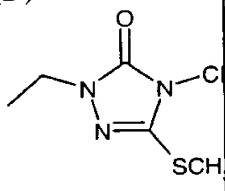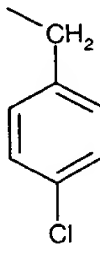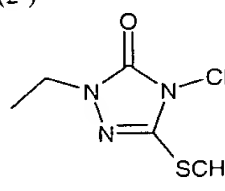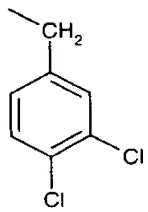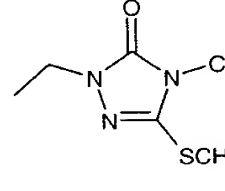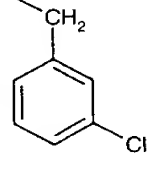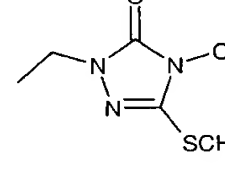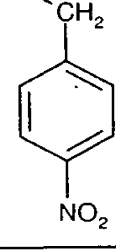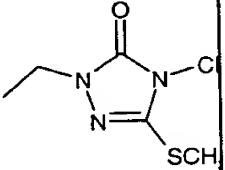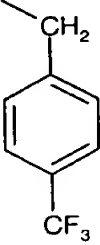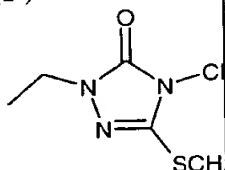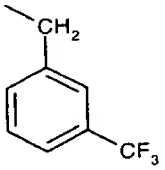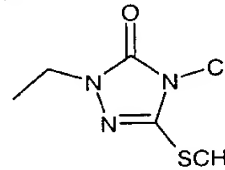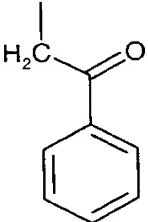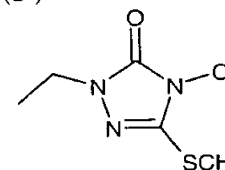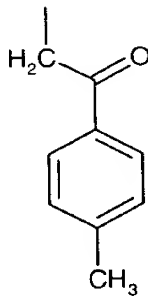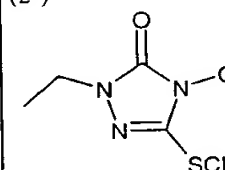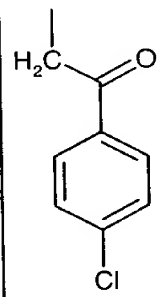
Ex. No.	R^1	R^2	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Y	Physical data
ID-12	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	SO$_2$–CH$_3$	logP = 2.82 [a]
ID-13	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	CH$_3$	logP = 2.74 [a]
ID-14	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	C$_2$H$_5$	logP = 2.82 [a]
ID-15	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	i-C$_3$H$_7$	logP = 3.11 [a]
ID-16	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	CH$_2$–HC=CH$_2$	logP = 2.99 [a]
ID-17	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	CH$_2$–C$_6$H$_5$	logP = 3.45 [a]
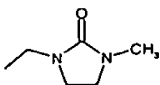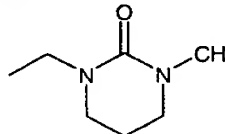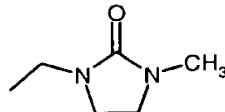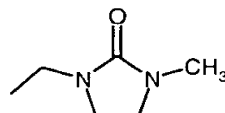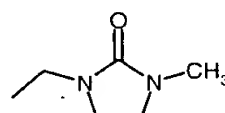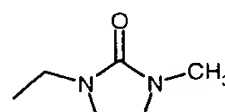
Ex. No.	R^1	R^2	(Position) R^3	(Position) $(R^4)_n$	(Position) -A-Z	Y	Physical data
ID-18	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone, N-ethyl, 4-Cl, OC_2H	benzyl, 4-CH_3	logP = 3.79 [a]
ID-19	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone, N-ethyl, 4-Cl, OC_2H	benzyl, 4-Br	logP = 3.97 [a]
ID-20	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone, N-ethyl, 4-Cl, OC_2H	benzyl, 4-CN	logP = 3.12 [a]
ID-21	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone, N-ethyl, 4-Cl, OC_2H	benzyl, 4-F	logP = 3.49 [a]
ID-22	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone, N-ethyl, 4-Cl, OC_2H	benzyl, 4-Cl	logP = 3.85 [a]
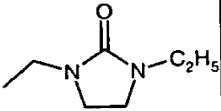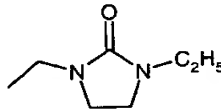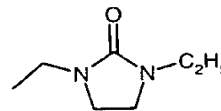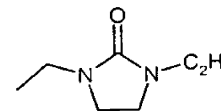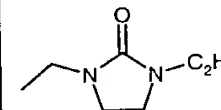
Ex. No.	R¹	R²	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Y	Physical data
ID-23	C_2H_5	H	(4-) CF_3	-	(2-)		$logP = 4.26$ [a]
ID-24	C_2H_5	H	(4-) CF_3	-	(2-)		$logP = 3.84$ [a]
ID-25	C_2H_5	H	(4-) CF_3	-	(2-)		$logP = 3.33$ [a]
ID-26	C_2H_5	H	(4-) CF_3	-	(2-)		$logP = 3.98$ [a]
ID-27	C_2H_5	H	(4-) CF_3	-	(2-)		$logP = 3.94$ [a]

Ex. No.	R^1	R^2	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Y	Physical data
ID-28	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.57 [a)]
ID-29	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.75 [a)]
ID-30	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	CH$_3$	logP = 2.65 [a)]
ID-31	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	C$_2$H$_5$	logP = 2.71 [a)]
ID-32	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)	i-C$_3$H$_7$	logP = 3.00 [a)]

Ex. No.	R^1	R^2	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Y	Physical data
ID-33	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 2.89 [a]
ID-34	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.37 [a]
ID-35	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.71 [a]
ID-36	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.89 [a]
ID-37	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.06 [a]

Ex. No.	R^1	R^2	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Y	Physical data
ID-38	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.41 [a)]
ID-39	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.78 [a)]
ID-40	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 4.17 [a)]
ID-41	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.76 [a)]
ID-42	C$_2$H$_5$	H	(4-) CF$_3$	-	(2-)		logP = 3.26 [a)]

Ex. No.	R¹	R²	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Y	Physical data
ID-43	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone (ethyl, =O, Cl, SCH_2)	CH_2-phenyl-CF_3 (para)	logP = 3.89 [a]
ID-44	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone (ethyl, =O, Cl, SCH_2)	CH_2-phenyl-CF_3 (meta)	logP = 3.85 [a]
ID-45	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone (ethyl, =O, Cl, SCH_2)	H_2C-C(=O)-phenyl	logP = 3.19 [a]
ID-46	C_2H_5	H	(4-) CF_3	-	(2-) triazolinone (ethyl, =O, Cl, SCH_2)	H_2C-C(=O)-phenyl-CH_3 (para)	logP = 3.47 [a]
ID-47	C_2H_5	H	(4-) CF	-	(2-) triazolinone (ethyl, =O, Cl, SCH_2)	H_2C-C(=O)-phenyl-Cl (para)	logP = 3.64 [a]

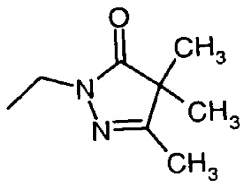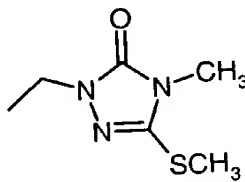| Ex. No.	R¹	R²	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Y	Physical data
ID-48	C_2H_5	H	(2-) OCH_3	(4-) Cl	(3-) imidazolidinone (1-ethyl, 3-methyl)	H	
ID-49	C_2H_5	H	(2-) OCH_3	(4-) Cl	(3-) tetrahydropyrimidinone (1-ethyl, 3-CH)	H	
ID-50	CH_3	H	(2-) Cl	(4-) Cl	(3-) imidazolidinone (1-ethyl, 3-CH_3)	H	logP = 1.41 [a)]
ID-51	CH_3	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone (1-ethyl, 3-CH_3)	H	logP = 1.38 [a)]
ID-52	C_2H_5	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone (1-ethyl, 3-CH_3)	H	logP = 1.56 [a)]
ID-53	t-C_4H_9	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone (1-ethyl, 3-CH_3)	H	logP = 2.79 [a)]

Ex. No.	R^1	R^2	(Position) R^3	(Position) $(R^4)_n$	(Position) -A-Z	Y	Physical data
ID-54	CH_3	H	(2-) Cl	(4-) Cl	(3-) imidazolidinone, N-ethyl, N'-C_2H_5	H	logP = 1.62 [a]
ID-55	CH_3	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone, N-ethyl, N'-C_2H_5	H	logP = 1.57 [a]
ID-56	C_2H_5	H	(2-) Cl	(4-) Cl	(3-) imidazolidinone, N-ethyl, N'-C_2H_5	H	logP = 1.88 [a]
ID-57	C_2H_5	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone, N-ethyl, N'-C_2H_5	H	logP = 1.79 [a]
ID-58	t-C_4H_9	CH_3	(2-) Cl	(4-) Cl	(3-) imidazolidinone, N-ethyl, N'-C_2H_5	H	logP = 3.15 [a]

The logP values given in Tables 1 and 2 were determined in accordance with EEC Directive 79/831 Annex V.A8 by HPLC (High Performance Liquid Chromatography) on a reversed-phase column (C 18). Temperature: 43°C.

5

(a) Mobile phases for the determination in the acidic range: 0.1% aqueous phosphoric acid, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked a).

5   (b) Mobile phases for the determination in the neutral range: 0.01 molar aqueous phosphate buffer solution, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked b).

Calibration was carried out using unbranched alkan-2-ones (having 3 to 16 carbon

10   atoms) with known logP values (determination of the logP values by the retention times using linear interpolation between two successive alkanones).

The lambda max values were determined in the maxima of the chromatographic signals using the UV spectra from 200 nm to 400 nm.

15

- 85 -

Starting materials of formula (III):

Example (III-1)
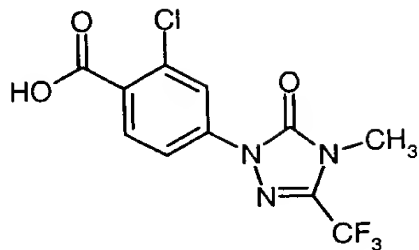
5

4.5 g (15 mmol) of 2-(3-chloro-4-cyano-phenyl)-4-methyl-5-trifluoromethyl-2,4-dihydro-3H-1,2,4-triazol-3-one are taken up in 80 ml of 60% strength sulfuric acid, and the mixture is heated at reflux for 6 hours. After cooling to room temperature, the resulting crystalline product is isolated by filtration with suction.

10

This gives 4.5 g (91% of theory) of 2-(3-carboxy-4-chloro-phenyl)-4-methyl-5-trifluoromethyl-2,4-dihydro-3H-1,2,4-triazol-3-one of melting point 223°C.

Example (III-2)

15
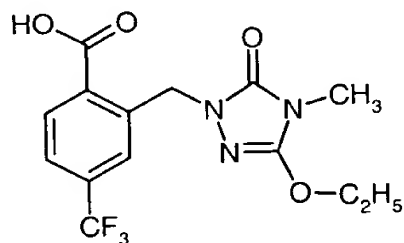
2 g (4.9 mmol) of 5-bromo-4-methyl-2-(2-ethoxycarbonyl-5-trifluoromethyl-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one (cf. Example IV-1) are dissolved in 30 ml of 10% strength ethanolic potassium hydroxide solution and heated at reflux for 2 hours. The reaction mixture is concentrated under waterpump vacuum and the residue is taken up in 20 ml of water and acidified with dilute hydrochloric acid. The precipitated solid is filtered and dried.

20

- 86 -

This gives 1.2 g (71% of theory) of 5-ethoxy-4-methyl-2-(2-carboxy-5-trifluoro-methyl-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one as a solid product.

5   logP: 2.18[a)]

Example (III-3)
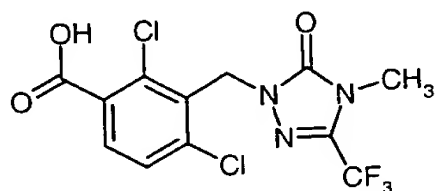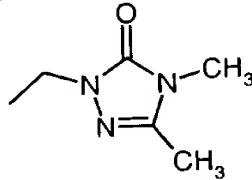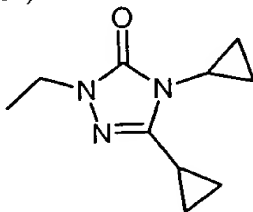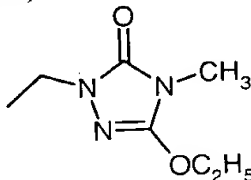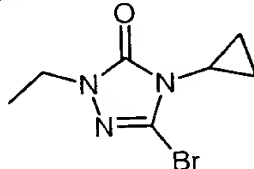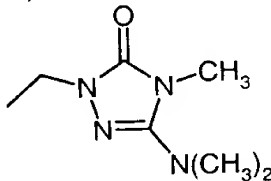
10   13.4 g (35 mmol) of 4-methyl-5-trifluoromethyl-2-(2,6-dichloro-3-methoxycarbonyl-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one are initially charged in 60 ml of 1,4-dioxane, and a solution of 1.54 g (38.5 mmol) of sodium hydroxide in 20 ml of water is slowly metered in at room temperature. The reaction mixture is stirred at 60°C for 150 minutes and then concentrated under waterpump vacuum. The residue
15   is dissolved in 100 ml of water, and the pH of the solution is adjusted to 1 by addition of conc. hydrochloric acid. The resulting crystalline product is isolated by filtration with suction.

This gives 11.7 g (90% of theory) of 4-methyl-5-trifluoromethyl-2-(2,6-dichloro-3-
20   carboxy-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one of melting point 207°C.
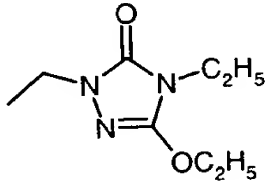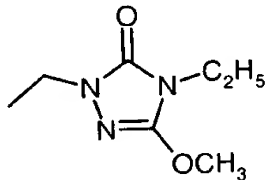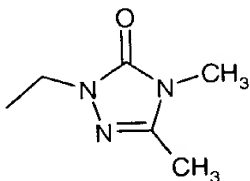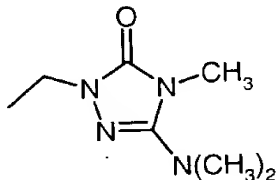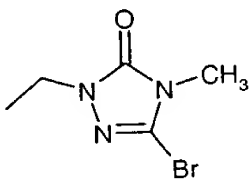
Similarly to Examples (III-1) to (III-3), it is also possible to prepare, for example, the compounds of the general formula (III) listed in Table 3 below.

(III)
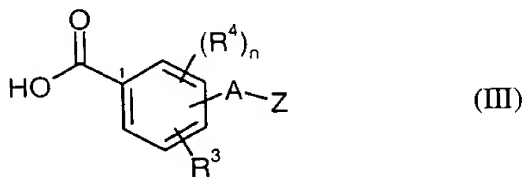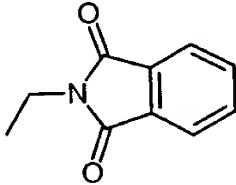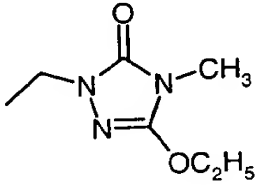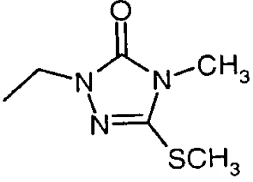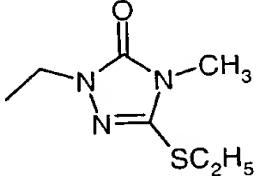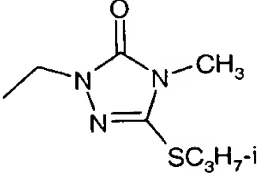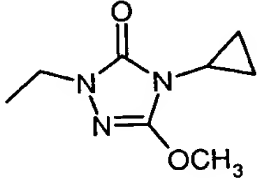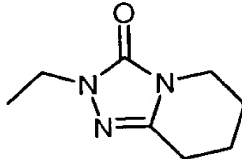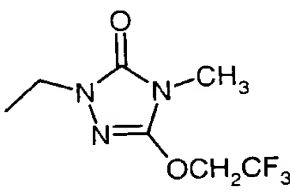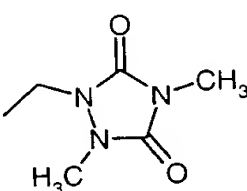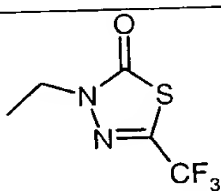
25   **Table 32:** Examples of the compounds of the formula (III)
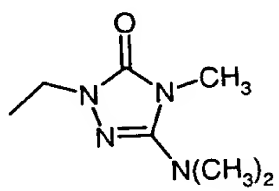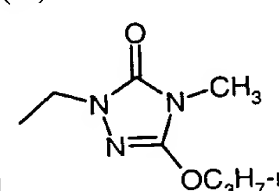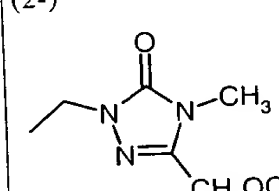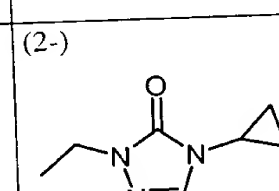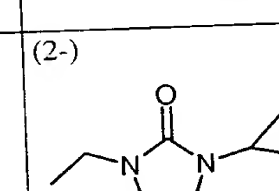
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-4	(4-) Cl	-	(2-)	logP = 1.39 [a)]
III-5	(4-) SO₂CH₃	-	(3-)	logP = 1.47 [a)]
III-6	(4-) F	-	(2-)	logP = 1.73 [a)]
III-7	(4-) CF₃	-	(2-)	logP = 1.65 [a)]
III-8	(4-) Br	-	(2-)	logP = 1.74 [a)]

Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
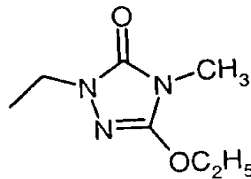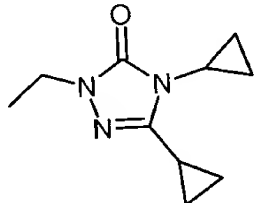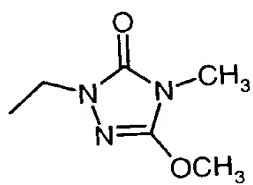
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-9	(4-) CF$_3$	-	(2-)	logP = 2.43 [a)]
III-10	(4-) CF$_3$	-	(2-)	logP = 2.12 [a)]
III-11	(4-) CF$_3$	-	(2-)	logP = 1.61 [a)]
III-12	(4-) CF$_3$	-	(2-)	logP = 1.93 [a)]
III-13	(4-) CF$_3$	-	(2-)	logP = 2.01 [a)]
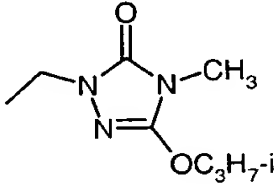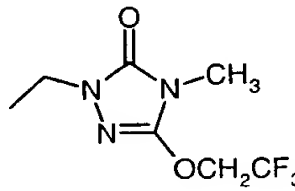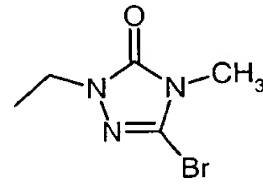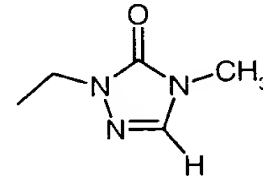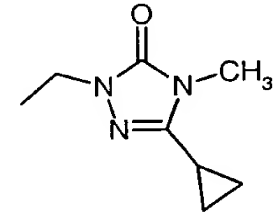
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-14	(4-) CF$_3$	-	(2-)	logP = 1.77 [a]
III-15	(3-) CH$_3$	-	(2-) OC$_2$H$_5$	logP = 1.70 [a]
III-16	(4-) SO$_2$CH$_3$	-	(2-) SCH$_3$	logP = 1.07 [a]
III-17	(4-) CF$_3$	-	(2-) SC$_2$H$_5$	logP = 2.35 [a]
III-18	(4-) CF$_3$	-	(2-) SC$_3$H$_7$-i	logP = 2.63 [a]

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-19	(4-) CF$_3$	-	(2-) [triazolone ring: N-ethyl, N-cyclopropyl, =O, OCH$_3$]	logP = 2.13 [a]
III-20	(4-) CF$_3$	-	(2-) [bicyclic triazolone ring: N-ethyl, =O, fused cyclohexane]	logP = 1.82 [a]
III-21	(4-) CF$_3$	-	(2-) [triazolone ring: N-ethyl, N—CH$_3$, =O, OCH$_2$CF$_3$]	logP = 2.48 [a]
III-22	(4-) CF$_3$	-	(2-) [triazolidinedione ring: N-ethyl, N—CH$_3$, two =O, H$_3$C—N]	logP = 1.73 [a]
III-23	(4-) CF$_3$	-	(2-) [thiadiazolone ring: N-ethyl, =O, S, CF$_3$]	logP = 3.11 [a]
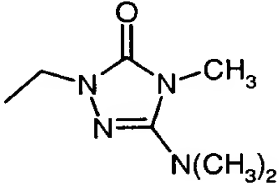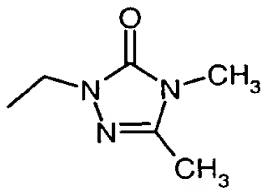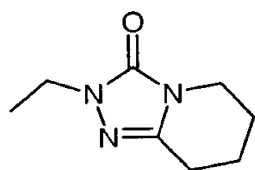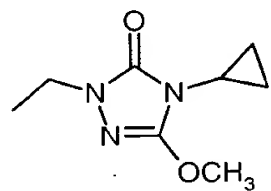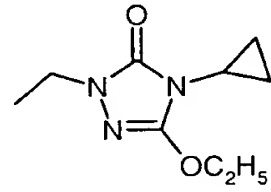
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-24	(4-) F	-	(2-)	logP = 1.43 [a]
III-25	(4-) F	-	(2-)	logP = 1.97 [a]
III-26	(4-) F	-	(2-)	logP = 1.30 [a]
III-27	(4-) F	-	(2-)	logP = 1.63 [a]
III-28	(4-) F	-	(2-)	logP = 1.93 [a]
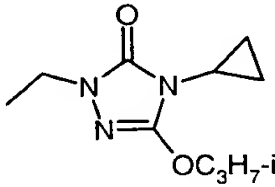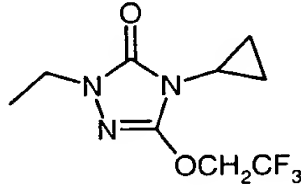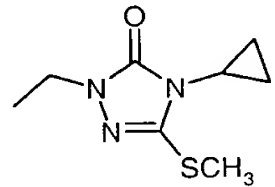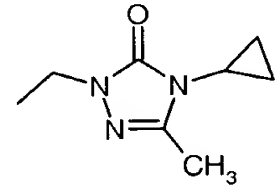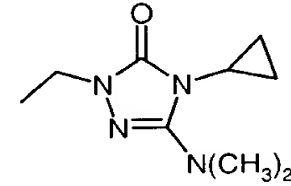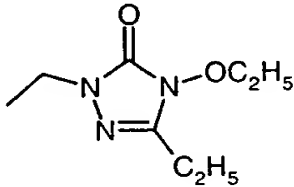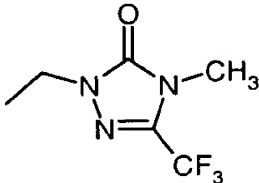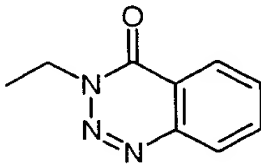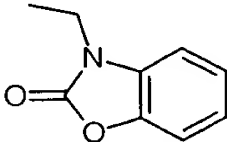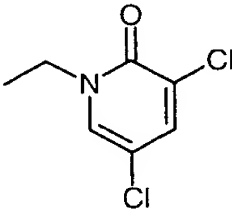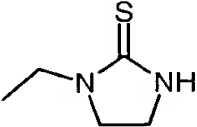
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-29	(4-) CF$_3$	-	(2-)	logP = 1.78 [a]
III-30	(2-) Cl	(4-) Cl	(3-)	m.p.: 230°C logP = 1.63 [a]
III-31	(2-) Cl	(4-) Cl	(3-)	m.p.: 190°C logP = 1.73 [a]
III-32	(2-) Cl	(4-) Cl	(3-)	m.p.: 210°C logP = 1.87 [a]
III-33	(2-) Cl	(4-) Cl	(3-)	m.p.: 210°C logP = 1.43 [a]
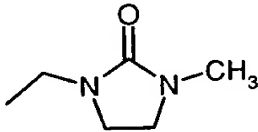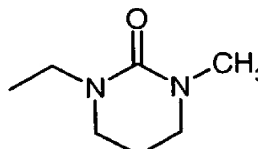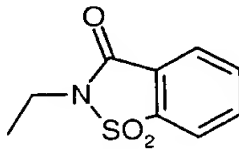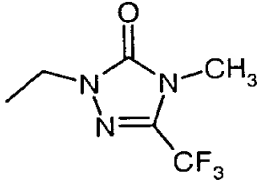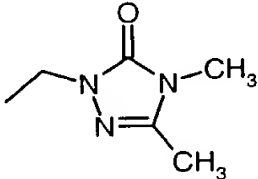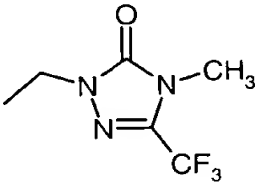
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-34	(2-) Cl	(4-) Cl	(3-)	m.p.: 164°C logP = 2.01 [a]
III-35	(2-) Cl	(4-) Cl	(3-)	m.p.: 168°C logP = 2.04 [a]
III-36	(2-) Cl	(4-) Cl	(3-)	m.p.: 218°C logP = 1.53 [a]
III-37	(2-) Cl	(4-) Cl	(3-)	m.p.: 259°C logP = 0.98 [a]
III-38	(2-) Cl	(4-) Cl	(3-)	m.p.: 210°C logP = 1.56 [a]
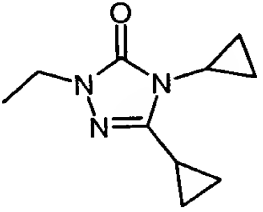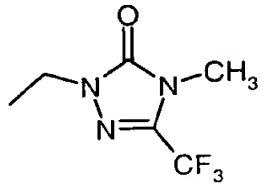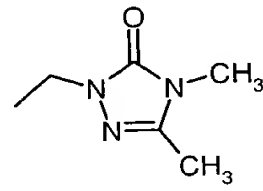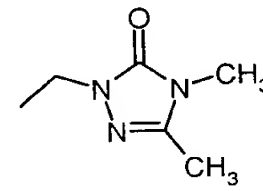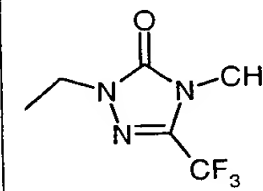
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-39	(2-) Cl	(4-) Cl	(3-)	m.p.: 197°C logP = 1.51 [a]
III-40	(2-) Cl	(4-) Cl	(3-)	m.p.: 262°C logP = 1.11 [a]
III-41	(2-) Cl	(4-) Cl	(3-)	m.p.: 249°C logP = 1.30 [a]
III-42	(2-) Cl	(4-) Cl	(3-)	m.p.: 200°C logP = 1.71 [a]
III-43	(2-) Cl	(4-) Cl	(3-)	m.p.: 189°C logP = 2.01 [a]
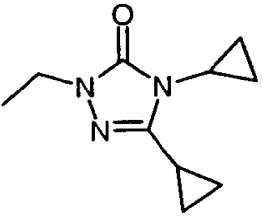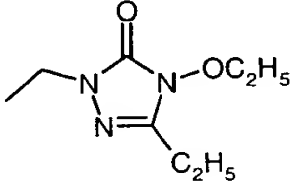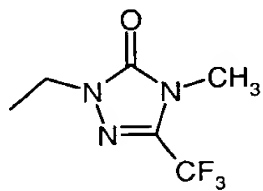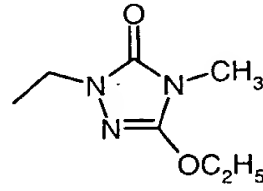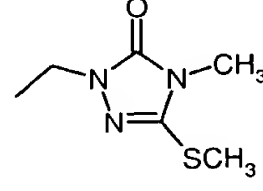
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-44	(2-) Cl	(4-) Cl	(3-) triazolone, N-ethyl, N-cyclopropyl, OC_3H_7-i	m.p.: 178°C $\log P = 2.28$ [a]
III-45	(2-) Cl	(4-) Cl	(3-) triazolone, N-ethyl, N-cyclopropyl, OCH_2CF_3	m.p.: 161°C $\log P = 2.31$ [a]
III-46	(2-) Cl	(4-) Cl	(3-) triazolone, N-ethyl, N-cyclopropyl, SCH_3	m.p.: 200°C $\log P = 1.98$ [a]
III-47	(2-) Cl	(4-) Cl	(3-) triazolone, N-ethyl, N-cyclopropyl, CH_3	m.p.: 201°C $\log P = 1.39$ [a]
III-48	(2-) Cl	(4-) Cl	(3-) triazolone, N-ethyl, N-cyclopropyl, $N(CH_3)_2$	m.p.: 207°C $\log P = 1.77$ [a]
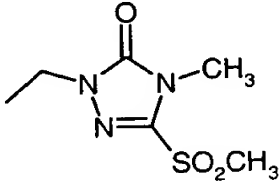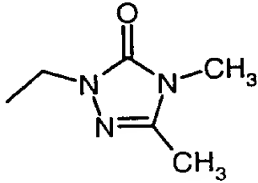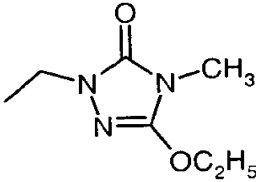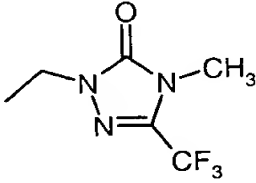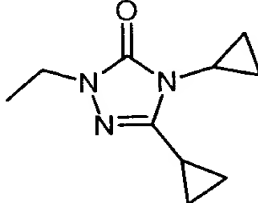
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-49	(2-) Cl	(4-) Cl	(3-)	m.p.: 140°C logP = 1.88 [a]
III-50	(4-) OCH$_2$CHF$_2$	-	(2-)	m.p.: 154°C logP = 2.14 [a]
III-51	-	-	(2-)	m.p.: 214°C logP = 1.87 [a]
III-52	-	-	(2-)	m.p.: 194°C logP = 2.07 [a]
III-53	-	-	(2-)	m.p.: 181°C logP = 1.97 [a]
III-54	-	-	(2-)	m.p.: 251°C logP = 1.14 [a]
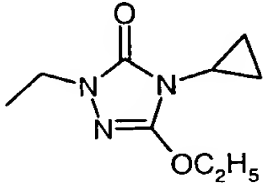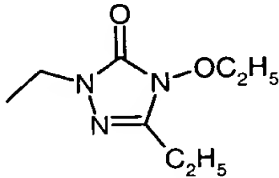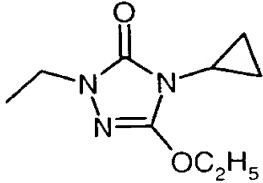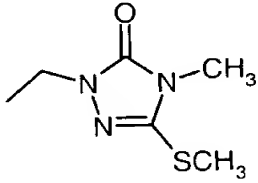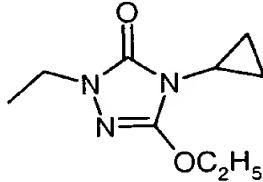
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-55	(2-) Cl	(4-) Cl	(3-)	logP = 1,38 [a]
III-56	(2-) Cl	(4-) Cl	(3-)	logP = 1.48 [a]
III-57	(2-) Cl	(4-) Cl	(3-)	
III-58	(4-) Cl	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.42 ppm.
III-59	(4-) CF₃	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.48 ppm.
III-60	(4-) CF₃	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.60 ppm. logP = 2.47 [a]
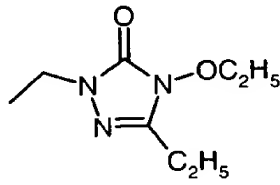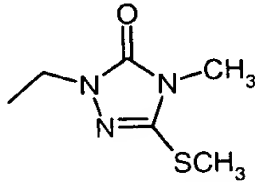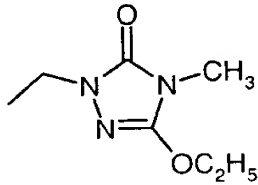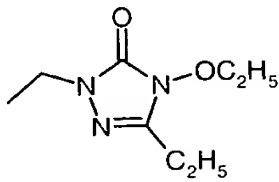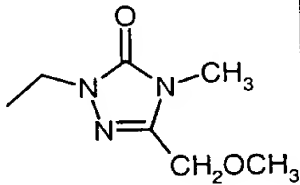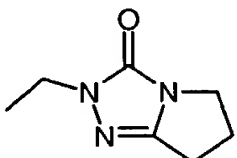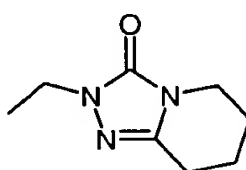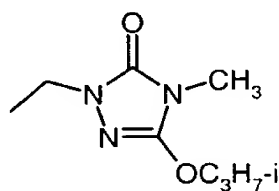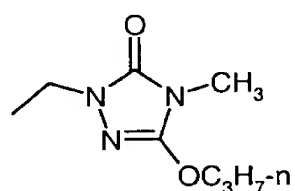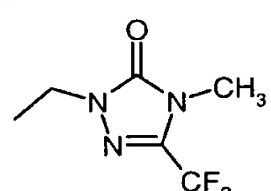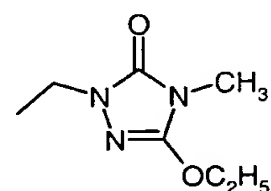
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-61	(4-) CF$_3$	-	(2-)	logP = 2.33 [a]
III-62	(4-) SO$_2$CH$_3$	-	(3-)	^{1}H-NMR (DMSO-D6, δ): 5.14 ppm.
III-63	(4-) SO$_2$CH$_3$	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.27 ppm.
III-64	(4-) Cl	-	(3-)	^{1}H-NMR (CDCl$_3$, δ): 5.12 ppm.
III-65	(4-) Cl	-	(3-)	^{1}H-NMR (DMSO-D6, δ): 5.20 ppm.

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-66	(4-) Cl	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.03 ppm.
III-67	(4-) Br	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.24 ppm.
III-68	(4-) Br	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.39 ppm.
III-69	(4-) F	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.19 ppm.
III-70	(4-) F	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.30 ppm.
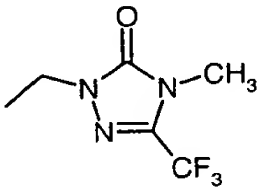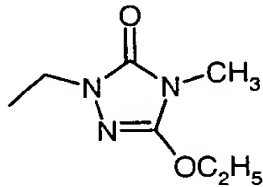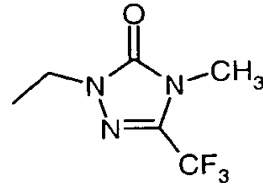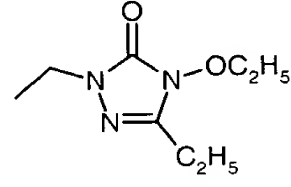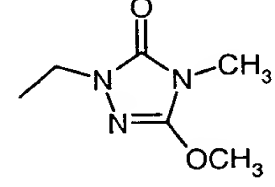
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-71	(4-) F	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.43 ppm.
III-72	(4-) Br	-	(3-)	¹H-NMR, (CDCl₃ δ): 5.10 ppm.
III-73	(4-) Br	-	(3-)	¹H-NMR (DMSO-D6, δ): 5.03 ppm.
III-74	(4-) Br	-	(3-)	¹H-NMR (DMSO-D6, δ): 5.19 ppm.
III-75	(4-) Br	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.01 ppm.
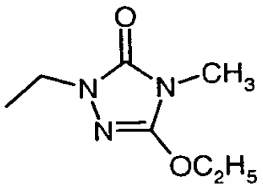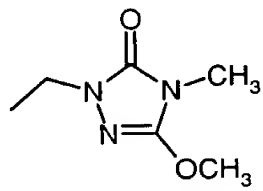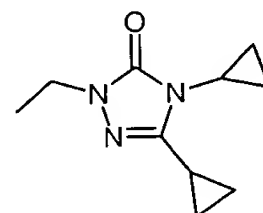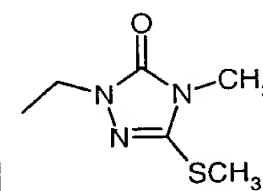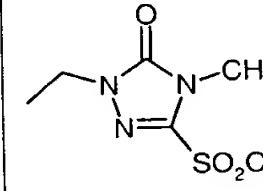
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-76	(4-) Cl	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.14 ppm.
III-77	(4-) Cl	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.25 ppm.
III-78	(4-) NO$_2$	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.23 ppm.
III-79	(4-) NO$_2$	-	(2-)	^{1}H-NMR (DMSO-D6, δ): 5.37 ppm.
III-80	(4-) CF$_3$	-	(2-)	logP = 2.46 $^{a)}$
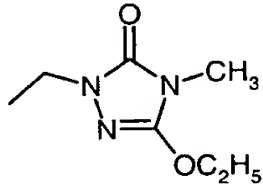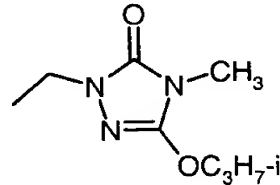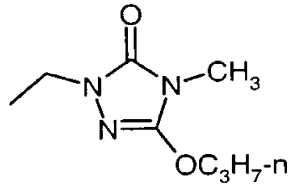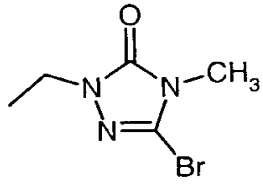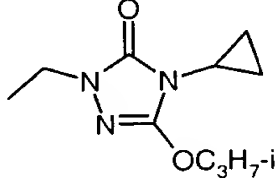
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-81	(4-) CF₃	-	(2-)	¹H-NMR (DMSO-D6, δ): 5.31 ppm.
III-82	(4-) CF₃	-	(2-)	logP = 2.08 [a]
III-83	(4-) OCH₃	-	(2-)	¹H-NMR (CDCl₃, δ): 5.38 ppm.
III-84	(4-) OCH₃	-	(2-)	¹H-NMR (CDCl₃, δ): 5.43 ppm.
III-85	(4-) CF₃	-	(2-)	¹H-NMR (CDCl₃, δ): 5.47 ppm.
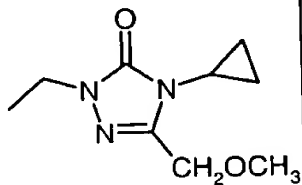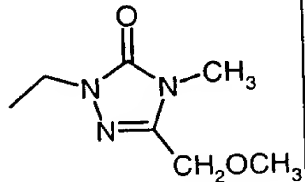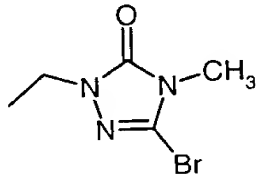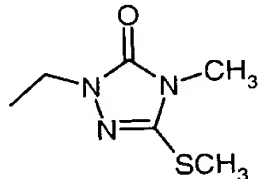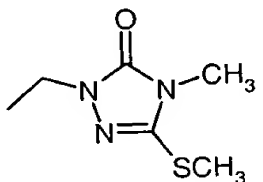
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-86	(4-) Br	-	(2-)	logP = 1.44 [a]
III-87	(4-) Br	-	(2-)	logP = 1.63 [a]
III-88	(4-) Br	-	(2-) OC$_3$H$_7$-i	logP = 2.27 [a]
III-89	(4-) Br	-	(2-) OC$_3$H$_7$-n	logP = 2.31 [a]
III-90	-	-	(2-) CF$_3$	logP = 1.82 [a]
III-91	(4-) Br	-	(2-) OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.32 ppm.
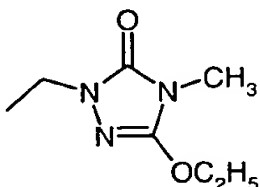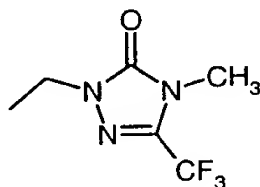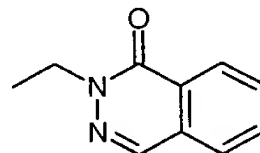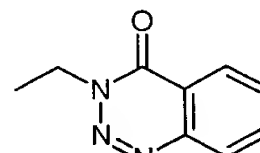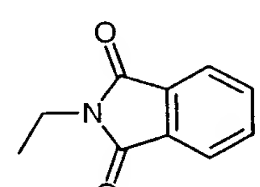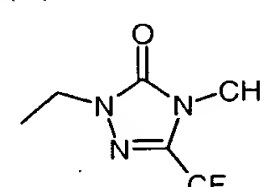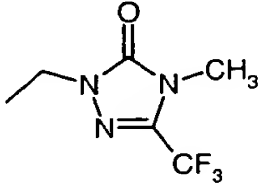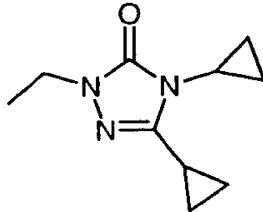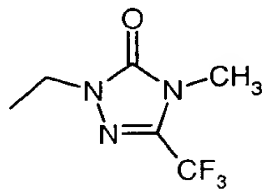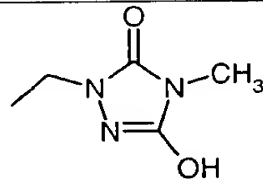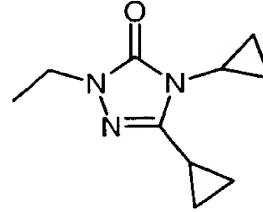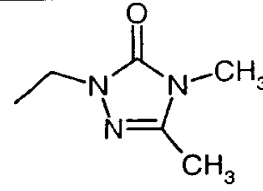
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-92	(4-) Br	-	(2-) ethyl-N-N-CH₃ triazolone with CF₃	¹H-NMR (CDCl₃, δ): 5.53 ppm.
III-93	(4-) F	-	(2-) ethyl-N-N-CH₃ triazolone with OC₂H₅	¹H-NMR (CDCl₃, δ): 5.39 ppm.
III-94	(4-) F	-	(2-) ethyl-N-N-CH₃ triazolone with CF₃	¹H-NMR (CDCl₃, δ): 5.57 ppm.
III-95	(4-) F	-	(2-) ethyl-N-N-OC₂H₅ triazolone with C₂H₅	¹H-NMR (CDCl₃, δ): 5.44 ppm.
III-96	(4-) F	-	(2-) ethyl-N-N-CH₃ triazolone with OCH₃	¹H-NMR (CDCl₃, δ): 5.41 ppm.
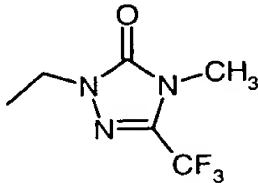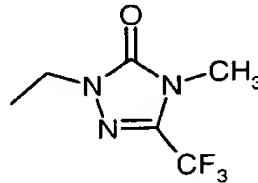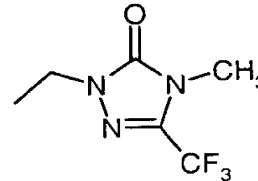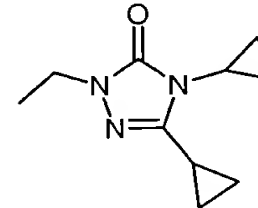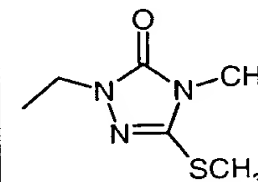
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-97	-	-	(2-) 	¹H-NMR (CDCl₃, δ): 5.34 ppm.
III-98	-	-	(2-) 	¹H-NMR (CDCl₃, δ): 5.38 ppm.
III-99	-	-	(2-) 	¹H-NMR (CDCl₃, δ): 5.26 ppm.
III-100	-	-	(2-) 	¹H-NMR (CDCl₃, δ): 5.43 ppm.
III-101	-	-	(2-) 	logP = 1.23 [a]
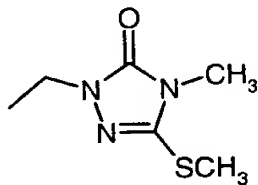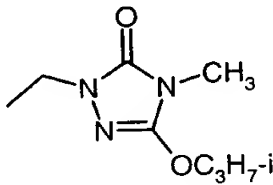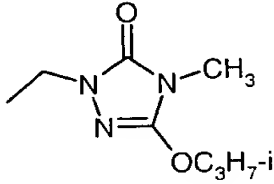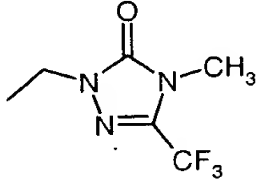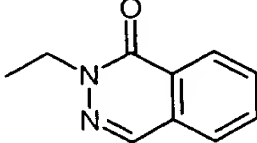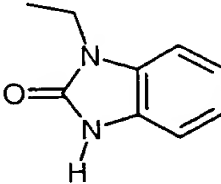
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-102	(4-) SO$_2$CH$_3$	-	(2-) structure with OC$_2$H$_5$	logP = 1.14 [a]
III-103	(4-) CF$_3$	-	(2-) structure with OC$_3$H$_7$-i	logP = 2.45 [a]
III-104	(4-) CF$_3$	-	(2-) structure with OC$_3$H$_7$-n	logP = 2.48 [a]
III-105	(4-) Br	-	(2-) structure with Br	logP = 1.85 [a]
III-106	(4-) CF$_3$	-	(3-) structure with cyclopropyl and OC$_3$H$_7$-i	logP = 2.74 [a]

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-107	(4-) CF$_3$	-	(2-)	logP = 2.01 [a]
III-108	(4-) CF$_3$	-	(2-)	logP = 1.79 [a]
III-109	(4-) CF$_3$	-	(2-)	logP = 1.65 [a]
III-110	(4-) Br	-	(2-)	logP = 1.90 [a]
III-111	(4-) Cl	-	(2-)	logP = 1.83 [a]

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-112	(4-) I	-	(2-)	logP = 2.06 [a]
III-113	(4-) I	-	(2-)	m.p.: 104°C logP = 2.39 [a]
III-114	(4-) Br	-	(2-)	m.p.: 191°C
III-115	(4-) Br	-	(2-)	m.p.: 213°C
III-116	-	-	(2-)	
III-117	-	-	(2-)	m.p.: 112°C

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-118	(4-) CF$_3$	-	(2-) triazolinone: N-ethyl, N-CH$_3$, CF$_3$	m.p.: 158°C
III-119	(4-) CF$_3$	-	(2-) triazolinone: N-ethyl, N-cyclopropyl, cyclopropyl	m.p.: 162°C
III-120	(4-) Cl	(5-) Cl	(2-) triazolinone: N-ethyl, N-CH$_3$, CF$_3$	m.p.: 167°C
III-121	-	-	(2-) triazolinone: N-ethyl, N-CH$_3$, OH	m.p.: 188°C
III-122	-	-	(2-) triazolinone: N-ethyl, N-cyclopropyl, cyclopropyl	—
III-123	-	-	triazolinone: N-ethyl, N-CH$_3$, CH$_3$	m.p.: 131°C

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	Physical data
III-124	(4-) Cl	-	(2-) 2-ethyl-4-methyl-5-(trifluoromethyl)-2,4-dihydro-3H-1,2,4-triazol-3-one	m.p.: 109°C
III-125	(4-) I	-	(2-) 2-ethyl-4-methyl-5-(trifluoromethyl)-2,4-dihydro-3H-1,2,4-triazol-3-one	m.p.: 104°C
III-126	(4-) Br	-	(2-) 2-ethyl-4-methyl-5-(trifluoromethyl)-2,4-dihydro-3H-1,2,4-triazol-3-one	m.p.: 99°C
III-127	(4-) Br	-	(2-) 4,5-dicyclopropyl-2-ethyl-2,4-dihydro-3H-1,2,4-triazol-3-one	m.p.: 174°C
III-128	-	-	(2-) 2-ethyl-4-methyl-5-(methylthio)-2,4-dihydro-3H-1,2,4-triazol-3-one	m.p.: 122°C

Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	Physical data
III-129	(4-) Br	-	(2-)	m.p.: 164°C
III-130	-	-	(2-)	m.p.: 154°C
III-131	(4-) Br	-	(2-)	m.p.: 161°C
III-132	(4-) CN	-	(2-)	m.p.: 196°C
III-133	-	-	(2-)	m.p.: 192°C
III-134	-	-		

The logP values given in Table 3 were determined in accordance with EEC Directive 79/831 Annex V.A8 by HPLC (High Performance Liquid Chromatography) on a reversed-phase column (C 18). Temperature: 43°C.

5

(a) Mobile phases for the determination in the acidic range: 0.1% aqueous phosphoric acid, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked [a].

10

(b) Mobile phases for the determination in the neutral range: 0.01 molar aqueous phosphate buffer solution, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked [b].

Calibration was carried out using unbranched alkan-2-ones (having 3 to 16 carbon atoms) with known logP values (determination of the logP values by the retention

15

times using linear interpolation between two successive alkanones).

The lambda max values were determined in the maxima of the chromatographic signals using the UV spectra from 200 nm to 400 nm.

Starting materials of the formula (IV):

Example (IV-1)
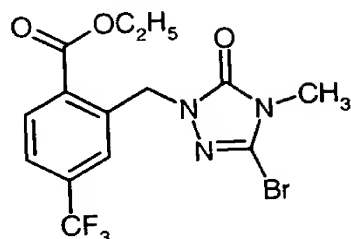
5

Step 1
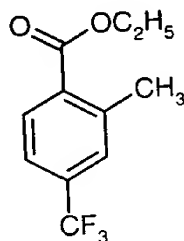
10   10 g (49 mmol) of 2-methyl-4-trifluoromethyl-benzoic acid are dissolved in 150 ml of ethanol and admixed with 1 ml of conc. sulfuric acid. The solution is heated at reflux for 24 hours and then concentrated, the residue is taken up in methylene chloride and the mixture is extracted with aqueous sodium bicarbonate solution. The methylene chloride phase is dried over sodium sulfate and concentrated under
15   waterpump vacuum.

This gives 9 g (80% of theory) of ethyl 2-methyl-4-trifluoromethyl-benzoate as an amorphous residue.

Step 2
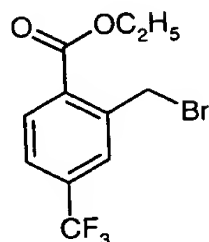
5   9 g (39 mmol) of ethyl 2-methyl-4-trifluoromethyl-benzoate are dissolved in 200 ml of carbon tetrachloride and admixed with 7 g (39 mmol) of N-bromo-succinimide and 0.1 g of dibenzoyl peroxide. The mixture is heated at reflux for 6 hours, and the precipitated succinimide is then filtered and the filtrate is concentrated under waterpump vacuum.

10   This gives 12 g of an amorphous residue which, in addition to ethyl 2-bromomethyl-4-trifluoromethyl-benzoate, also contains 17% of ethyl 2,2-dibromomethyl-4-trifluoromethyl-benzoate and 12% of ethyl 2-methyl-4-trifluoromethyl-benzoate.

15   Step 3
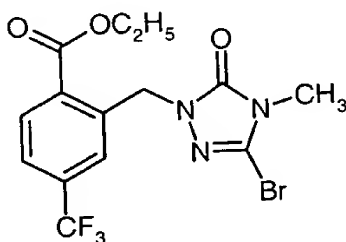
4 g of ethyl 2-bromomethyl-4-trifluoromethyl-benzoate (about 70% pure) and 2.28 g
20   (12.8 mmol) of 5-bromo-4-methyl-2,4-dihydro-3H-1,2,4-triazol-3-one are dissolved in 150 ml of acetonitrile, admixed with 5.3 g (38.4 mmol) of potassium carbonate and, with vigorous stirring, heated at reflux for 2 hours. The reaction mixture is taken up in water and extracted repeatedly with methylene chloride. The combined

methylene chloride phases are dried over sodium sulfate, concentrated under waterpump vacuum and chromatographed.

This gives 2 g (38% of theory) of 5-bromo-4-methyl-2-(2-ethoxycarbonyl-5-trifluoro-methyl-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one as an amorphous product.

^{1}H-NMR (CDCl$_3$, δ): 5.46 ppm.

Example (IV-2)
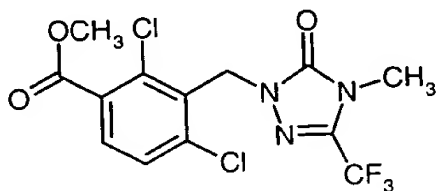
6.7 g (40 mmol) of 4-methyl-5-trifluoromethyl-2,4-dihydro-3H-1,2,4-triazol-3-one are initially charged in 150 ml of acetonitrile and stirred with 11 g (80 mmol) of potassium carbonate. The mixture is heated to 50°C, and a solution of 13.1 g (44 mmol) of methyl 3-bromomethyl-2,4-dichloro-benzoate in 20 ml of acetonitrile is then added dropwise with stirring, and the reaction mixture is heated at reflux with stirring for 15 hours. The mixture is then concentrated under waterpump vacuum and the residue is taken up in methylene chloride, washed with 1N hydrochloric acid, dried with sodium sulfate and filtered. The filtrate is concentrated under reduced pressure, the residue is digested with petroleum ether and the resulting crystalline product is isolated by filtration with suction.
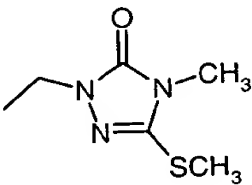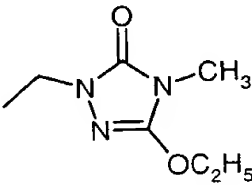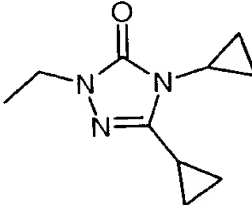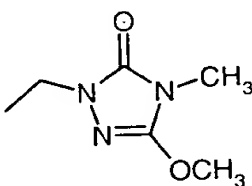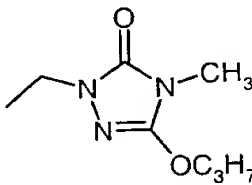
This gives 14.9 g (97% of theory) of 4-methyl-5-trifluoromethyl-2-(2,6-dichloro-3-methoxycarbonyl-benzyl)-2,4-dihydro-3H-1,2,4-triazol-3-one of melting point 109°C.
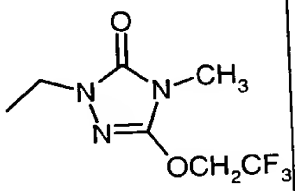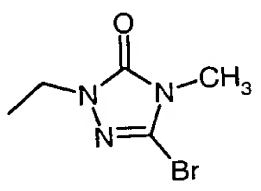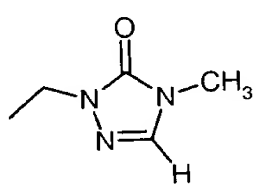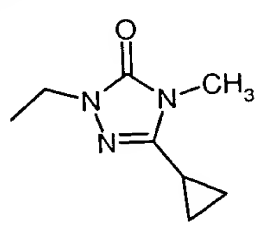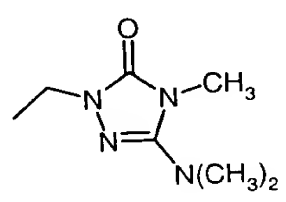
Similarly to Examples (IV-1) and (IV-2), it is also possible to prepare, for example, the compounds of the general formula (IV) listed in Table 4 below.
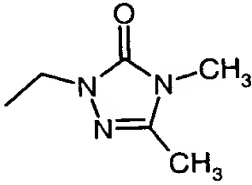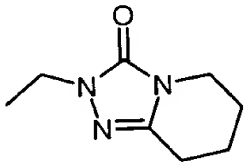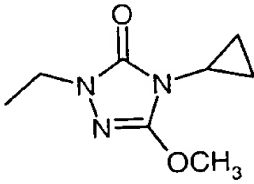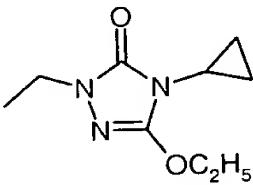
$$\text{X}-\overset{\overset{\text{O}}{\|}}{\text{C}}-\overset{(R^4)_n}{\underset{R^3}{\bigcirc}}-\text{A}-\text{Z} \qquad \text{(IV)}$$
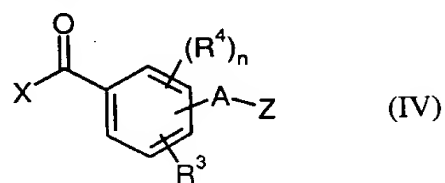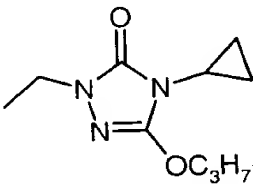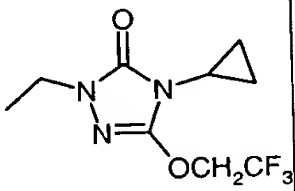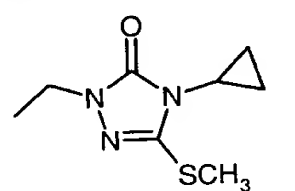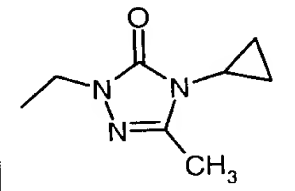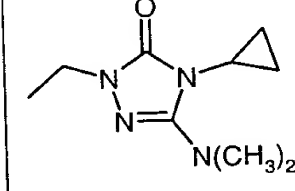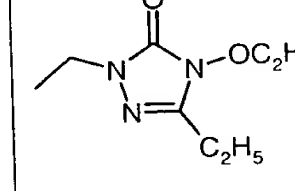
Table 4: Examples of the compounds of the formula (IV)

Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-3	(2-) Cl	(4-) Cl	(3-) triazolinone with N-ethyl, N-CH₃, SCH₃	OCH₃	m.p.: 229°C logP = 2.27 [a]
IV-4	(2-) Cl	(4-) Cl	(3-) triazolinone with N-ethyl, N-CH₃, OC_2H_5	OCH₃	m.p.: 120°C logP = 2.38 [a]
IV-5	(2-) Cl	(4-) Cl	(3-) triazolinone with N-ethyl, N-cyclopropyl, cyclopropyl	OCH₃	m.p.: 127°C logP = 2.55 [a]
IV-6	(2-) Cl	(4-) Cl	(3-) triazolinone with N-ethyl, N-CH₃, OCH₃	OCH₃	m.p.: 121°C logP = 2.04 [a]
IV-7	(2-) Cl	(4-) Cl	(3-) triazolinone with N-ethyl, N-CH₃, OC_3H_7-i	OCH₃	m.p.: 68°C logP = 2.73 [a]
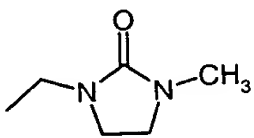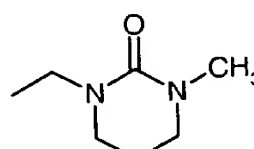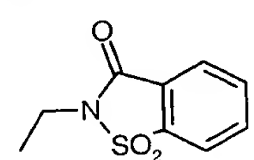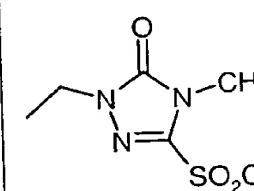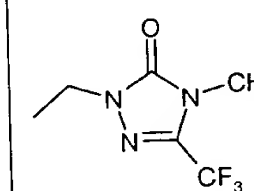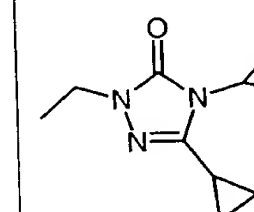
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-8	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 129°C logP = 2.72 [a]
IV-9	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 164°C logP = 2.18 [a]
IV-10	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 158°C logP = 1.55 [a]
IV-11	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 106°C logP = 2.16 [a]
IV-12	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 126°C logP = 2.11 [a]
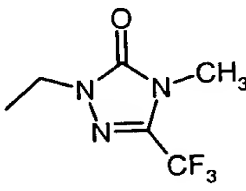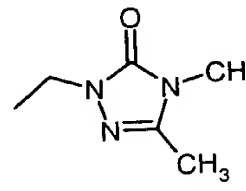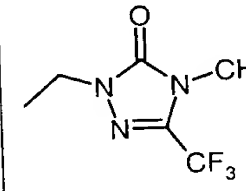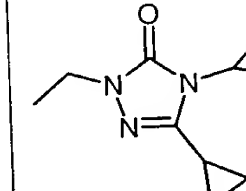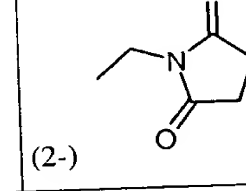
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-13	(2-) Cl	(4-) Cl	(3-)	OCH₃	m.p.: 146°C logP = 1.65 [a)]
IV-14	(2-) Cl	(4-) Cl	(3-)	OCH₃	m.p.: 178°C logP = 1.86 [a)]
IV-15	(2-) Cl	(4-) Cl	(3-)	OCH₃	m.p.: 97°C logP = 2.36 [a)]
IV-16	(2-) Cl	(4-) Cl	(3-)	OCH₃	m.p.: 99°C logP = 2.73 [a)]
IV-17	(2-) Cl	(4-) Cl	(3-)	OCH₃	m.p.: 56°C logP = 3.08 [a)]
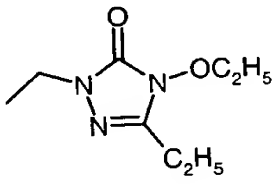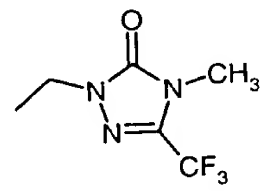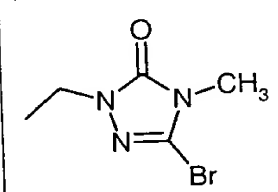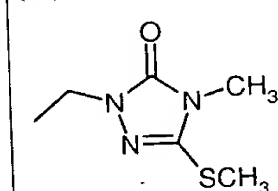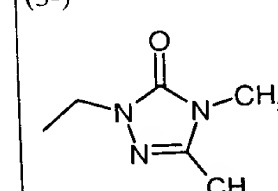
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-18	(2-) Cl	(4-) Cl	(3-) triazolinone ring with N-ethyl, N-cyclopropyl, OCH₂CF₃	OCH₃	m.p.: 102°C logP = 3.05 [a]
IV-19	(2-) Cl	(4-) Cl	(3-) triazolinone ring with N-ethyl, N-cyclopropyl, SCH₃	OCH₃	m.p.: 131°C logP = 2.70 [a]
IV-20	(2-) Cl	(4-) Cl	(3-) triazolinone ring with N-ethyl, N-cyclopropyl, CH₃	OCH₃	m.p.: 135°C logP = 1.97 [a]
IV-21	(2-) Cl	(4-) Cl	(3-) triazolinone ring with N-ethyl, N-cyclopropyl, N(CH₃)₂	OCH₃	m.p.: 143°C logP = 2.42 [a]
IV-22	(2-) Cl	(4-) Cl	(3-) triazolinone ring with N-ethyl, N-OC₂H₅, C₂H₅	OCH₃	m.p.: 85°C logP = 2.58 [a]
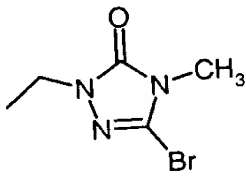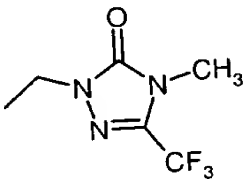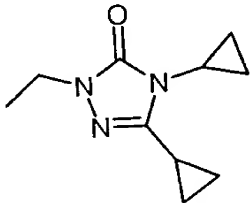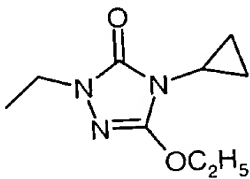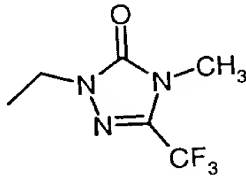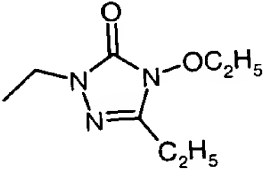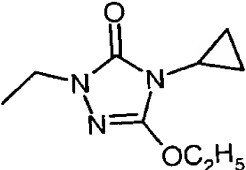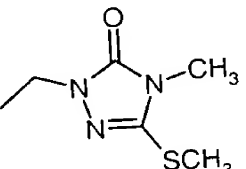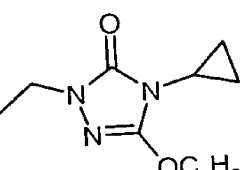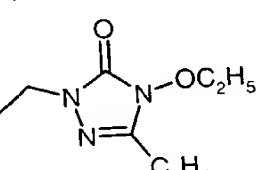
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-23	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	logP = 1.98 [a]
IV-24	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	logP = 2.07 [a]
IV-25	(2-) Cl	(4-) Cl	(3-)	OCH$_3$	m.p.: 157°C logP = 2.94 [a]
IV-26	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.53 ppm.
IV-27	(4-) NO$_2$	-	(3-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.48 ppm.
IV-28	(4-) NO$_2$	-	(3-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.30 ppm.
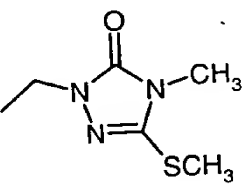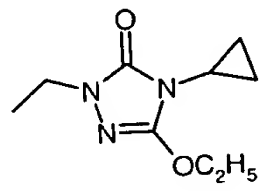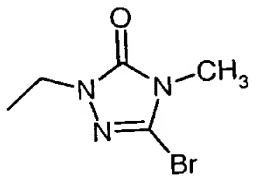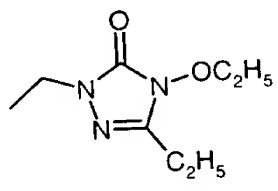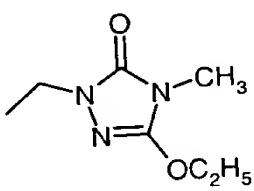
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-29	(4-) SO₂CH₃	-	(3-)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.61 ppm.
IV-30	(4-) Cl	-	(3-)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.08 ppm.
IV-31	(4-) Cl	-	(3-)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.17 ppm.
IV-32	(4-) Cl	-	(3-)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.00 ppm
IV-33	(4-) SO₂CH₃	-	(2-)	OC₂H₅	logP = 1.53 [a)
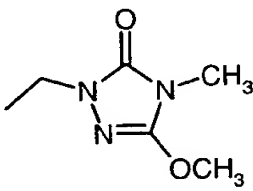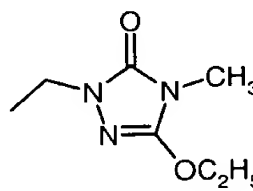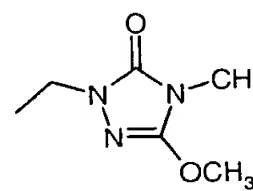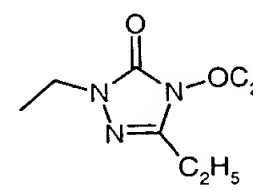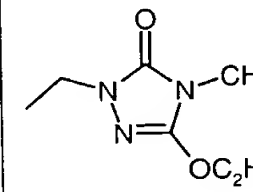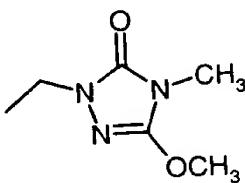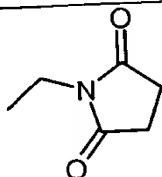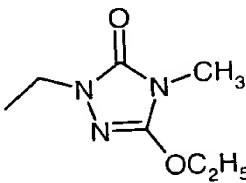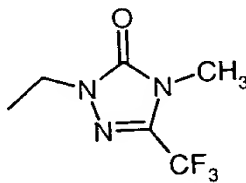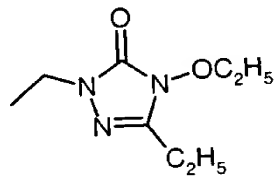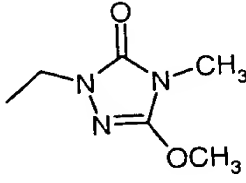
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-34	(4-) Br	-	(2-) 1-ethyl-4-(OC$_2$H$_5$)-5-(C$_2$H$_5$)-1,2,4-triazol-3(4H)-one	OC$_2$H$_5$	logP = 3.24 [a]
IV-35	(4-) Br	-	(2-) 1-ethyl-4-CH$_3$-5-CF$_3$-1,2,4-triazol-3(4H)-one	OC$_2$H$_5$	logP = 3.40 [a]
IV-36	(4-) F	-	(3-) 1-ethyl-4-CH$_3$-5-Br-1,2,4-triazol-3(4H)-one	OC$_2$H$_5$	logP = 2.41 [a]
IV-37	(4-) F	-	(2-) 1-ethyl-4-CH$_3$-5-SCH$_3$-1,2,4-triazol-3(4H)-one	OC$_2$H$_5$	logP = 2.45 [a]
IV-38	(4-) Br	-	(3-) 1-ethyl-4-CH$_3$-5-CH$_3$-1,2,4-triazol-3(4H)-one	OC$_2$H$_5$	logP = 2.06 [a]
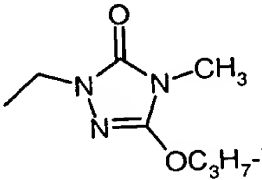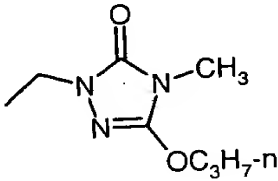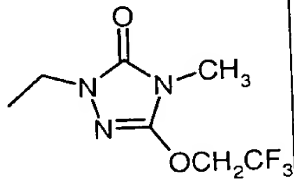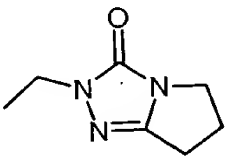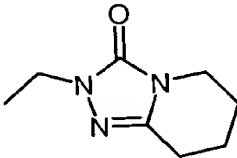
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-39	(4-) Br	-	(3-)	OC₂H₅	logP = 2.64 [a]
IV-40	(4-) Br	-	(3-)	OC₂H₅	logP = 3.23 [a]
IV-41	(4-) Br	-	(3-)	OC₂H₅	logP = 3.02 [a]
IV-42	(4-) Cl	-	(2-)	OC₂H₅	logP = 3.23 [a]
IV-43	(4-) Cl	-	(2-)	OC₂H₅	logP = 3.31 [a]
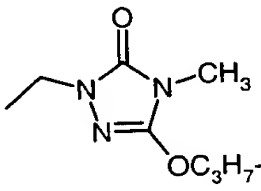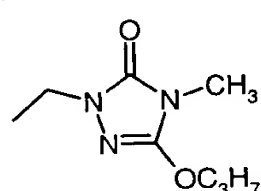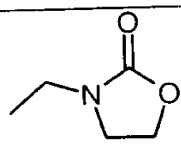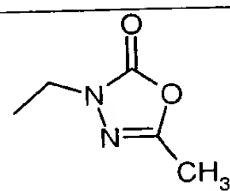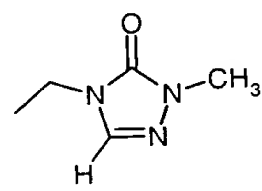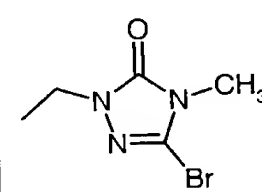
Ex. No.	(Position) R^3	(Position) $(R^4)_n$	(Position) -A-Z	X	Physical data
IV-44	(4-) Cl	-	(2-)	OC_2H_5	logP = 3.14 [a]
IV-45	(4-) NO_2	-	(2-)	OC_2H_5	logP = 2.42 [a]
IV-46	(4-) NO_2	-	(2-)	OC_2H_5	logP = 2.82 [a]
IV-47	(4-) CF_3	-	(2-)	OC_2H_5	logP = 3.48 [a]
IV-48	(4-) CF_3	-	(2-)	OC_2H_5	logP = 3.38 [a]
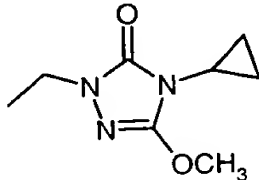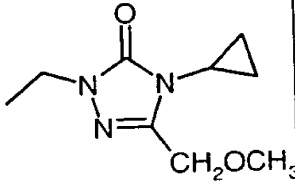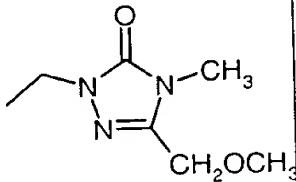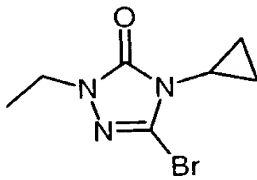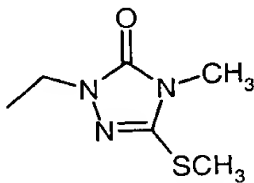
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-49	(4-) CF$_3$	-	(2-) triazolinone with N-ethyl, N-CH$_3$, SCH$_3$	OC$_2$H$_5$	logP = 3.02 [a]
IV-50	(4-) CF$_3$	-	(2-) triazolinone with N-ethyl, N-cyclopropyl, OC$_2$H$_5$	OC$_3$H$_7$	logP = 3.91 [a]
IV-51	(4-) OCH$_3$	-	(2-) triazolinone with N-ethyl, N-CH$_3$, Br	OC$_2$H$_5$	
IV-52	(4-) OCH$_3$	-	(2-) triazolinone with N-ethyl, N-OC$_2$H$_5$, C$_2$H$_5$	OC$_2$H$_5$	
IV-53	(4-) CF$_3$	-	(2-) triazolinone with N-ethyl, N-CH$_3$, OC$_2$H$_5$	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.37 ppm.
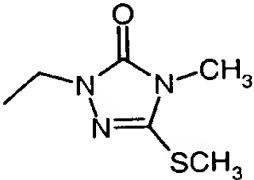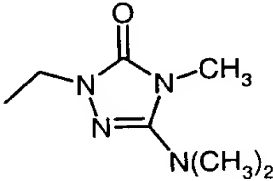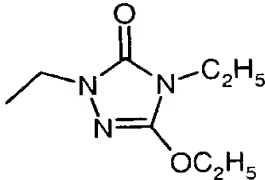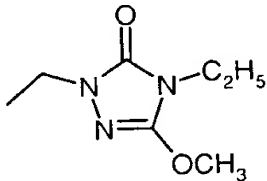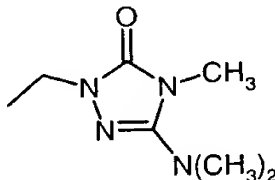
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-54	(4-) CF₃	-	(2-) triazolone ring, N-ethyl, 4-CH₃, 5-OCH₃	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.37 ppm.
IV-55	-	-	(2-) triazolone ring, N-ethyl, 4-CH₃, 5-OC₂H₅	OC₂H₅	
IV-56	-	-	(2-) triazolone ring, N-ethyl, 4-CH₃, 5-OCH₃	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.37 ppm.
IV-57	-	-	(2-) triazolone ring, N-ethyl, 4-OC₂H₅, 5-C₂H₅	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.40 ppm.
IV-58	(4-) Br	-	(2-) triazolone ring, N-ethyl, 4-CH₃, 5-OC₂H₅	OC₂H₅	logP = 2.95 [a)]

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-59	(4-) Br	-	(2-) *(triazolinone ring, N-ethyl, N-CH$_3$, OCH$_3$)*	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.31 ppm.
IV-60	(4-) Br	-	(2-) *(N-ethylsuccinimide ring)*	OC$_2$H$_5$	logP = 2.44 [a]
IV-61	(4-) F	-	(2-) *(triazolinone ring, N-ethyl, N-CH$_3$, OC$_2$H$_5$)*	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.35 ppm.
IV-62	(4-) F	-	(2-) *(triazolinone ring, N-ethyl, N-CH$_3$, CF$_3$)*	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.53 ppm.
IV-63	(4-) F	-	(2-) *(triazolinone ring, N-ethyl, N-OC$_2$H$_5$, C$_2$H$_5$)*	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.40 ppm.
IV-64	(4-) F	-	(2-) *(triazolinone ring, N-ethyl, N-CH$_3$, OCH$_3$)*	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.36 ppm.
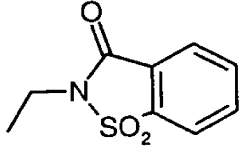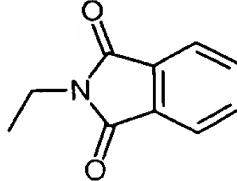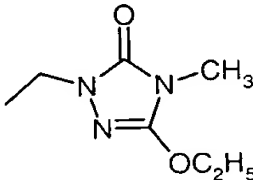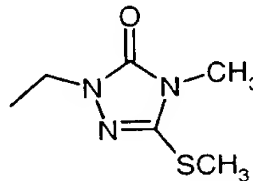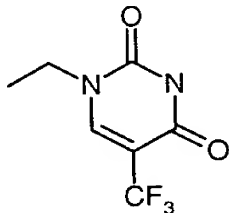
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-65	(4-) Br	-	(2-) ethyl-triazolone with CH₃ and OC₃H₇-i	OC₂H₅	logP = 3.34 [a]
IV-66	(4-) Br	-	(2-) ethyl-triazolone with CH₃ and OC₃H₇-n	OC₂H₅	logP = 3.38 [a]
IV-67	(4-) Br	-	(2-) ethyl-triazolone with CH₃ and OCH₂CF₃	OC₂H₅	logP = 3.31 [a]
IV-68	(4-) Br	-	(2-) ethyl fused bicyclic triazolone	OC₂H₅	logP = 2.16 [a]
IV-69	(4-) Br	-	(2-) ethyl fused bicyclic triazolone	OC₂H₅	logP = 2.41 [a]
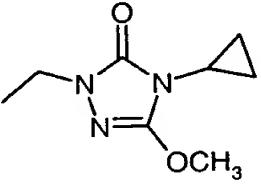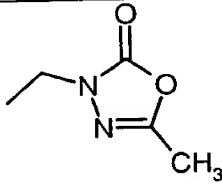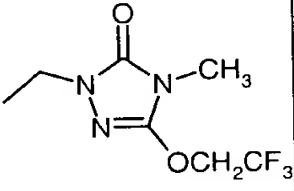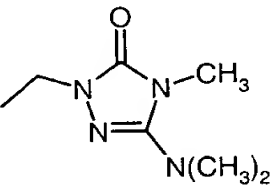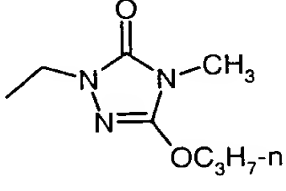
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-70	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	logP = 3.51 [a]
IV-71	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	logP = 3.54 [a]
IV-72	(4-) Br	-	(2-)	OC$_2$H$_5$	logP = 2.36 [a]
IV-73	(4-) Br	-	(2-)	OC$_2$H$_5$	logP = 2.88 [a]
IV-74	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	logP = 2.68 [a]
IV-75	(4-) Br	-	(2-)	OC$_2$H$_5$	logP = 2.80 [a]

Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-76	(4-) CF₃	-	(3-)	OC₂H₅	logP = 3.87 [a]
IV-77	(4-) CF₃	-	(2-)	OC₂H₅	logP = 2.88 [a]
IV-78	(4-) CF₃	-	(2-)	OC₂H₅	logP = 2.60 [a]
IV-79	(4-) CF₃	-	(2-)	OC₂H₅	logP = 3.35 [a]
IV-80	(4-) Br	-	(2-)	OC₂H₅	logP = 2.86 [a]
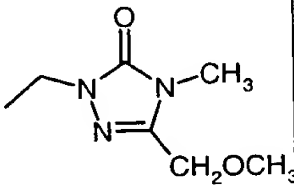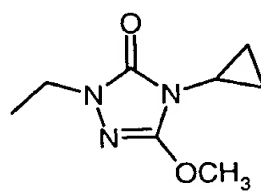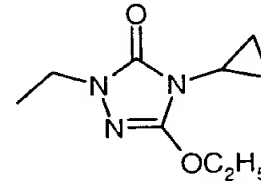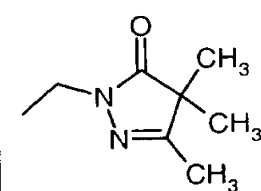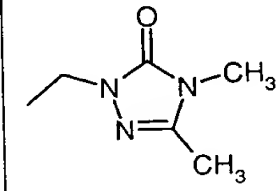
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-81	(4-) Cl	-	(2-) (triazolinone ring with ethyl, N-CH₃, SCH₃)	OC₂H₅	logP = 2.83 [a]
IV-82	(4-) Br	-	(2-) (triazolinone ring with ethyl, N-CH₃, N(CH₃)₂)	OC₂H₅	logP = 2.60 [a]
IV-83	(4-) CF₃	-	(2-) (triazolinone ring with ethyl, N-C₂H₅, OC₂H₅)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.36 ppm.
IV-84	(4-) CF₃	-	(2-) (triazolinone ring with ethyl, N-C₂H₅, OCH₃)	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.37 ppm.
IV-85	(4-) CF₃	-	(2-) (triazolinone ring with ethyl, N-CH₃, N(CH₃)₂)	OC₂H₅	logP = 2.79 [a]
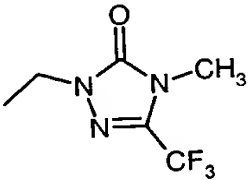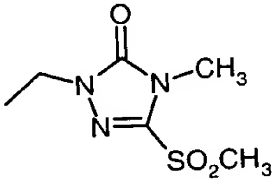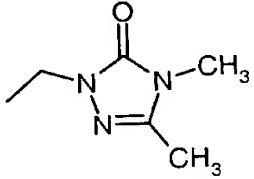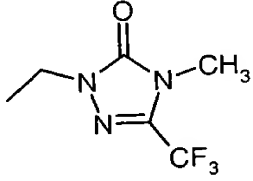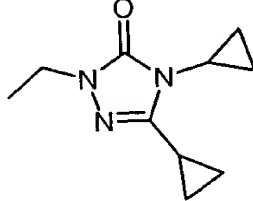
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-86	(4-) CF_3	-	(2-)	OC_2H_5	logP = 3.67 [a]
IV-87	(4-) CF_3	-	(2-)	OC_2H_5	logP = 3.80 [a]
IV-88	(3-) CH_3	-	(2-)	OC_2H_5	logP = 2.54 [a]
IV-89	(4-) SO_2CH_3	-	(2-)	OC_2H_5	logP = 1.82 [a]
IV-90	(4-) CF_3	-	(2-)	OC_2H_5	logP = 2.93 [a]
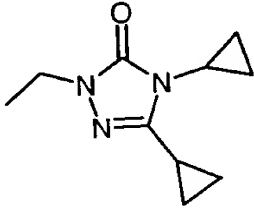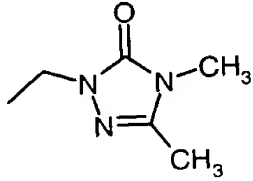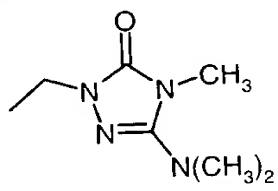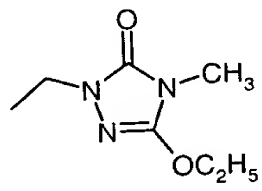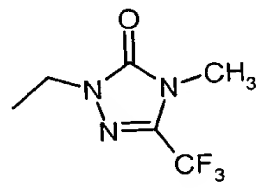
Ex. No.	(Position) R^3	(Position) $(R^4)_n$	(Position) -A-Z	X	Physical data
IV-91	(4-) CF_3	-	(2-) (1-ethyl-4-cyclopropyl-5-methoxy-1,2,4-triazol-3(4H)-one)	OC_2H_5	logP = 3.08 [a]
IV-92	(4-) CF_3	-	(2-) (3-ethyl-5-methyl-1,3,4-oxadiazol-2(3H)-one)	OC_2H_5	logP = 3.04 [a]
IV-93	(4-) CF_3	-	(2-) (1-ethyl-4-methyl-5-(OCH_2CF_3)-1,2,4-triazol-3(4H)-one)	OC_2H_5	logP = 3.45 [a]
IV-94	(4-) F	-	(2-) (1-ethyl-4-methyl-5-$N(CH_3)_2$-1,2,4-triazol-3(4H)-one)	OC_2H_5	logP = 2.21 [a]
IV-95	(4-) F	-	(2-) (1-ethyl-4-methyl-5-(OC_3H_7-n)-1,2,4-triazol-3(4H)-one)	OC_2H_5	logP = 2.96 [a]
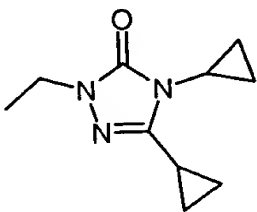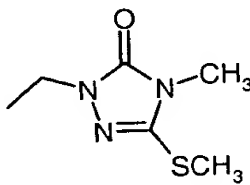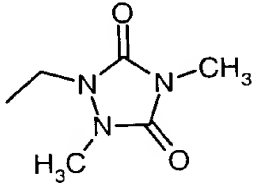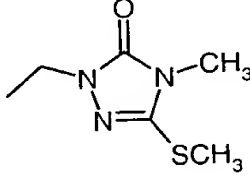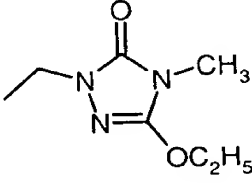
Ex. No.	(Position) R³	(Position) (R⁴)ₙ	(Position) -A-Z	X	Physical data
IV-96	(4-) F	-	(2-) structure with CH₃, CH₂OCH₃	OC₂H₅	logP = 2.05 [a]
IV-97	(4-) F	-	(2-) structure with cyclopropyl, OCH₃	OC₂H₅	logP = 2.50 [a]
IV-98	(4-) F	-	(2-) structure with cyclopropyl, OC₂H₅	OC₂H₅	logP = 2.89 [a]
IV-99	(4-) CF₃	-	(2-) structure with CH₃, CH₃, CH₃	OC₂H₅	logP = 2.91 [a]
IV-100	(4-) Cl	-	(2-) structure with CH₃, CH₃	OC₂H₅	¹H-NMR (CDCl₃, δ): 5.39 ppm.
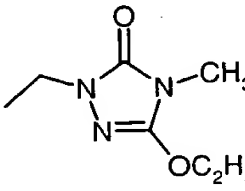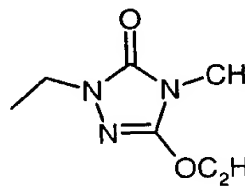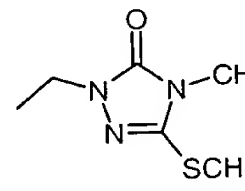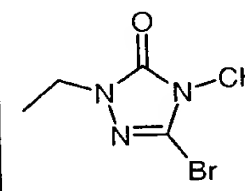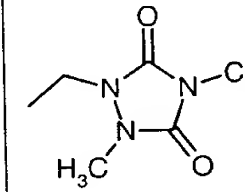
Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-101	(4-) Cl	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.50 ppm.
IV-102	(4-) Cl	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.49 ppm.
IV-103	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.29 ppm.
IV-104	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.53 ppm.
IV-105	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.34 ppm.

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-106	(4-) SO$_2$CH$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.39 ppm.
IV-107	(4-) SO$_2$CH$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.43 ppm.
IV-108	(4-) SO$_2$CH$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.40 ppm.
IV-109	(4-) SO$_2$CH$_3$	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.38 ppm.
IV-110	(4-) Br	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.49 ppm.

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-111	-	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.3 ppm.
IV-112	-	-	(2-)	OC$_2$H$_5$	^{1}H-NMR (CDCl$_3$, δ): 5.44 ppm.
IV-113	(4-) CF$_3$	-	(2-)	OC$_2$H$_5$	logP = 2.58 [a]
IV-114	(4-) SO$_2$CH$_3$	-	(2-)	OCH$_3$	logP = 1.53 [a]
IV-115	(4-) SO$_2$CH$_3$	-	(2-)	OCH$_3$	logP = 1.59 [a]

Ex. No.	(Position) R^3	(Position) (R^4)$_n$	(Position) -A-Z	X	Physical data
IV-116	(4-) I	-	(2-) triazolone, N-ethyl, N-CH$_3$, OC$_2$H$_5$	OCH$_3$	logP = 2.68 [a]
IV-117	(4-) CF$_3$	-	(2-) triazolone, N-ethyl, N-CH$_3$, OC$_2$H$_5$	OCH$_3$	logP = 2.74 [a]
IV-118	(4-) CF$_3$	-	(2-) triazolone, N-ethyl, N-CH$_3$, SCH$_3$	OCH$_3$	logP = 2.65 [a]
IV-119	(4-) CF$_3$	-	(2-) triazolone, N-ethyl, N-CH$_3$, Br	OC$_2$H$_5$	logP = 2.96 [a]
IV-120	-	-	(2-) triazolidinedione, N-ethyl, N-CH$_3$, N-CH$_3$ (H$_3$C)	OCH$_3$	m.p.: 106°C

The logP values given in Tables 4 were determined in accordance with EEC Directive 79/831 Annex V.A8 by HPLC (High Performance Liquid Chromatography) on a reversed-phase column (C 18). Temperature: 43°C.

(a) Mobile phases for the determination in the acidic range: 0.1% aqueous phosphoric acid, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked [a].

5

(b) Mobile phases for the determination in the neutral range: 0.01 molar aqueous phosphate buffer solution, acetonitrile; linear gradient from 10% acetonitrile to 90% acetonitrile - corresponding measurement results in Table 1 are marked [b].

10   Calibration was carried out using unbranched alkan-2-ones (having 3 to 16 carbon atoms) with known logP values (determination of the logP values by the retention times using linear interpolation between two successive alkanones).

The lambda max values were determined in the maxima of the chromatographic
15   signals using the UV spectra from 200 nm to 400 nm.

Use Examples:

Example A

Pre-emergence test

5

Solvent: 5 parts by weight of acetone

Emulsifier: 1 part by weight of alkylaryl polyglycol ether

To produce a suitable preparation of active compound, 1 part by weight of active
10   compound is mixed with the stated amount of solvent, the stated amount of
emulsifier is added and the concentrate is diluted with water to the desired
concentration.

Seeds of the test plants are sown in normal soil. After about 24 hours, the soil is
15   sprayed with the preparation of active compound such that the particular amount of
active compound is applied per unit area. The concentration of the spray liquor is
chosen such that the particular amount of active compound desired is applied in
1000 liters of water per hectare.

20   After three weeks, the degree of damage to plants is rated in % damage in
comparison to the development of the untreated control.

The figures denote:

25   0%   =   no effect (like untreated control)

100%   =   total destruction

In this test, for example, the compounds of Preparation Example 2 and 3 exhibit
strong activity against weeds, and they are tolerated well by crop plants such as, for
30   example, corn.

Example B

Post-emergence test

5   Solvent: 5 parts by weight of acetone

Emulsifier: 1 part by weight of alkylaryl polyglycol ether

To produce a suitable preparation of active compound, 1 part by weight of active compound is mixed with the stated amount of solvent, the stated amount of

10   emulsifier is added and the concentrate is diluted with water to the desired concentration.

Test plants of a height of 5 - 15 cm are sprayed with the preparation of active compound such that the particular amounts of active compound are applied per unit

15   area. The concentration of the spray liquor is chosen such that the particular amounts of active compound desired are applied in 1000 l of water/ha.

After three weeks, the degree of damage to the plants is rated in % damage in comparison to the development of the untreated control.

20

The figures denote:

0%   = no effect (like untreated control)

100% = total destruction

25

In this test, for example, the compound of Preparation Example 2 and 3 exhibit strong activity against weeds.